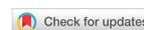


Review

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# Phonon and electron transport engineering for enhanced thermoelectric performance and the challenges of device integration

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## Abstract

Thermoelectricity has long been recognized as a transformative technology for power generation and cooling, owing to its capability to convert heat directly into electricity and vice versa, thereby facilitating cost-effective and environmentally friendly energy conversion. Following a period of modest activity, the field has experienced a remarkable resurgence since 2000, driven by significant advancements in the development of a diverse array of new materials and compounds, alongside enhanced capabilities for controlled nanostructuring. This rapid growth and the innovative breakthroughs observed over the past two decades can be largely attributed to a deeper understanding of the physical properties at the nanoscale. Among the various thermoelectric materials, nanostructured variants exhibit the highest potential for commercial application due to their unprecedented thermoelectric performance, which arises from substantial reductions in thermal conductivity. However, further advancements will not rely solely on nanostructuring; they will also necessitate novel electronic structure design concepts that require a comprehensive understanding of the complexities of electronic and phonon transport. These developments present significant opportunities for thermoelectric energy harvesting, power generation, and cooling applications. This article aims to summarize and elucidate the breakthroughs reported in recent years, discuss future avenues that integrate nanostructuring concepts with the rich electronic structures of novel materials, and provide a critical overview of the future directions in thermoelectric materials research. Additionally, it offers a comprehensive overview of state-of-the-art thermoelectric materials and devices and a summary of the



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challenges associated with transitioning these materials into practical devices.

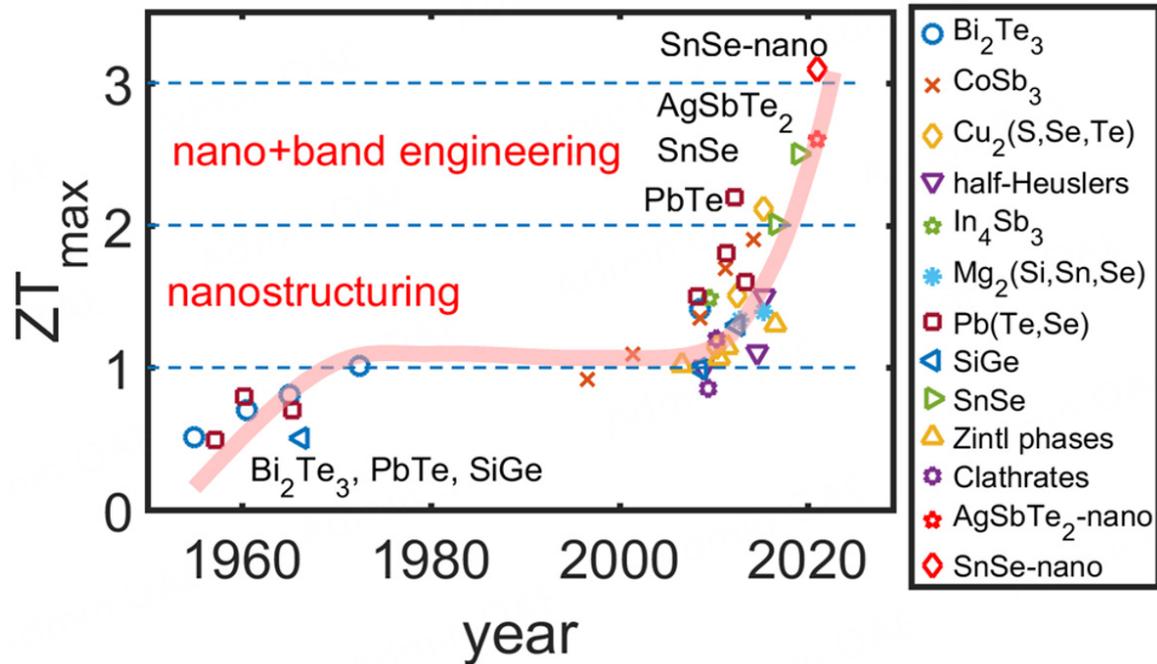
**Keywords:** Thermoelectricity, nanostructuring, phonon and electron transport, energy harvesting,  $zT$  figure of merit, thermoelectric devices

## INTRODUCTION

Thermoelectric (TE) devices are solid-state systems capable of generating electrical power from heat<sup>[1]</sup>. They have many advantages over conventional thermomechanical energy conversion devices due to their solid-state nature, absence of moving parts and high reliability. These advantages have led to some noteworthy applications, including the creation of direct electrical power (e.g., power generators for spacecraft), automotive climate control seats, small solid-state cooling diode lasers and infrared sensors, potential applications for power generation from solar irradiation, waste heat recovery, and powering Internet of Things (IoT) devices<sup>[2,3]</sup>. Companies such as RCA, 3M, Texas Instruments, Marlow Industries, and several other start-ups, have already developed and begun producing TE devices. However, their low efficiency compared to conventional thermomechanical cycles has limited their use to only niche applications and those for which the conventional cycles cannot be easily applied.

The ability of a TE material to convert heat into electricity is quantified by the  $zT$  figure of merit, given by  $zT = S^2\sigma T/\kappa$ , where  $\sigma$  is the electrical conductivity,  $S$  is the Seebeck coefficient (the product  $S^2\sigma$  is referred to as the power factor or  $PF$ ),  $T$  is the absolute temperature and  $\kappa$  is the thermal conductivity determining the losses of the process, composed of the electronic and phononic (or lattice) parts as  $\kappa = \kappa_e + \kappa_p$ . While these transport coefficients of TE materials have been studied for decades, their negative interdependence ( $\sigma$  is inversely proportional to  $S$  and directly proportional to  $\kappa_e$ ) has hindered the overall increase in  $zT$ <sup>[4]</sup>. The best commercial materials are based on compounds and alloys of Bi, Te and Pb, and provide  $zT \sim 1$ , which corresponds to  $\sim 10\%$  of the Carnot efficiency<sup>[5]</sup>. Since  $\text{Bi}_2\text{Te}_3$  and  $\text{PbTe}$  with  $zT \sim 1$  were developed more than 40 years ago [Figure 1], there have been many efforts to increase  $zT$  further. Alloy systems and carrier concentration optimization have been investigated as the main drivers to improve conventional thermoelectrics; however, the necessity of optimizing three adversely interdependent parameters in the same material was a fundamental restriction. As a result, around the mid-1970s [Figure 1], optimism faded, basic materials research slowed down, and the mainstream impression in the scientific community was that  $zT$  was constrained to unity. For such rather low  $zT$  values, the efficiency was insufficient to support competitive products compared to other technologies.

The resurgence in TE materials research started in the 90s after two influential studies published in 1993 by Hicks and Dresselhaus on Bi nanowires, suggesting an increase in performance from the use of low-dimensional TE materials<sup>[6,7]</sup>. They essentially provided a paradigm change, pointing out that quantum confinement could offer a fresh approach to improving the efficiency of TE materials. The core premise was that the low-dimensional density of states (DOS) provides sharp features in energy, an element that could largely increase the Seebeck coefficient, independently from the electrical conductivity (i.e., without reducing it), such that large  $PF$ s are reached. Although this initial suggestion that quantum confinement will improve the  $PF$  was never realized for reasons that we and others explained in later works<sup>[8-11]</sup>, research on nanomaterials quickly led to the realization that nanostructuring offered enormous opportunities for thermal conductivity reductions, from only mild changes all the way to even orders of magnitude. Such works became more frequent in the literature from around the year 2000 onwards, when material synthesis improved to the degree at which adequate control over nanomaterial design was possible. Funding agencies across the globe increased financial support, which led to a resurgence in the field [see Figure 1<sup>[12-18]</sup> and Table 1].



**Figure 1.** The evolution of the  $zT$  figure of merit over decades indicates enormous progress over the last 20 years for many different classes of materials. Data are indicative for the performance of different material families and have been extracted from several Refs.<sup>[12-18]</sup>. After the initial boost in performance from nanostructuring, the addition of band engineering drives further improvements (regions only roughly denoted).

Nanostructuring has had such a strong impact on  $zT$  through drastic thermal conductivity reductions, allowing  $zT > 2$  across materials and operating temperatures, and dominated the research field in the last 20 years. With thermal conductivities reaching nowadays the amorphous limit and below for many materials, however, further benefits from this approach could be reaching their limits. It is worth mentioning that the reduction of thermal conductivity below the amorphous limit arises mostly from recalibrated models accounting for reduced heat capacity or strong scattering, rather than a fundamental breach of physical limits. For example, porous SiGe nanocrystalline materials reported thermal conductivities of as low as  $0.5 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  due to ultra-strong phonon boundary scattering, ostensibly below the traditionally cited amorphous limit ( $1\text{-}2 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$ )<sup>[31]</sup>. Similar adjustments have been applied to hierarchical nanostructures in PbTe-SrTe, where alloy scattering parameters differ from pristine materials<sup>[32]</sup>. Another additional strategy is now emerging for further advancement in the field, namely band engineering of alloyed TE materials, which targets  $PF$  improvements. The combination of the two has already demonstrated examples of  $zT$ s close to, or even beyond  $zT > 3$ <sup>[12-14,33]</sup>, with even an extraordinary estimated value of  $zT \sim 5$  in one case<sup>[34]</sup>.

With the above crucial considerations in mind, this paper describes the main directions for the TE research field over the last period and promising future opportunities. The rest of the paper is organized as follows: Section "REDUCING THERMAL CONDUCTIVITY BY HIERARCHICAL NANOSTRUCTURING" discusses the mechanisms that allowed improvements to  $zT$  by nanostructuring, specifically targeting thermal conductivity reductions, and summarizes some of the important milestone works. Section "USING NANOSTRUCTURING TO IMPROVE THE POWER FACTOR" follows with the main approaches to improve the  $PF$  of materials, again using nanostructuring and nanomaterials. Section "BAND STRUCTURE ENGINEERING" presents the improvements that band engineering can additionally bring and the promising strategies that are currently followed towards this. Section "TRANSITIONING HIGH  $zT$

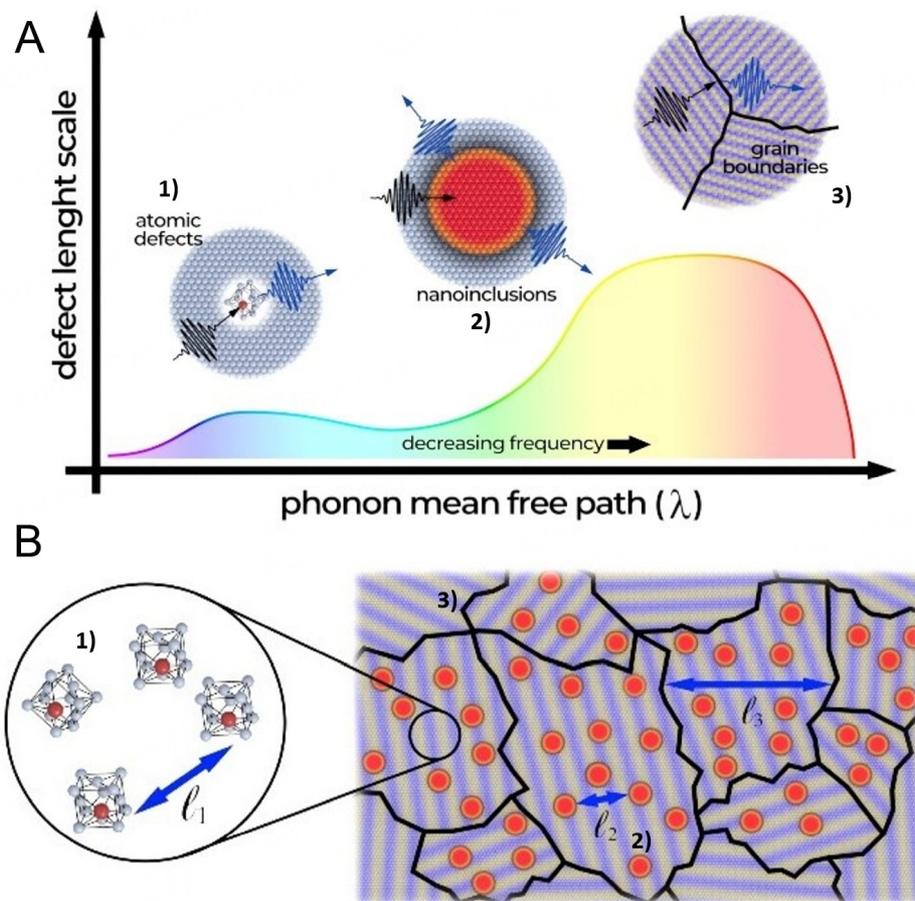
**Table 1. Data reported in Figure 1 according to their optimal operating temperature range**

