



