Temperature range	Material	Optimal T (K)	zT	Ref.
Near Room	Bi <sub>2</sub> Te <sub>3</sub>	320	1.4	[19,20]
	AgSbTe <sub>2</sub>	320	1.5	[14]
	Iodine-doped Cu <sub>2</sub> Se	400	2.3	[21]
Medium	AgSbTe <sub>2</sub>	573	2.6	[14]
	Zintl phase (Mg <sub>3</sub> Sb <sub>2</sub> )	725	1.65	[15,22]
	SnSe polycrystalline	783	3.1	[12]
	InSb	773	1.28	[15,23]
	CoSb <sub>3</sub>	835	1.9	[15,24]
	PbTe	820	2.2	[15,25]
	SnSe single crystal	923	2.6	[18]
High	Cu <sub>2-x</sub> Se	1,000	1.5	[15,26]
	Cu <sub>2</sub> Se	1,030	2.62	[15,27]
	SiGe	1,073	1.84	[28]
	Half-Heuslers (Hf,Zr, Ti)CoSb <sub>0.8</sub> Sn <sub>0.2</sub> )	1,073	1.0	[15,29]
	Half-Heuslers (FeNbSb)	1,200	1.5	[15,30]

"THERMOELECTRIC MATERIALS TO DEVICES" discusses current state-of-the-art efforts for TE devices: their merits and challenges in practical applications and the main challenges for transitioning from materials to devices. Section "CONCLUSION AND OUTLOOK" finally concludes the review with our wrapping up of the main global considerations on the topic.

## REDUCING THERMAL CONDUCTIVITY BY HIERARCHICAL NANOSTRUCTURING

In the context of thermoelectricity, the process and consequence of nanostructuring is essentially to introduce a variety of defects in the channel material to act as phonon scattering centers, reducing the lattice's thermal conductivity. Of course, these also form scattering centers for electrons which reduce the electrical conductivity, but in general, the thermal conductivity is reduced much more since the mean-free-paths (mfps) for scattering of the dominant phonons are in most materials significantly longer compared to those of electrons. The thermal energy-carrying phonons have a distribution of mfps, ranging from nanometers to micrometers, even millimeters. Due to this vast distribution of mfps, one of the most successful strategies to reduce thermal conductivity is to *hierarchically* nanostructure the materials. This involves the incorporation of a variety of nano-features in the material, each targeting to scatter phonons with different mfps (see Figure 2A). Phonon scattering is characterized by complex dynamics, roughly: (i) atomic-scale defects or alloying that can effectively scatter short wavelength phonons, (ii) nanoscale defects that can scatter short and medium wavelength phonons (with mfps up to ~100 nm), and (iii) micro and mesoscale defects (e.g., grain boundaries) that can scatter long wavelength, low-frequency phonons (with mfps from 100 s of nm to several μm, even up to ~1 mm). Due to their intrinsic size, these nano- and micro-features are also naturally distanced in a hierarchical fashion according to their intrinsic dimension as well, matching the corresponding mfp of the respective targeted phonons, as shown in Figure 2B. Note that the closer these features are placed, the stronger phonon scattering they cause (one could imagine a nanocrystalline material with grains of only several nm). However, this also results in much stronger electron scattering and subsequent *PF* reduction; electrons are mainly affected by nanoscale defects and their short-range arrangement, with the longer-range defects and arrangements being less relevant. The hierarchical architecture allows taking advantage of the fact that (i) different types of defects target distinct phonons, while (ii) the distance between them is also of the order of those phonon mfps, optimizing strong phonon, but weak electron scattering. In this way, thermal conductivity reduction from phonons across the



**Figure 2.** (A) Schematic of hierarchical (panoscopic) nanostructuring, where atomic defects, nano-inclusions, and grain boundaries - spanning atomic to macro scales - act as phonon scattering centers across different mean-free-paths (mfps), reducing lattice thermal conductivity<sup>[35]</sup>. The lower color plot illustrates typical frequency and mfp contributions to thermal conductivity. Combining various defect types and sizes is essential to target the full phonon spectrum. (B) Nanostructured material schematic showing optimal placement of: (1) Atomistic defects (e.g., vacancies, dopants) for high-frequency phonon scattering, with controlled concentrations to minimize electron scattering; (2) Nano-inclusions of varied sizes and interface types to scatter a wide mfp range while preserving electron transport; and (3) Grain boundaries in nanocrystalline materials for mid- to long-mfp phonon scattering<sup>[36]</sup>, with secondary phases at boundaries to enhance phonon blocking without significantly impeding electron flow.

entire spectrum can be achieved, while the electronic conductivity suffers significantly less<sup>[37]</sup>.

Although nanostructuring is currently considered the most promising way forward, one needs to consider that some nanostructures, such as grain boundaries, for example, may suffer from stability and reliability issues. This can be the case if the material is exposed to temperatures higher than those used to produce the material. Thus, to effectively benefit from enhanced phonon scattering along the lifetime of the device/material, it must be operated under conditions at which the nanostructures do not evolve.

One of the earliest and most successful examples of nanostructuring was demonstrated by Biswas *et al.* for PbTe in 2012, for a system that reached an ultra-low lattice thermal conductivity of  $\kappa_p < 0.9 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  at 915 K and a record  $zT$  of  $\sim 2.2$  at the time<sup>[17]</sup>. In that and subsequent studies, the authors were able to nanostructure PbTe/SrTe in what they referred to as “hierarchical” or “panoscopic” nanostructuring, which included atomic-, nano-, and mesoscale defects to target a range of phonon mfps and reduce the lattice thermal conductivity across the spectrum<sup>[17,38,39]</sup>. Later, using this method for the p-type  $\text{Pb}_{0.98}\text{Na}_{0.02}\text{Te-SrTe}$

system, Tan *et al.* reported an even lower lattice thermal conductivity of  $\kappa_p = 0.5 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  and a higher  $zT$  of 2.5 at 923 K<sup>[32]</sup>. This multi-scale, or all-size phonon scattering center inclusion is now a widely used approach to reduce  $\kappa_p$ . Around that period and since then, many more examples of successful hierarchical nanostructuring emerged in the literature, with nanocomposites characterized by multi-scale phonon scattering centers of different nature and sizes (see [Figure 2A](#)).

Such nanostructuring was also able to raise interest for materials that were not traditionally considered good TEs due to their high  $\kappa_p$ , such as the abundant Si. It was realized early on that Si in the form of nanowires<sup>[40-42]</sup>, superlattices<sup>[43,44]</sup>, porous media<sup>[31,45]</sup>, and combinations of those could result in two orders of magnitude reduction in  $\kappa_p$ , reaching values close to or even below the amorphous limit. For example, a porous SiGe nanocrystalline material<sup>[31]</sup> provided scattering of phonons due to alloying, pore scattering and boundary scattering, together with reduced heat capacity because of porosity, which resulted in thermal conductivities of  $\kappa_p = 0.5 \text{ Wm}^{-1}\text{K}^{-1}$ .

Another highly discussed material is SnSe. Initial reports for the TE performance of the pristine single crystal p-type material indicated a  $zT$  value of 2.6<sup>[18]</sup>, sharply peaking at high temperatures due to strong bond anharmonicity, resulting in ultra-low thermal conductivities. This was a new  $zT$  record at the time. The corresponding n-type performance reached a  $zT$  of 2.2<sup>[46]</sup>. Many subsequent studies on polycrystalline samples, on the other hand, reported values consistently below unity (although one would have expected that nanostructuring would reduce the thermal conductivity even lower). Later on, another study has, however, treated the samples in a specific way to remove the localized oxygen trapped around the grain boundaries. This resulted in another  $zT$  record of 3.1 [see [Figure 1](#)]<sup>[12,33]</sup>, and better average  $zT$  over temperature overall, indicating that the trapped oxygen was hurting the performance in two ways: increasing the thermal conductivity across the grain boundaries, and reducing the electronic conductivity. In other examples, local lattice distortions - such as rattling atoms in skutterudites or strain fields in half-Heuslers - reduce  $\kappa_p$  by 15%-30% without significantly altering  $\sigma$ <sup>[47]</sup>. Recent work on Cd-doped-AgSbTe<sub>2</sub> nanocomposites demonstrates  $zT > 2.5$  via strain-induced phonon localization, validated by transmission electron microscopy (TEM)<sup>[14]</sup>.

Although initially bottom-up nanostructuring demonstrations were targeted, nowadays top-down approaches, such as mechanical alloying and high-energy ball milling followed by hot pressing sintering or spark plasma sintering (SPS), are widely used to synthesize nanoparticle powders and bulk-size TE materials. This is now widely applied across material families. In this way, the  $zT$  values of the different materials were gradually doubled and sometimes even tripled compared to their pristine values. For example, the abundant and non-toxic family of silicides also raised significant interest due to their ability to be nanostructured and extensive studies on these materials exist (Mg<sub>2</sub>Si<sup>[48,49]</sup>, metal silicides including CrSi<sub>2</sub><sup>[50]</sup>, CoSi<sub>2</sub><sup>[51]</sup>, TiSi<sub>2</sub>, VSi<sub>2</sub><sup>[52]</sup>, and YbSi<sub>2</sub><sup>[53]</sup>). Nanostructuring methods can produce nanoscale precipitates located within larger grain formations in the Si matrix, which reduces  $\kappa_p$  drastically, as another example of a successful hierarchically nanostructured geometry. Moreover advances in MnSi<sub>1.7</sub> (higher manganese silicide) have demonstrated a  $zT$  of 0.8 at 800 K, outperforming CrSi<sub>2</sub> and CoSi<sub>2</sub>. This improvement stems from intrinsic vacancy engineering, which enhances phonon scattering while maintaining hole mobility<sup>[54]</sup>.

Nanostructuring, thus, is a proven strategy that over the past two decades has increased the nominal  $zT$  values of the most prominent TEs to the range between 1-2 and sometimes even higher (see [Figure 1](#)). There is still some fascinating work to be done to fully understand the effect of nanostructuring on phonon scattering and to create materials that can eliminate most of the phonon contribution to the thermal conductivity (we discuss below an example of a three-dimensional (3D) network material<sup>[55]</sup> which

approaches such point). However, improving the  $zT$  further requires improvements to the  $PF$  as well, on top of retaining the nanostructured material geometry, for which targeting the roadblock of the adverse interdependence of the electrical conductivity and Seebeck coefficient is encountered. In this regard, directions such as energy filtering and band engineering are shown to be very powerful tools, enabled by developments in the control of nanostructure and alloy synthesis. This was also driven by developments in theoretical and computational tools that enhanced our understanding of transport in complex materials and how chemical bonding affects electronic structure features<sup>[13,56]</sup>.

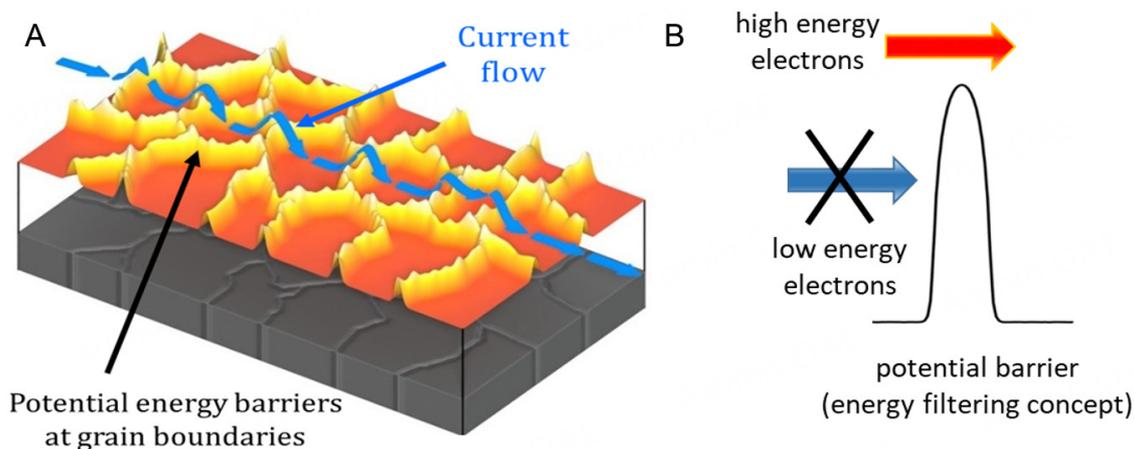
Importantly, in some of the  $zT$  record-breaking works mentioned above, care was taken such that the insertion of nanoinclusions does not degrade the  $PF$  significantly. Typically, nanostructuring reduces the electrical conductivity and increases the Seebeck coefficient slightly, but the reduction in conductivity has a stronger influence on the  $PF$ <sup>[57]</sup>, which is also reduced, albeit at a smaller degree compared to the thermal conductivity. Some of the ways this  $PF$  reduction is mitigated are by aligning the band edges of the constituent material phases such that electronic transport is not noticeably interrupted<sup>[17,38,39]</sup> alloying using iso-valent substitutions in the lattice (which minimize electron-alloy scattering) using atoms with different atomic weight, like Sr instead of Pb in lead tellurides<sup>[17]</sup>, or Zr and Hf instead of Ti in TiNiSn half-Heusler compounds<sup>[58,59]</sup>; introducing energy filtering through potential barriers which improve the Seebeck coefficient<sup>[57]</sup>; through charge transfer from doped islands (modulation doping)<sup>[60]</sup>, or doped interstitial voids<sup>[61]</sup> in the crystal matrix, which improves conductivity without degrading the mobility, but still reducing the thermal conductivity, *etc.*

## USING NANOSTRUCTURING TO IMPROVE THE POWER FACTOR

### Developments in energy filtering

From the above strategies to mitigate the reduction by nanostructuring on the  $PF$ , energy filtering is a common underlying cause when  $PF$  improvements are encountered. It is typically achieved by the introduction of potential barriers that form at the interfaces between different materials phases, at the grain boundaries and material discontinuities, in the presence of nanoscale defects, *etc.* These barriers allow high-energy electrons to propagate more easily, while blocking low-energy electrons, thus directly increasing the Seebeck coefficient (see Figure 3). Of course, potential barriers degrade the electrical conductivity. However, design "sweet spots" can be found for which the  $PF$  is improved, but even so not significantly, only at the order of 30%<sup>[57,62-65]</sup>. Nevertheless, a surge in efforts to use energy filtering and design the grain/grain-boundary system efficiently in a variety of materials has recently emerged<sup>[66-69]</sup>.

A few experimental designs, however, backed by theoretical calculations, have demonstrated extraordinarily high  $PF$ s in Si-based nanocrystalline materials. Specifically, a simultaneous improvement in both the electrical conductivity and the Seebeck coefficient in p-type Si layers was achieved, which led to  $PF$  improvements in various instances from  $2 \times 10^3$ <sup>[68-73]</sup>. For this to be realized, certain design "ingredients" needed to be fit in place: (i) the presence of energy filtering barriers between the nanograins, (ii) the size of the nanograins had to be a few 10 s of nanometers, (iii) ultra-high doping up to the levels of solid solubility had to be used, and finally, and (iv) the materials had to be annealed at high temperatures to allow precipitation of the dopants at the grain boundaries. Compared to bulk Si, whose maximum room temperature  $PF$  is approximately  $\sim 4-5 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  for p-type and  $\sim 5 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  for n-type<sup>[74]</sup>, values of  $\sim 6.5 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  were reported for n-type films<sup>[72]</sup>,  $\sim 15 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  for ultra-highly doped p-type nanocrystalline Si films with grain sizes  $\sim 30 \text{ nm}$ <sup>[70]</sup>,  $\sim 22 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  in the latter after the additional inclusion of nanopores<sup>[71]</sup>, and up to  $\sim 33 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  in such ultra-highly doped nanocrystalline films after they had undergone a dehydrogenated process<sup>[73]</sup>.



**Figure 3.** (A) An illustration of the effect of energy filtering at the grain boundaries of a nanocrystalline material. The vertical elevations are a plot of the energy level diagram at the interfaces that the electrons will find, to visually explain the energy filtering process and the potential energy barriers (in yellow) at the grain boundaries, which act as filtering barriers for high-energy electrons. The energy of the current flow (blue arrows) increases locally near the barriers (electrons absorb phonons and gain energy to overpass them) which locally increases the Seebeck coefficient. (B) A schematic of high-energy electrons passing through a grain boundary (potential barrier), where only carriers with energy greater than the barrier height can pass through, while the rest are blocked.

Theoretical calculations have shown that the properly designed grain/grain-boundary (well/barrier) interface can indeed lead to very high  $PF$ s<sup>[74]</sup>. Such a design includes (i) heavily doped, narrow nanograins that allow access to high carrier energies and velocities, and do not allow significant carrier energy relaxation (to mitigate the local reduction in Seebeck in the grains<sup>[64,75]</sup>); and (ii) grain-boundaries that provide energy filtering barriers, but themselves and the regions around them are less doped or ‘clean’ from dopants to allow higher carrier mobilities (and mitigate the local reduction in conductivity by the boundaries). Very recently,  $PF$  values of  $\sim 11 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  were demonstrated in p-type Si that implemented this concept under controlled nanofabrication conditions on a Silicon-on-Insulator (SOI) wafer, a step towards on-chip energy harvesting and cooling<sup>[76]</sup>. Although this was demonstrated for Si as a flexible platform to control the nano-feature and barrier details<sup>[77]</sup>, this design direction, or parts of it, can widely be applied to other materials as well during the process of nanostructuring.

### Developments in 3D nanostructured networks

Even though many nanostructures outperform bulk materials with the same material composition in terms of TE performance, it is crucial to develop structures with nanoscale properties covering wider size scales for use in commercial TE applications. The idea is to combine the advantages of nanostructuring, low-dimensionality, and various interfaces that effectively scatter phonons, into a 3D “size” material. Some of the first examples of this were superlattices, where a periodic arrangement of dissimilar materials creates interfacial scattering and band structure modifications<sup>[43,44]</sup>. A new direction is the use of proximity field nanopatterning (PnP), a technique that enables the construction of structures with nanoscale feature sizes, while preserving a continuous structure over a sizable region. As an example, a periodic 3D pattern of  $\text{Bi}_{1.5}\text{Sb}_{0.5}\text{Te}_3$  ternary alloys made up of a square array of nanopillars can be created to improve their performance<sup>[78]</sup>.

Another recent example, which in addition to low thermal conductivity targets high  $PF$ s, is a material formed by a 3D nanowire network<sup>[55]</sup>. A recent study employing 3D/1D anodic alumina templates has demonstrated a remarkable  $\sim 5$ -fold reduction in thermal conductivity in 3D-CuNi nanonetworks. Notably, this significant suppression, relative to the bulk material, has minimal impact on the Seebeck coefficient and

electrical conductivity, resulting in an approximately five-fold enhancement in the  $zT$ <sup>[79]</sup>. Similarly, a 3D arrangement of stoichiometric and highly oriented [110] bismuth telluride was created by electrodeposition, consisting of 55 nm diameter longitudinal nanowires joined by 20 nm diameter transversal nanowires. In the longitudinal direction, it was shown that the electrical conductivity of the 3D nanowire network is comparable to the high conductivity typical of  $\text{Bi}_2\text{Te}_3$  films produced by the same technique<sup>[80]</sup>. On the other hand, the Seebeck coefficient increases to values up to  $\sim 130 \mu\text{V}\cdot\text{K}^{-1}$ <sup>[55]</sup>, when they are measured as 3D nanowires networks. Along the transverse nanowire interconnection direction, due to nanostructuring, the thermal conductivity at room temperature reaches values as low as  $0.5 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$ , lower than 1D nanowires of similar diameter<sup>[81]</sup>. This value suggests that the phonon part of the thermal conductivity along the direction of the transverse interconnection is negligible. The electronic component is thus the only one contributing to thermal conductivity in that direction, and as a result the Seebeck coefficient and the Lorenz number are the only variables that affect the  $zT$  in these 3D nano-networks (in the direction along the transverse interconnections), opening a new research direction for performance enhancement.

### Developments in 2D materials

Monolayers of two-dimensional (2D) materials are emerging as promising candidates for TE materials, offering a novel platform where high electrical conductivity coexists with a high Seebeck coefficient<sup>[82,83]</sup>. This unique combination is attributed to their low dimensionality, atomically clean surfaces, absence of dangling bonds, and, when prepared with precision, minimal roughness of the 2D surface, which results in reduced quantum well thickness variation. Furthermore, recent studies have uncovered novel phenomena such as metal-insulator transitions<sup>[84,85]</sup> and electronic correlations that generate charge density waves<sup>[86,87]</sup>, which can enhance TE performance. The ability to create heterostructures also allows for bandgap tuning<sup>[88]</sup> and renormalization<sup>[89]</sup>, expanding the functional capabilities of these materials.

Recent advancements have indicated the potential for large Berry curvature in stacked 2D materials, which can lead to significant anomalous Nernst coefficients (in the absence of a magnetic field). This property opens avenues for realizing TE spin-Nernst currents, which are particularly relevant for spintronic applications<sup>[90-93]</sup>. Additionally, 2D materials typically exhibit moderate to low thermal conductivities, especially in the cross-plane direction when monolayers are stacked<sup>[82,94,95]</sup>, which is advantageous for TE efficiency.

The experimental techniques employed to fabricate and investigate the TE characteristics of 2D materials include mechanical exfoliation, which produces single flakes but often results in structural or chemical defects. This straightforward method allows for rapid assessment of TE properties, with promising  $PF$ s reported for various 2D materials, such as  $\text{WSe}_2$ , which has demonstrated a  $PF$  of approximately  $3.7 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$ <sup>[96]</sup>. Continuous 2D films have also been produced through methods such as screen or inkjet printing and vacuum filtration; however, these techniques typically rely on 2D flakes generated via exfoliation, which can compromise film quality<sup>[97-99]</sup>. While selected 2D materials in film form such as  $\text{WSe}_2$  achieve  $PF$ s rivaling bulk  $\text{Bi}_2\text{Te}_3$ , most monolayer 2D materials (e.g.,  $\text{MoS}_2$ ,  $\text{WS}_2$ ) underperform due to their simplified band structures and interfacial scattering (see Table 2); thus, more research is required in terms of materials development and integration to improve performance.

Alternative approaches, such as intercalation techniques previously applied to non-oxidized graphene, have shown promise in enhancing  $PF$ s to over  $0.6 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$ <sup>[100-104]</sup>. Flexible 2D thermoelectric generators (TEGs) are beginning to emerge in the literature, with examples based on  $\text{NbSe}_2$  (p-type,  $PF \sim 0.026 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$ ) and  $\text{WS}_2$  (n-type,  $PF \sim 0.005 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$ ) at room temperature<sup>[105]</sup>. Other 2D semiconductors, including  $\text{MoS}_2$  and  $\text{MoTe}_2$ , have exhibited  $PF$ s around 0.03 and  $0.8 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$ ,

**Table 2. Comparative power factors for Bi<sub>2</sub>Te<sub>3</sub> bulk and various thin films**

Material	Power factor (mW·m <sup>-1</sup> ·K <sup>-2</sup> )	Temperature (K)	Ref.
Bi <sub>2</sub> Te <sub>3</sub> (bulk)	4.5	300	[110]
Sb <sub>2</sub> Te <sub>3</sub> /MoS <sub>2</sub> (multilayer)	4.9	316	[82]
WSe <sub>2</sub> (2D film)	3.7	300	[96]
NbSe <sub>2</sub> (nanosheets)	0.026	300	[105]
WS <sub>2</sub> (nanosheets)	0.005	300	[105]
Graphene films	0.6	320	[104]
MoS <sub>2</sub> (p-type)	0.0303	460	[106]
MoTe <sub>2</sub> (n-type)	0.815	670	[107]
Mo-based MXenes	0.3	800	[109]

respectively<sup>[106,107]</sup>. Additionally, a family of exotic materials known as MXenes<sup>[108]</sup>, which are based on metal carbides or nitrides, is being explored for TE applications. Notably, Mo-based MXenes have experimentally demonstrated *PF*s of 0.3 mW·m<sup>-1</sup>·K<sup>-2</sup> at elevated temperatures (800 K)<sup>[109]</sup>.

Thus, 2D materials could have the potential of providing insight into novel physical effects and hold some promise for high performance on par with or even exceeding that of bulk materials<sup>[110]</sup>. If in the future they overcome challenges related to scalability, stability, *etc.*, they could be at the forefront of the TE technology, particularly in terms of low-power, flexible technologies that enable devices for the IoT and new device concepts to be developed. They will complement the work on films for flexible, room temperature IoT applications, which is also picking up. For this, some promising 2D materials possess the potential to unveil novel physical phenomena and provide high TE performance. The ongoing research in this area is yielding promising results, paving the way for advancements in flexible TE films aimed at room temperature IoT applications<sup>[111,112]</sup>.

## BAND STRUCTURE ENGINEERING

Some of the most promising directions for *PF* improvement in TE materials are summarized by the term “valleytronics”<sup>[113]</sup>. This involves the utilization of materials with certain features that the valleys in their electronic structure should have, or the engineering of electronic structures to achieve valleys that meet certain criteria. Below we describe these optimal criteria and then some of the efforts undertaken for achieving those through materials engineering strategies. Some examples are shown in Figure 4.

### Electronic structure optimal features for TEs

To anticipate at first order if the band structure of a given material will allow for a desirable TE *PF*, quality factors (or descriptors) have been established. One of the most commonly used is the quality factor,  $B \propto \mu_w/\kappa_{\text{lat}}$ , which consists of the weighted mobility,  $\mu_w = \mu \cdot N_v \cdot (m_b^*)^{1.5}$  which is the product of the electron mobility ( $\mu$ ), the carrier band mass ( $m_b^*$ ) and the number of equivalent bands in the Brillouin zone ( $N_v$ )<sup>[114-117]</sup>. Since the mobility is in general a quantity that cannot be identified from the band structure before accurate transport calculations or matching to experiment, another descriptor identified is  $N_v \varepsilon_r / D^2 m_c$  where  $N_v$  is the number of valleys per band,  $\varepsilon_r$  is the dielectric constant of the material,  $D$  is the dominant deformation potential, and  $m_c$  is the conductivity effective mass<sup>[118]</sup>. Most of these quantities are dictated by the band structure.

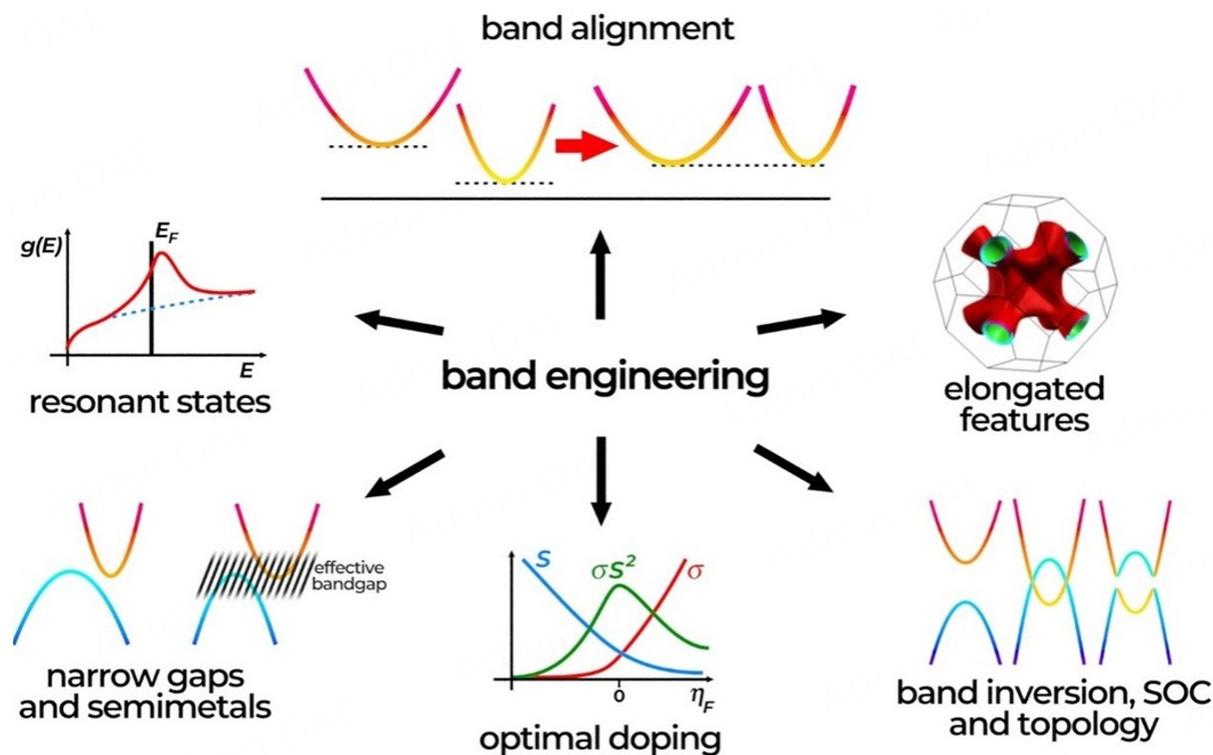
It has been identified in many studies that for optimal *PF*s, the Fermi level needs to reside very close to the band edge of the material. In fact, optimal doping is not always achieved in TEs, but it is an important direction that can allow large *PF* improvements<sup>[119]</sup>. If a material with its Fermi level placed at its band edge

is realized, then many valleys (or transport carrier pockets) would provide increased carrier density and conductivity. Another way to appreciate the importance of the number of valleys, is when examining the material at a fixed carrier density. In this case, a large band degeneracy increases the overall density-of-states effective mass ( $m_{\text{DOS}}$ ) and improves the Seebeck coefficient (the reason being that at a fixed carrier density the Fermi level will be positioned lower if  $m_{\text{DOS}}$  is higher). On the other hand, a low conductivity effective mass ( $m_c$ ) leads to high carrier mobility and conductivity. The favorable situation will be many light valleys for large  $m_{\text{DOS}}$  and low  $m_c$ . An additional important aspect is the degree of inter-valley scattering between these many valleys. This is desired to remain low, because the additional scattering could negate the increase in carrier density and conductivity that large  $N_v$  brings<sup>[120,121]</sup>. In many TE materials, the dominant scattering mechanisms are polar optical phonons and ionized impurity scattering (IIS), since TE materials are heavily doped and many are polar. Fortunately, both scattering mechanisms are anisotropic and their strength decays fast with the distance between the valleys in momentum space. Thus, ideally, the optimal TE material electronic structure will consist of: (i) many bands and valleys per band packed closely in energy, (ii) but as further apart as possible in the k-space of the Brillouin zone<sup>[118,122]</sup> (a definition of  $N_v$  as the average number of valleys per band captures some of this<sup>[118]</sup>), (iii) with light conductivity effective masses, and (iv) large dielectric constant values for enhanced screening and reduction of Coulomb and polar phonon scattering. Such behavior is typically encountered, for example, in most of the materials that exhibit large *PFs*, such as half-Heusler materials (primarily p-type), chalcogenides, some Zintl phases, silicides, and others. Note that at high carrier densities exceeding  $10^{20} \text{ cm}^{-3}$ , screening effects are typically strong and reduce the strength of polar optical phonons. Typically, screening does not become strong enough to do the same for IIS, since IIS also increases with the density of ionized dopants, which is equal to the carrier density. On the other hand, these are anisotropic (small angle) scattering mechanisms, and at elevated densities and Fermi levels, where the Fermi surfaces are large, they only scatter carriers locally in the Brillouin zone, according to the DOS in the vicinity of the initial state, thus not proportionally to the entire DOS. This could make their relative strength reduced compared to non-polar optical phonon scattering, for example, especially at higher temperatures. In other materials such as PbTe with ultra-high dielectric constants, the IIS strength can be diminished, since the scattering strength is inversely proportional to the square of the dielectric constant<sup>[123]</sup>.

Another band structure-related feature that allows for large *PFs* in complex materials is the presence of complex shaped, elongated energy surfaces in the electronic structure (see Figure 4). Compared to isotropic spherical, or even ellipsoidal bands, elongated, non-parabolic and highly anisotropic features can deliver higher *PFs* as they can provide simultaneously light effective masses for large electrical conductivity and heavy masses for enhanced Seebeck coefficient. In the limiting case, largely elongated tube-like bands of seemingly lower dimensionality can also provide possibilities for *PF* improvements and are actively investigated.

### Band alignment materials engineering

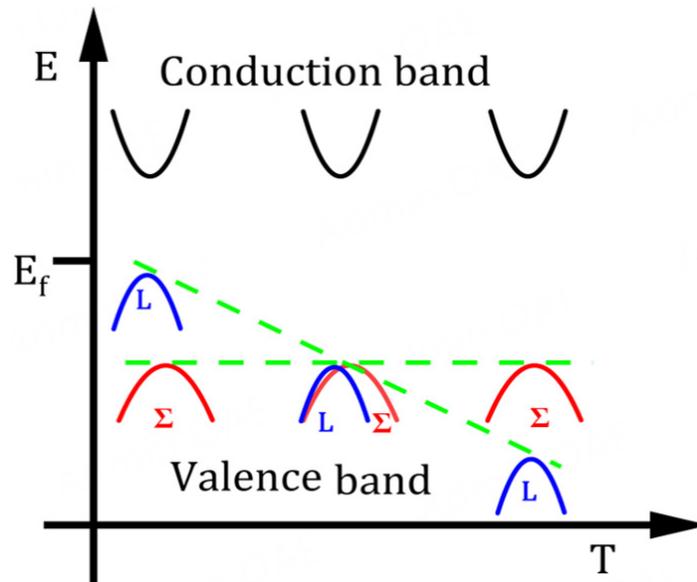
For materials with many valleys, but misaligned in energy, significant effort is devoted to properly aligning these valleys using iso-electric alloying (referred to as band alignment or convergence), with very promising results across materials<sup>[124]</sup>. This is a major band engineering optimization direction currently undertaken (see Figure 4). Many promising examples of materials whose *PF* benefits upon band alignment can be found in the literature with 10%-50% *PF* improvements<sup>[120]</sup>. An effective strategy to achieve band convergence at proper compositions is to use compounds in solid solutions with different band ordering. One example is the solid solution of  $\text{Mg}_2\text{X}$ , where X can be Si or Sn. These materials present a conduction band with a dual-band electronic structure (with one heavy and another light band) with an inverted band order for  $\text{Mg}_2\text{Si}$  and  $\text{Mg}_2\text{Sn}$ . A mixing of the two compounds, around composition  $\text{Mg}_2\text{Si}_{0.35}\text{Sn}_{0.65}$ , makes the band edges of the light and heavy conduction bands coincide<sup>[125]</sup>. It is very popular to use solid solutions in thermoelectricity to reduce the lattice thermal conductivity, but equally important is to examine the band



**Figure 4.** Schematics for a few dominant and frequently encountered band structure engineering strategies (clockwise from top): (i) band alignment for increasing the valley degeneracy; (ii) elongated energy surface features; (iii) band inversion followed by spin-orbit-coupling (SOC) for increasing degeneracy and operation close to topological states; (iv) optimal doping; (v) narrow gap materials engineering; (vi) resonant states for Seebeck improvement.

structure of the parent compounds and establish opportunities for band convergence. For this, understanding and following the rules of bonding chemistry plays a significant role. For example, in Mn-doped PbTe a smaller energy difference  $\Delta E$  between the L and  $\Sigma$  pockets is achieved due to the anti-bonding of Te- $p$  and Mn- $d$  orbitals, resulting in the second  $\Sigma$  pocket being pushed upwards<sup>[126,127]</sup>. Half-Heusler alloys also have multiple pockets in their band structure and are prone to such favorable alloying<sup>[30,128]</sup>. Moreover, it has been shown that the high TE performance in certain materials is achieved from band convergence with temperature. For example, the high performance of CoSb<sub>3</sub> skutterudites arises from the convergence of a secondary conduction band with 12 carrier pockets at high temperatures, rather than from the previously assumed linear band behavior<sup>[129]</sup>. It is also predicted that the prominent PbTe will also show band alignment of the L and  $\Sigma$  pockets with temperature as well, which will lead to better performance<sup>[130]</sup> (see Figure 5).

Other than forming compounds, band alignment can naturally occur in some multi-band materials at certain temperatures where the different temperature evolutions of the various bands could eventually cause their energy convergence. Materials, such as SnS with three valence band valleys, have temperature dependent valley positions and at certain temperatures alignment can occur which can also reflect on improved  $PFs$ <sup>[131]</sup>. This has also been observed in PbTe and similar rock salt IV-VI compounds and SnSe<sub>1-x</sub>S<sub>x</sub><sup>[32,128,131,132]</sup>. These compounds have non-parabolic bands with two or even three valley pockets energetically nearby. For example, Kim *et al.*<sup>[133]</sup> (and lately others<sup>[134]</sup>) used *ab initio* molecular dynamics simulations to observe the convergence of the L and  $\Sigma$  pockets in the Brillouin zone of PbTe at 450 K, further supported by the results of Pei *et al.* for PbTe where they observe the convergence above 700 K when



**Figure 5.** An illustration of the two-valence band (L and  $\Sigma$ ) convergence with increasing temperature in certain materials.

both lattice thermal expansion and electron-phonon interaction are considered<sup>[135]</sup>. However, more efforts need to be taken to achieve an in-depth understanding of the mechanism of temperature-induced band convergence.

Band alignment can also be achieved by tuning the material's structural parameters in certain material families, for example as shown for tetragonal chalcopyrite semiconductors. In cubic zinc-blende compounds, the valence band consists of  $\Gamma_{5V}$  and  $\Gamma_{4V}$  bands, which are split in energy by the crystal field with splitting  $\Delta_{CF} = \Gamma_{5V} - \Gamma_{4V}$  under a non-unity structural parameter  $\eta = c/2a$ <sup>[136]</sup>. However, it is shown that for  $\eta \approx 1$  (pseudo-cubic structure), band convergence can be achieved with an increase in the  $PF$ <sup>[137]</sup>. This can be achieved through compound optimization by tuning the composition, doping and solid solutions between compounds, which modifies  $\eta$  towards unity. It has been experimentally shown in ternary and quaternary chalcogenides<sup>[137,138]</sup>.

When attempting to reach band alignment, however, details regarding the strength of the inter-valley scattering and the details of the aligned bands need to be considered. For example, theory has shown that in the presence of strong inter-valley scattering, band alignment is only beneficial to the  $PF$  when a light mass valley is brought closer to the band extrema to participate in transport (whereas a heavy mass valley can be detrimental). Furthermore, in that case, the bands should not be fully aligned, but an optimal (but small) separation must exist for optimal  $PF$  conditions<sup>[120]</sup>. In the case of weak inter-valley scattering, then any band that is brought into the transport energy window provides  $PF$  benefits. Finally, care needs to be taken such that the modifications to the band structure upon band alignment do not alter the masses of the valleys by making them heavier, which will harm the mobility.

### Semimetals and narrow bandgap bipolar thermoelectric materials engineering

The bandgap is a crucial component for the TE performance of materials because it creates the necessary anisotropy in the DOS that the Seebeck coefficient requires. TEs require large carrier densities and low effective masses, which are favored by small bandgaps; thus, the interplay between those quantities and some finite bandgap provides some of the best performance TE materials. Lately, however, many reports

have appeared in the literature on large  $PF$  performance in semimetals or materials with small bandgaps. Most of these cases (that we describe below) have large  $PF$ s, but also large thermal conductivities and only moderate  $zT$ s (with values not large enough to make it in the plot of Figure 1). Still, high  $PF$  materials could be more significant in situations of power generation from abundant heat sources, whereas efficiency could be less important.

A prominent example that has raised significant attention is the full Heusler  $Fe_2VAl$ . In reference<sup>[34]</sup>, for example, an exceptional  $PF$  was reported, originating from a very high Seebeck coefficient, and it is claimed that the n-type  $Fe_2VAl$  could provide  $zT$  even up to 5. This material is claimed to be semi-metallic from *ab initio* calculations, although a large degree of its behavior can be understood by assuming a small bandgap of 0.02 - 0.04 eV. In fact, with such a small bandgap, due to the large concentration of intrinsic defects (of the order of  $10^{20} \text{ cm}^{-3}$ ), nominally undoped samples appear metallic<sup>[56]</sup>. Nevertheless, in Refs.<sup>[139,140]</sup>, it was shown by density functional theory (DFT) calculations that partial Ti or Ta and Si co-substitution in  $Fe_2V_{1-x}Ti_xAl_{1-y}Si_y$  and  $Fe_2V_{1-x}Ta_xAl_{1-y}Si_y$  drives an opening of a pseudo-gap, which could explain the large experimental measurements of exceptionally large thermopowers and  $PF$ s of the order of  $7.3\text{-}10.3 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$  at room temperature. Such large  $PF$ s are reported in a series of reports lately, stressing the direction of opening small bandgaps in semimetals to improve  $PF$ s<sup>[56,139,140]</sup>. It is worth mentioning that stabilizing the  $L_{21}$  ordered phase significantly enhances the TE performance of  $Fe_2VAl$ , exhibiting a two-fold increase in the Seebeck coefficient compared to its disordered B2/A2 phases. This improvement is attributed to the flattening of the DOS and the widening of the bandgap induced by the disorder-order transition<sup>[141]</sup>. We note here that half-Heuslers have some of the highest  $PF$ s of all TE materials and using these techniques they can be improved even more. The reason this happens is that in those cases, the DOS(E) increases sharply near the bandgap, which allows for very large Seebeck coefficients at significant conductivity. Note also that there are several metallic/semi-metallic Heuslers that could offer large possibilities for materials exploration. Recent works that have demonstrated exceptionally high  $PF$ s<sup>[34,139,140]</sup> stress the importance of defect engineering to obtain a small effective bandgap, but which can also introduce impurity delta-function states as a result of Anderson localization transitions in the material induced by defects. These can result in effects like those observed in the presence of increased DOS due to resonant states (discussed below), which are beneficial for the Seebeck coefficient and the  $PF$ .

Another example of how semimetals and/or narrow bandgap materials can provide very high Seebeck coefficients and  $PF$ s is when a large degree of anisotropy in transport exists between the conduction and valence bands<sup>[118,142,143]</sup>. Typically, bipolar transport degrades the Seebeck coefficient which undergoes a zero crossing when the material changes polarity from n-type to p-type when changing the Fermi level position. Furthermore, when the Fermi level is in the bandgap and near the charge neutrality point, where the material is intrinsic (undoped), the carriers experience phonon-limited mobility, which is higher compared to the mobility under heavy doping conditions. In addition, bipolar contributions from both electrons and holes increase the conductivity. Advanced theory and simulations have shown that if quantities such as the DOS, mobility, scattering rates, effective masses, etc., differ significantly between the conduction and valence bands, then the zero crossing of the Seebeck coefficient and the charge neutrality point of high mobility/conductivity occur at different Fermi level positions<sup>[143]</sup>. This allows for a finite or even high Seebeck at the high conductivity charge neutrality point, and exceptionally high  $PF$ s can be realized, even by an order of magnitude higher compared to the unipolar values. Since the electronic thermal conductivity is also large in the bipolar regime, the  $zT$  is shown to only increase at most by a factor of 2, which is still significant, while this effect is more evident at higher temperatures<sup>[143]</sup>. However, in general, bipolar conduction degrades performance and needs to be avoided, because it reduces the Seebeck coefficient and increases the electronic thermal conductivity through conduction of carriers of both polarities, in addition

to the bipolar thermal conductivity term (which can be thought as recombination of electrons and holes at the contacts of the material). For example, in narrow-gap materials such as SnSe, bipolar transport degrades the  $zT$  at elevated temperatures by introducing such counteracting hole/electron currents<sup>[144]</sup>. This material-level challenge necessitates bandgap widening via alloying in SnSe, thereby achieving a significant reduction in bipolar conduction<sup>[145,146]</sup>. Another bandgap engineering direction is related to 2D materials. Theoretical studies have shown that there exists an optimal value for the  $PF$  as the bandgap of these materials is reduced as a consequence of the bands becoming more linear, which benefits the electronic conductivity. For even lower bandgaps, however, the  $PF$  is reduced because bipolar effects gradually diminish the Seebeck coefficient<sup>[147]</sup>.

### Resonant states for Seebeck improvement

Another strategy that led to  $PF$  improvements in certain cases is the use of resonant levels which result from interactions between defects and host materials and could lead to an increase in the DOS. This happens in rare cases where certain defects can introduce resonance states and were first observed for TI defects in PbTe<sup>[148]</sup>. Typically, the electronic configuration of such dopants is very close to that of the host, as when using periodic table elements from neighboring columns for doping. The goal is to identify a dopant that near the Fermi level forms defect states. The main idea is to increase the DOS locally in energy, such that a broadened delta-function-like feature is added to the underlying DOS of the pristine material. The sharp increase in the DOS is translated into an increase in the Seebeck coefficient, as this coefficient is proportional at first order to the energy derivative of the DOS. As this extra DOS appears at elevated energies into the bands (and assuming that the Fermi level is placed in that vicinity), typically the electronic conductivity does not suffer significantly, or at least  $PF$  improvements can be observed. For instance, TI-doped PbTe exhibits a 3-5 $\times$  reduction in mobility compared to the single crystal, but shows a 1.7-3 $\times$  enhancement in the Seebeck coefficient<sup>[148]</sup>, resulting in an overall increase in  $PF$  and  $zT$ . Since the initial observation<sup>[148]</sup>, a few other cases have been reported. Some other examples are the use of IIIA elements for rock-salt IV-VI structures and functioning as p-type dopants<sup>[149-151]</sup> and the use of IVA elements in  $V_2VI_3$  compounds<sup>[152]</sup>. Additionally, cases such as Sb-doping on the Te-site in  $CuGaTe_2$ <sup>[153]</sup>, Pb doping on the Bi-site in  $BiCuSeO$ <sup>[154]</sup>, and even anti-site defects in  $ZrNiSn$ <sup>[155]</sup>, are known for the formation of band edge resonant levels. More lately such effects arising from defect states in Heusler alloys are claimed to provide significant  $PF$  improvements<sup>[139,140]</sup>.

### Other promising and exploratory directions for PF improvements

Other than the mainstream directions for  $PF$  improvements described above, it is worth mentioning exploratory examples based on novel physical phenomena.

One such direction is the increase of band degeneracy in materials with inverted bands in the presence of spin-orbit-coupling (SOC) (see [Figure 4](#)). Upon band inversion beyond a critical degree, and under the influence of SOC, a bandgap opens, accompanied by an increase in the number of carrier pockets, the band degeneracy, and as a result it could largely increase the  $PF$ . Such effects have for example been predicted recently for rock-salt IV-VI compounds<sup>[156,157]</sup>. With regards to further utilization of SOC in TEs, recent reports also speculate that the Rashba SOC in 2D materials can modify the DOS distribution and result in lower-dimensional states, which could positively influence the Seebeck coefficient under certain cases<sup>[158]</sup>.

A second exploratory direction is that of topological effects, which have recently emerged with possibly large  $PF$  potential. Indeed, a lot of high-performance TE materials are also topological materials. Topological materials have demonstrated a variety of unconventional TE effects that can lead to very high performance. Topological insulators have surface states with almost ballistic conductivity, as backscattering is forbidden by time-reversal symmetry arguments. Although the metallic topological edge states should

eliminate  $S$ , on the contrary, very large  $S$  can be achieved upon SOC induced bandgap opening, further assisted by the low-dimensionality and band linearity<sup>[159,160]</sup>. This, in addition to the ballistic topological transport (protected from backscattering) which improves  $\sigma$ , could allow for unprecedented  $PF$ s. Furthermore, the presence of a non-zero Berry curvature under certain conditions allows the realization of a very large anomalous Nernst effect (ANE), which can also be modulated by gating<sup>[90,91,161-163]</sup>. These are novel phenomena, but at this point, it is not yet clear if they can provide larger performance compared to the conventional operation. A study on such a Heusler material heterostructure has speculated an incredible  $zT \sim 5$ , importantly as a result of its extremely high  $PF$  which reaches values as high as  $\sim 50 \text{ mW}\cdot\text{m}^{-1}\cdot\text{K}^{-2}$ <sup>[34]</sup>. Although not yet reproduced successfully, it could be that topological effects, or operating in the presence of those, could have a lot to offer to thermoelectricity. We mentioned a few exploratory directions, but we emphasize that this is not an exhaustive list; other novel physics-based effects are investigated, ranging from Anderson-type transport, atomic ordering<sup>[14]</sup>, to spin-Seebeck, to magnon-based effects, to techniques that soften the lattice and reduce thermal conductivity<sup>[164]</sup>, to utilizing  $4f$  electrons and their effect on the DOS<sup>[165]</sup>, to utilizing scattering of  $s$ -orbital electrons by  $d$ -orbitals in metals<sup>[166]</sup>, and many more. It yet remains to be seen if some of these will result in high  $PF$  performance, especially at room and elevated temperatures.

Finally, we need to stress that there are many important developments in materials with important performance features, but not high enough  $zT$  to make it into [Figure 1](#), and because of this they sometimes receive less attention. Other than the half-Heuslers mentioned earlier with very high  $PF$ s, but  $zT$  only around 1-1.5<sup>[167,168]</sup>, we would like to mention the case of  $\text{Mg}_3\text{Sb}_2$  and its derivatives<sup>[22,169-172]</sup>. This material has a  $zT$  of  $\sim 0.7$ , but importantly this is at room temperature, which is a range where not many candidates can challenge  $\text{Bi}_2\text{Te}_3$ . Such a non-Te base material composed of abundant elements can have a much larger impact on room-temperature applications.

Beyond energy conversion and cooling applications, significant interest for TE effects is also encountered in the field of spintronics, where spin currents or voltages need to be created. In particular, the spin-Seebeck effect-SSE (the spin analog of the Seebeck effect), discovered in 2008, has been explored for generation of “spin voltages” driven by thermal gradients (having the ability for spin injection into attached conductors). This gave birth to the field of “spin-caloritronics”. Unlike typical thermoelectrics, which are optimized for highly doped semiconductors, the SSE can be realized in a variety of magnets from - metals to semiconductors and insulators; thus, it could enable insulator-based thermoelectrics<sup>[173,174]</sup>. SSE was observed in both the transverse configuration, in which a spin current perpendicular to the temperature gradient can be measured, and the longitudinal configuration, in which a spin current flowing in parallel to the temperature gradient is created in the longitudinal direction; thus, this is referred to as the Longitudinal Spin Seebeck Effect (LSSE). The latter has been observed in paramagnetic metal/ferromagnetic insulator junctions, and due to its simplified structure is the mainstream SSE direction<sup>[175,176]</sup>. However, in this configuration, proximity effects from the ferromagnetic material can result in the creation of the ANE in the metal, in the same direction as the LSSE. Thus, when studying the SSE, separation of the two signals is an important task that is undertaken<sup>[177]</sup>. Only in magnetic insulators which do not develop the ANE, has the LSSE been measured unequivocally<sup>[177]</sup>. In general, from the application perspective, the output voltage of LSSE is much smaller than that of conventional devices, but it can be used for spintronic applications, and sensors.

Utilizing the diverse array of materials production and manipulation methods discussed above, a wide range of materials has been explored for TE applications, including Chalcogenides<sup>[178,179]</sup>, Skutterudites<sup>[180,181]</sup>, Zintl phases<sup>[182-184]</sup>, Clathrates<sup>[185-187]</sup>, Full-Heusler<sup>[139,188-190]</sup>, Half-Heusler<sup>[191-193]</sup>, Silicides<sup>[194,195]</sup>, Oxides<sup>[196-198]</sup>,

Chalcopyrites<sup>[199,200]</sup>, high-entropy alloys<sup>[201,202]</sup>, transition metal dichalcogenides<sup>[203-206]</sup>, Tetrahedrites<sup>[207,208]</sup>, composites<sup>[209-211]</sup>, and organic materials<sup>[212,213]</sup>, among others<sup>[214]</sup>. This diverse array of materials encompasses a wide range of structures and properties, aiming to enhance efficiency and performance across different temperature ranges<sup>[1,215]</sup>.

### Near-room temperature applications

$\text{Bi}_2\text{Te}_3$  and its alloys have been the benchmark for near-room-temperature TE applications for energy harvesting and cooling<sup>[216,217]</sup>, achieving a  $zT$  of  $\sim 1.4$  for  $\text{Bi}_2\text{Te}_3$  with various modifications around 300 K<sup>[19,20]</sup>. However, brittleness, instability under high-temperature gradients, high material costs, and concerns about toxicity, scarcity of the components, and environmental impact, have driven ongoing research into alternative materials. In this sense,  $\text{Ag}_3\text{Se}$  has emerged as a more cost-effective and environmentally friendly candidate, achieving near room temperature  $zT$  values  $\sim 1.2$ <sup>[218-223]</sup>. However, the challenge remains to enhance the stability and performance of these materials under operational conditions, particularly regarding thermal cycling, when operating at temperatures above the phase transition, and mechanical stress.

Other TE materials have also been tested for near-room temperature applications.  $\text{PbTe}$ <sup>[224-226]</sup>,  $\text{Sb}_2\text{Te}_3$ <sup>[178,227,228]</sup>,  $\text{Mg}_3(\text{Sb/Bi})_2$ <sup>[229,230]</sup>, chalcogenide  $\text{PbSe}$ <sup>[231]</sup>, and the half-Heusler  $\text{AgCuTe}$ <sup>[232]</sup> have shown promising potential for achieving high TE performance ( $zT$ ) near room temperature.

### Medium-temperature applications

For medium-temperature applications, materials such as Skutterudites and Heusler alloys have demonstrated significant potential, with  $zT$  values reaching  $\sim 2$ . Skutterudites, particularly those based on  $\text{CoSb}_3$ <sup>[233,234]</sup>, have been extensively studied for their ability to achieve high  $zT$  values due to their complex crystal structures, which facilitate low thermal conductivity while maintaining good electrical conductivity. Similarly, both half and full Heusler alloys have shown considerable potential in this temperature range, with several compositions yielding  $zT$  of  $\sim 1.5$ <sup>[58,167]</sup>. Another well-known material in this temperature range is  $\text{Mg}_2\text{Si}$ , which boasts the advantages of being low-cost and non-toxic, showing a  $zT$  of approximately 1-1.5 in several studies<sup>[235-237]</sup>.

However, a notable challenge faced by these materials is bipolar conduction, which adversely affects their performance at elevated temperatures. Unlike narrow-bandgap semiconductors or semimetals that excel at lower temperatures, the presence of both holes and electrons in these materials leads to a reduction in the Seebeck coefficient and an increase in thermal conductivity. This interplay ultimately results in a deterioration of the overall  $zT$ .

In addition to Skutterudites and Heusler alloys, other materials such as  $\text{GeTe}$ <sup>[238]</sup>,  $\text{Cu}_2\text{Se}$ <sup>[239]</sup> and  $\text{PbTe}$ <sup>[240]</sup> have also demonstrated  $zT \sim 2$  in the medium-temperature range. Among these,  $\text{SnSe}$ <sup>[18,241-244]</sup> has garnered significant attention due to its exceptional TE performance, achieving  $zT$  values exceeding 3.1 for polycrystalline  $\text{SnSe}$ <sup>[12]</sup>, underscoring its potential as a simple and cost-effective binary compound with increasing high-performance capabilities. This remarkable performance is attributed to its ultra-low thermal conductivity.

Furthermore, the semiconductor nature of  $\text{SnSe}$  allows for tunability in both p-type<sup>[245,246]</sup> and n-type<sup>[46,247-249]</sup> conduction through the introduction of external dopants, enabling the maintenance of a high  $zT$  across different compositions. However, the practical application of  $\text{SnSe}$  and other selenides is hindered by their sensitivity to degradation, particularly due to the oxidation of Sn, which necessitates extreme purity during

synthesis. This sensitivity poses substantial challenges for the scalability and long-term stability of SnSe-based TE devices, which must be, for example, encapsulated. Again, bipolar conduction, prevalent in narrow-bandgap semiconductors, further limits their effectiveness at elevated temperatures, highlighting the need for ongoing research into alternative materials and strategies to mitigate these issues.

### High-temperature applications

High-temperature TE materials are essential for applications in remote power stations, deep-space spacecraft exploration powered by radioisotope TEG, *etc.* Oxides and silicides are predominant in this domain, i.e., metal oxides, including ZnO<sup>[250-252]</sup> doped with Al<sup>[253,254]</sup>, have been widely studied for such applications. Other oxides such as MgO, Cr/Mo/Ru/MnFe<sub>2</sub>O<sub>4</sub><sup>[197]</sup>, Ba<sub>1/3</sub>CoO<sub>2</sub><sup>[255]</sup>, cobalt-oxide<sup>[198]</sup>, and SrTiO<sub>3</sub>-based<sup>[256]</sup> have also garnered attention. There are non-oxidic materials like Cu<sub>2</sub>Se (which has demonstrated  $zT$  of  $\sim 2$  around 1,000 K<sup>[239]</sup>), while Cu<sub>2</sub>Se-BiCuSeO-graphene composites have shown exceptional potential, achieving a  $zT = 2.82$  at 1,000 K<sup>[257]</sup>, and Si-Ge<sup>[258]</sup> alloys have shown promise at high temperatures. However, these materials must endure significant thermal stress, which can induce mechanical failure and maintain chemical instability, which poses a challenge for their long-term performance. Si-Ge alloys have been shown to have long stability and have been used by the National Aeronautics and Space Administration (NASA) in the spacecraft Voyager, Galileo, Ulysses, Cassini, and New Horizons<sup>[258,259]</sup>. Although theoretical predictions suggest a thermoelectric  $zT$  exceeding 2 for Si-Ge systems<sup>[260]</sup>, experimental studies have reached approximately 1.84 at 1,073 K<sup>[28]</sup>.

It is also important to note that at very high temperatures, the strategy of nanostructuring may be compromised, as materials must endure extreme temperatures for extended periods. Table 3 provides a detailed list of the state-of-the-art TE materials and their performance according to their operational temperature.

## TRANSITIONING HIGH $zT$ THERMOELECTRIC MATERIALS TO DEVICES

The landscape of TE materials has evolved significantly over the past few decades due to advancements in phonon and electron transport engineering. While materials with  $zT > 2$  are now commonplace in laboratory settings, their translation into commercially viable TEGs requires addressing critical challenges at the device integration level. This section gathers critical insights to highlight the most significant challenges and considerations in this transition and future priorities.

**Scalability and processing challenges:** The first major bottleneck is scalability and manufacturing. Current synthesis methods for hierarchical nanostructures (e.g., ball milling, SPS) face limitations in producing homogeneous bulk materials at industrial scales. For instance, while PbTe-SrTe nanocomposites achieve  $\kappa_p \approx 0.5 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  in lab settings, batch-to-batch variability exceeding 15% persists due to incomplete SrTe nanoparticle dispersion during mechanical alloying<sup>[32,126]</sup>. Such variation in overall TE performance can also be noticed for other high performance materials such as Ag/Cu or Sn based selenides<sup>[239,270,271]</sup>. For example, SPS introduces anisotropic transport properties in materials such as BiSbTe alloys, resulting in conductivity variations by a factor of 2 between the perpendicular ( $\sigma_{\perp} \approx 900 \text{ S}\cdot\text{cm}^{-1}$ ) and parallel ( $\sigma_{\parallel} \approx 450 \text{ S}\cdot\text{cm}^{-1}$ ) directions relative to the pressing axis<sup>[272]</sup>. This anisotropy stems from grain alignment during uniaxial compression, where (001)-plane textures develop preferentially<sup>[272,273]</sup>. Fine-tuning and optimizing the processing parameters, such as temperature profile and pressure can help minimize the directional performance and potentially alleviate batch to batch variabilities.

**Materials selection:** Another significant factor is the selection of TE materials, which must exhibit optimal TE properties, including a high Seebeck coefficient, electrical conductivity, and low thermal conductivity. While traditional materials such as bismuth telluride (Bi<sub>2</sub>Te<sub>3</sub>) and lead telluride (PbTe) have been

**Table 3. State-of-the-art TE materials according to optimal temperature and their respective zT**

Temperature range	Materials	Details	zT	Ref.	
Near room	Bi <sub>2</sub> Te <sub>3</sub>	Bi <sub>1.8</sub> Sb <sub>0.2</sub> Te <sub>2.7</sub> Se <sub>0.3</sub> + 15 wt% Te	1.4	[19]	
		(Bi <sub>2</sub> Te <sub>3</sub> ) <sub>0.93</sub> (Ga <sub>2</sub> Te <sub>5</sub> ) <sub>0.07</sub> /Bi <sub>1.99</sub> Ga <sub>0.01</sub> Te <sub>3</sub>	1.5/1.2	[20]	
		Bi <sub>2</sub> Se <sub>0.02</sub> Te <sub>3</sub> Cu <sub>0.03</sub> (Cu <sub>2</sub> S) <sub>0.0125</sub>	1.6	[211]	
		BiSb <sub>0.95</sub> Ag <sub>0.05</sub> Te <sub>2</sub> Se	1.7	[261]	
	Ag <sub>2</sub> Se	Zn-doped β-Ag <sub>2</sub> Se	1.3	[220]	
		Ag <sub>2</sub> Se nanorod arrays	1.15	[221]	
		β-Ag <sub>2</sub> Se (porous polycrystalline bulk)	0.9	[222]	
		Ag <sub>2</sub> Se via PHRMS	1.2	[218]	
	Mg <sub>3</sub> Bi <sub>2</sub>	Te-doped Mg <sub>3</sub> Sb <sub>2</sub> single crystal	0.5	[171]	
		Mg <sub>3.2</sub> Bi <sub>1.5</sub> Sb <sub>0.498</sub> Te <sub>0.002</sub> Cu <sub>0.01</sub>	0.9	[229]	
	Mg <sub>3</sub> Sb <sub>2</sub>	Mg <sub>3</sub> Sb <sub>1.48</sub> Bi <sub>0.48</sub> Te <sub>0.04</sub>	0.6	[22]	
	Sb <sub>2</sub> Te <sub>3</sub>	Sb <sub>2</sub> Te <sub>3</sub> /MoS <sub>2</sub> Multilayer	2	[82]	
		Sb <sub>1.97</sub> Nb <sub>0.03</sub> Ag <sub>0.005</sub> Te <sub>3</sub>	0.6	[227]	
	Medium	AgCuTe	Sb <sub>2</sub> Te <sub>3</sub> + 15% Bi <sub>2</sub> Te <sub>3</sub>	1	[178]
			(AgCu) <sub>0.995</sub> Te <sub>0.9</sub> Se <sub>0.1</sub>	1.1	[232]
		SnSe	Polycrystalline SnSe	3.1	[12]
			Single crystals along the b-axis	2.6	[18]
		PbTe	Bi-doped SnSe single crystals	2.2	[46]
			Pb <sub>0.98</sub> Na <sub>0.02</sub> Te-8%SrTe	2.5	[32]
		PbSe	Pb <sub>0.97</sub> Na <sub>0.03</sub> Te-2%MgTe-0.75%GeTe	2.8	[224]
2.5% K-doped PbTe <sub>0.7</sub> S <sub>0.3</sub>			2.2	[240]	
Mg <sub>3</sub> Sb <sub>2</sub>		Cu <sub>0.005</sub> PbSe <sub>0.99</sub> Te <sub>0.01</sub>	1.7	[61]	
		Mg <sub>3.1</sub> Co <sub>0.1</sub> Sb <sub>1.5</sub> Bi <sub>0.49</sub> Te <sub>0.01</sub>	1.7	[170]	
CoSb <sub>3</sub>		Mg <sub>3</sub> Sb <sub>1.48</sub> Bi <sub>0.48</sub> Te <sub>0.04</sub>	1.65	[22]	
		(Sm,Mm) <sub>0.15</sub> Co <sub>4</sub> Sb <sub>12</sub>	2.1	[262]	
GeTe		CoSb <sub>2.875</sub> Te <sub>0.125</sub>	1.1	[233]	
		Yb <sub>0.35</sub> Co <sub>4</sub> Sb <sub>12</sub>	1.5	[263]	
ZrCoBi		Bi <sub>0.05</sub> Ge <sub>0.99</sub> Te	2.0	[238]	
		ZrCoBi <sub>0.65</sub> Sb <sub>0.15</sub> Sn <sub>0.20</sub>	1.42	[167]	
(Zr, Hf) NiSn		Ti <sub>0.5</sub> (Zr <sub>0.5</sub> Hf <sub>0.5</sub> ) <sub>0.5</sub> NiSn <sub>0.998</sub> Sb <sub>0.002</sub>	1.5	[58]	
		Bi <sub>0.94</sub> Pb <sub>0.06</sub> CuSeO	1.14	[154]	
AgSbTe <sub>2</sub>		AgSb <sub>0.94</sub> Cd <sub>0.06</sub> Te <sub>2</sub>	2.6	[14]	
		Mg <sub>2</sub> Si	Mg <sub>2</sub> Si <sub>0.785</sub> Sn <sub>0.2</sub> Sb <sub>0.015</sub>	0.95	[236]
Mg <sub>2.03</sub> (Si <sub>0.3</sub> Sn <sub>0.7</sub> ) <sub>0.993</sub> Bi <sub>0.007</sub>	1.3		[237]		
High	SiGe	Mg <sub>2.08</sub> Si <sub>0.364</sub> Sn <sub>0.6</sub> Sb <sub>0.036</sub>	1.5	[264]	
		Polycrystalline n-type SiGe alloys	1.84	[28]	
	Ba <sub>1/3</sub> CoO <sub>2</sub>	Si <sub>80</sub> Ge <sub>20</sub> B <sub>0.8</sub> (TaC) <sub>1</sub>	1.06	[258]	
		Ba <sub>1/3</sub> CoO <sub>2</sub> epitaxial film	0.55	[255]	
	Yb <sub>14</sub> MnSb <sub>11</sub>	Yb <sub>14</sub> MnSb <sub>11</sub> with 5% excess Mn	1.2	[265]	
		Yb <sub>14</sub> MnSb <sub>11</sub> prepared with MgH <sub>2</sub>	1.26	[266]	
	Cu <sub>2</sub> Se	Cu <sub>2</sub> Se - Co-coated B nanoparticles	2.23	[267]	
		Cu <sub>2</sub> Se-BiCuSeO-graphene	2.82	[257]	
	FeNbSb	FeNb <sub>0.88</sub> Hf <sub>0.12</sub> Sb	1.5	[30]	
	Pr <sub>3</sub> Te <sub>4</sub>	Pr <sub>2.74</sub> Te <sub>4</sub>	1.7	[165]	
	La <sub>3</sub> Te <sub>4</sub>	La <sub>2.2</sub> Ca <sub>0.78</sub> Te <sub>4</sub>	1.2	[268]	
	SrTiO <sub>3</sub>	SrTiO <sub>3</sub> -10 mol% La-10 mol% Nb	0.65	[269]	

extensively used, they face limitations at elevated temperatures, necessitating the exploration of alternative materials. Recent studies have highlighted the potential of materials such as germanium telluride (GeTe) and skutterudites, which have shown promising  $zT$  values in the medium-temperature range. However, the performance of these materials can be adversely affected by bipolar conduction at higher temperatures, which necessitates ongoing research into alternative compositions and doping strategies to enhance their TE performance.

Innovative methods of doping and structural optimization further augment performance. For example, entropy engineering, Te-capping, doping, *etc.*, in  $\text{Bi}_2\text{Te}_3$  has significantly reduced tellurium sublimation losses, contributing to minimal  $zT$  degradation after extensive testing<sup>[274-276]</sup>. The exploration of materials such as  $\text{Ag}_2\text{Se}$ ,  $\text{PbSe}$ ,  $\text{GeTe}$  and  $\text{Sb}_2\text{Te}_3$ , in conjunction with state-of-the-art skutterudites and Heusler alloys, indicates the move towards developing high-performance TE materials for various applications<sup>[219,227,238,277]</sup>.

**Interface engineering and diffusion barriers:** Another important bottleneck in enhancing device performance and stability lies within the interfaces between TE materials and electrodes. A primary challenge is preventing interdiffusion, which can lead to the formation of undesirable phases and subsequent performance degradation. Each interface must be tailored to the specific TE material to ensure optimal efficiency and reliability. Notably, interfaces within multilayered segmented TE legs necessitate extensive characterization and optimization to leverage high-performance materials such as  $\text{Bi}_2\text{Te}_3$ ,  $\text{PbTe}$ , and  $\text{Mg}_3\text{Sb}_2$ , among others, that have shown significant  $zT$  values<sup>[278,279]</sup>. These interfaces critically influence thermal and electrical contact resistance, directly affecting overall device efficiency. Therefore, strategies to mitigate interdiffusion and minimize electrical resistance while minimizing thermal losses are essential for improving device TE performance<sup>[280]</sup>.

Innovative interface engineering techniques have been utilized to address interdiffusion challenges. The application of diffusion barrier materials, such as Ni-P coatings or TiN/Mo in  $\text{PbTe}$ -based and  $\text{Bi}_2\text{Te}_3$ -based TE modules, has effectively prevented detrimental interdiffusion with solder materials, while also enhancing bonding strength and resistance to intermetallic formation, ensuring greater reliability at elevated temperatures<sup>[278]</sup>. Further, Co-P diffusion barriers have been employed to stabilize joints in  $\text{PbTe}$  TE materials, significantly reducing atomic interdiffusion and prolonging the device lifespan at high operational temperatures<sup>[281]</sup>. Additionally, titanium layers have demonstrated improved interfacial stability against copper diffusion in  $\text{Bi}_2\text{Te}_3$ , maintaining low contact resistivity and notable mechanical strength<sup>[282]</sup>.

Moreover, recent advancements have explored novel barrier materials like  $\text{Mg}_2\text{Ni}$ , which prove compatible with TE phases such as  $\text{Mg}_3\text{Sb}_2$ <sup>[280]</sup>. This approach minimizes interfacial stresses, thereby enhancing overall device performance<sup>[280,283]</sup>. Furthermore, ongoing research into metal-semiconductor interfaces continues to present promising avenues for optimizing TE efficiency.

**Soldering and bonding process:** Soldering and bonding processes used to connect TE materials to electrodes also present critical challenges. Traditional soldering materials may not provide adequate thermal and electrical conductivity; thus, the exploration of new soldering materials is essential to improve performance. The impact of soldering processes on the microstructure, including crystal orientation and grain boundaries, must be thoroughly assessed to understand how these factors influence the TE properties of assembled devices. Traditional Sn-based soldering materials often exhibit limited temperature stability. In contrast, Ag-based bonding materials serve as a viable alternative, utilized in both micron-sized flakes and nanoscale paste forms, yet they do carry cost and environmental concerns<sup>[284]</sup>. Recent developments of transient liquid phase bonding<sup>[285]</sup> offer low-temperature solutions while maintaining high-temperature

stability, alleviating the detrimental effects of thermal damage on the TE material and interface.

**Operational degradation:** The operational degradation of TE materials constitutes an additional challenge, driven by mechanisms such as elemental diffusion, self-diffusion, and thermal stresses<sup>[286]</sup>. Critical design considerations, including encapsulation techniques, have emerged as effective strategies for mitigating these degradation mechanisms. Encapsulation protects TE materials from environmental factors, such as oxidation, which can undermine their stability at high temperatures. For example, encapsulation strategies in Ag<sub>2</sub>Se-based flexible devices have been demonstrated to prevent the propagation of microcracks, significantly enhancing mechanical flexibility while maintaining performance integrity even after 2,000 bending cycles<sup>[287]</sup>. Moreover, to mitigate surface-related degradation and ensure long-term stability - such as oxidation and magnesium loss in Mg<sub>3</sub>Bi<sub>2</sub>-based systems - coating or encapsulation with chemically inert materials such as BN, MgO, and SiC has proven to be an effective approach<sup>[288]</sup>. Furthermore, advanced techniques, such as Mg-vapor annealing, bolster thermal stability by preventing adverse phase transitions<sup>[288, 289]</sup>. In similar vein, Sulfur infusion is reported to enhance both TE performance and stability of sulfides<sup>[290]</sup>. The application of Al<sub>2</sub>O<sub>3</sub> atomic-layer-deposited coatings of 50 nm for on semiconducting single-walled carbon nanotube has demonstrated impressive retention of 95% of conductivity and *PF* after 300 h in air, underscoring the need of encapsulating materials to enhance TE resilience<sup>[291]</sup>.

**Material degradation:** The degradation of TE materials during operation is another critical challenge. Several mechanisms contribute to this degradation, including elemental diffusion and self-diffusion within the TE materials and contacts, sublimation of dopants, moisture-induced degradation, structural defects accumulation, thermal stresses, chemical interactions with other materials in the TE system and the formation of intermetallic compounds. These can compromise the mechanical integrity and TE performance of the modules over time<sup>[292,293]</sup>. Furthermore, the coefficients of thermal expansion (CTE) of different materials in a TEG must be carefully matched to prevent mechanical stresses during thermal cycling. Mismatches can lead to cracking and delamination at the interfaces, significantly affecting the durability and performance of the device<sup>[294]</sup>. The design of the module must account for these differences to ensure long-term stability under operational conditions. Such considerations also hold true for high-performing Ag/Cu/Sn-based selenides, which exhibit excellent TE performance, but undergo several phase transitions at elevated temperatures. Sublimation of Se and the need to use an extremely pure environment to prevent oxidation are critically important to maintaining the high performance of these materials<sup>[12]</sup>.

**Mechanical stability:** The mechanical stability and reliability of TEGs under varying thermal and mechanical loads is a significant concern. The design must ensure that the modules can withstand harsh environments and repeated thermal cycling without failure. This requires a thorough understanding of the mechanical properties of the materials and the effects of thermal stress on the interfaces. Operational control mechanisms are essential for maintaining consistent performance in TEGs. Effective temperature regulation requires advanced control systems that can monitor and adjust operational parameters, such as temperature and electrical output. The integration of power management systems is also vital to ensure compatibility with external loads, which adds complexity to the overall system design.

**Integration with modern electronics:** The solid-state nature and versatility in size and shape of TE make them suitable candidates for integration into modern electronics and sensors<sup>[279,287]</sup>, as well as integration with other systems such as photovoltaic<sup>[295]</sup>, wearables<sup>[296,297]</sup>, and power electronics<sup>[298]</sup>, *etc.* For integration with power electronics and other applications, modern TEGs require co-design with wide-bandgap semiconductors (SiC, GaN) to handle maximum power point tracking (MPPT) at > 90% efficiency<sup>[299-301]</sup>. Moreover, heterogeneous integration using ultra-deep through-silicon vias (TSVs) reduces parasitic

resistance in Si-based devices<sup>[302]</sup>.

Finally, the performance testing and its optimization is something that must also be researched to be able to transition from high  $zT$  materials to high  $ZT$  devices. The performance of TEG modules must be rigorously tested under various conditions to ensure reliability and efficiency. This includes evaluating their performance, along with other parameters such as internal resistance, and overall stability of the modules under different atmospheric conditions and temperature cycles. While significant advancements have been made in the development of TE materials, each category presents specific challenges that must be addressed to facilitate their transition into the market. Ongoing research is essential to optimize these materials for stability, performance, and cost-effectiveness, ensuring their viability for commercial application.

Several examples of the transformation from materials to functioning TE devices - with increasing overall device efficiency and their key limitations - have been reported in the literature, as discussed in Table 4. Various materials such as  $\text{Mg}_3\text{Sb}_2$ <sup>[309]</sup>,  $\text{GeTe}$ <sup>[311]</sup>, and half-Heusler<sup>[318]</sup> phase compounds-based TE modules have demonstrated efficiencies exceeding 10% across diverse temperature gradients, which is higher than commercially available  $\text{Bi}_2\text{Te}_3$ -based modules. For example, NASA has been employing SiGe- and PbTe-based TE modules in various space exploration missions, with an efficiency of 6.3% at a temperature gradient of 700 K and 328 K for SiGe- and PbTe-based systems, respectively<sup>[323]</sup>.

Lastly, it is vital to consider the market positioning of TE modules in relation to competing technologies. From an energy generation perspective, alternatives such as internal combustion engines (ICEs), organic Rankine cycle (ORC) systems, photovoltaics (PV), fuel cells, and piezoelectric or triboelectric generators pose significant competition. For refrigeration applications, vapor compression systems (VCS), magnetocaloric refrigeration (MCR), phase-change materials, and microfluidic cooling are key rivals. Currently available TE systems, which exhibit an efficiency of around  $\sim 10\%$  as generators and a coefficient of performance (COP) of less than 1 as coolers, offer lower performance compared to ICEs, PV, and fuel cells. The same is true when compared to VCS and MCR. However, their solid-state nature, reliability, low maintenance, and size versatility make them viable for a wide range - from large-scale systems to microelectronic devices, including various autonomous and sensor applications. In a similar vein, cooling using thermoelectric coolers (TECs) offers higher precision and control, and their compact nature enables easy and scalable integration. In fact, integration of thermoelectrics with other sustainable solutions such as PV and ICEs has shown improvement in the overall system efficiency. It is also important to note that TEGs outperform piezoelectric and triboelectric generators, and TECs can be an effective cooling solution for microelectronics. It is worth mentioning that while TEGs excel in niche applications - such as wearables<sup>[296]</sup>, space power<sup>[306]</sup> - they face stiff competition from ORCs in waste heat recovery scenarios. Techno-economic analyses indicate that TEGs become commercially viable at heat fluxes exceeding  $5 \text{ W}\cdot\text{cm}^{-2}$ , where ORC systems struggle due to high maintenance costs<sup>[324]</sup>. Recent advances include the demonstration of a 600 W TEG prototype integrated into a BMW X6, highlighting progress toward the U.S. Department of Energy's  $\$1/\text{W}$  cost target<sup>[325,326]</sup> for automotive applications. In parallel, material innovations such as Cu/Ag co-doping in bismuth telluride alloys have been shown in laboratory studies to reduce tellurium content by up to 40%, potentially contributing to future cost reductions, while also working on element recovery after the car is not more in use. One promising solution lies in circular economy approaches, such as the electrochemical recovery of critical elements like copper, bismuth, tellurium, and antimony commonly used in TE modules<sup>[327,328]</sup>.

Therefore, the broader adoption of thermoelectric modules depends on a holistic approach that advances materials development and improves device integration, as discussed in detail above, while simultaneously addressing both cost-effectiveness, environmental sustainability and recyclability.

**Table 4. State-of-the-art thermoelectric module efficiencies ( $\eta$ ) across various temperature gradients ( $\Delta T$ ) and their associated limitations**

Material	$\eta$ (%)	$\Delta T$ (K)	Key limitations	Ref
Bi <sub>2</sub> Te <sub>3</sub>	8	230	Bi <sub>2</sub> Te <sub>3</sub> modules suffer from Te scarcity (€1100/kg) and degradation above 250°C	[80,303-305]
PbTe (Apollo mission)	6.3	328	High-cost; Both <i>elements</i> are toxic; limited temperature stability; Interfacial reactions	[306,307]
GeTe-Mg <sub>3</sub> SbBi	10	350	High materials cost (Ge/Te); toxicity (Te/Sb); Other considerations for MgSb- and GeTe-based systems as discussed below	[308]
Mg <sub>3</sub> Sb <sub>2</sub> -single leg	10	350	Stability issue at higher temperatures; Mg sublimates above 450°C, requiring encapsulation	[309]
GeTe-Cu <sub>2</sub> Te-PbSe- single leg	14	440	High Cost of Ge (€3500/Kg according to Aldrich webpage) and toxicity concerns (Ge/Pb/Te); Phase transitions at elevated temperatures	[310]
GeTe-based	13.3	506	Expensive; Te-toxicity; Diffusion of materials and electrodes; Service reliability concerns, as the secondary phases are observed	[311]
Ge <sub>0.89</sub> Cu <sub>0.06</sub> Sb <sub>0.08</sub> Te-Yb <sub>0.3</sub> Co <sub>4</sub> Sb <sub>12</sub>	12	545	Aging ingresses the interface resistivity. Phase transition in GeTe at elevated temperature	[312]
Mg <sub>3</sub> Sb <sub>2</sub> -MgAgSb	7.3	593 (hot side)	Brittleness; Oxidization and moisture sensitivity; Bipolar effect	[313,314]
PbTe-Bi <sub>2</sub> Te <sub>3</sub> cascaded module	12	590	Deterioration of the interfaces between the legs and electrodes; toxic and high-cost material	[315]
(Nb <sub>0.8</sub> Ta <sub>0.2</sub> ) <sub>0.8</sub> Ti <sub>0.2</sub> FeSb -Hf <sub>0.5</sub> Zr <sub>0.5</sub> NiSn <sub>0.98</sub> Sb <sub>0.02</sub>	8.3	655	Challenges in scaling up homogeneous materials for large-scale applications; dopant Ta hard to melt and diffuse.	[316,317]
(Nb, Ta, Ti, V) FeSb	15.2	670	Significant efficiency drops at lower temperature; Though stable contacts long term stability concerns.	[318,319]
SiGe (Voyager mission)	6.3	700	Low efficiency; Increasing cost of Ge; Prone to microcracks and mechanical failures; Interdiffusion of electrodes	[306,320-322]

## CONCLUSION AND OUTLOOK

In this paper, we have reviewed the progress of the thermoelectric research field and discussed the most promising future directions. We focused mainly on novel concepts that have been introduced over the years, which have allowed for the remarkable progress experienced. Over the last 20 years, the resurgence in thermoelectric research and the advances in the  $zT$  are attributed significantly to nanostructuring and the large reduction in the thermal conductivity that accompanies it. Advanced nanostructuring methods have been developed, in which defects are introduced in a pristine material at the atomic-, nano-, and macro-scale, scattering phonons of different mean-free paths and reducing the thermal conductivity across the phonon spectrum.  $zT$ s have more than doubled to values of  $zT \sim 2$ , not only for the traditional thermoelectric materials, but for many more material families, which entered the field due to this drastic reduction in thermal conductivity. Some of the best results were reached when care was taken not to reduce the electronic conductivity significantly, for example, by using iso-electronic doping/alloying and nano inclusions whose band edges are aligned with the matrix material. Nanostructuring can also offer significant advantages to the power factor as well, utilizing the concept of energy filtering, which allowed unprecedented  $PF$  values in some cases that followed specific design guidelines. An even higher boost in the  $zT$  is on its way by utilizing band structure engineering techniques to improve the  $PF$  further. The most

studied method is the so-called band alignment (or band-convergence) in which multi-band electronic structures undergo specific alloying such that many bands are aligned in energy and all contribute to transport. Many other concepts are explored, such as resonant states, topological effects, defect engineering, and transport in low bandgap materials and/or semimetals. The exploration of these newly emerged physics-based concepts and their integration into different materials is only the starting point, and many exciting outcomes are expected in the future in the field of thermoelectrics.

However, the transition from thermoelectric materials to TEG devices presents its own set of challenges that must be addressed for successful commercialization of these materials with high efficiency. Key issues include interface engineering, where the complexity of interfaces between different TE materials can lead to increased thermal and electrical contact resistance, negatively impacting efficiency. Material degradation during operation, particularly due to chemical reactions with moisture and diffusion processes, can compromise the integrity and performance of the modules. Additionally, careful matching of thermal expansion coefficients is crucial to prevent mechanical stresses that can lead to cracking and delamination. The soldering and bonding processes used to connect TE materials to electrodes also present challenges, as traditional soldering materials may not provide adequate thermal and electrical conductivity. Finally, the mechanical stability of TEGs under varying thermal and mechanical loads must be ensured to withstand harsh environments and repeated thermal cycling. Addressing these challenges is essential for the successful integration of TEG technology into practical applications, and ongoing research is vital to optimize materials for stability, performance, and cost-effectiveness.

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### **Authors' contributions**

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### **Availability of data and materials**

Not applicable.

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### **Conflicts of interest**

All authors declared that there are no conflicts of interest.

### **Ethical approval and consent to participate**

Not applicable.

### **Consent for publication**

Not applicable.

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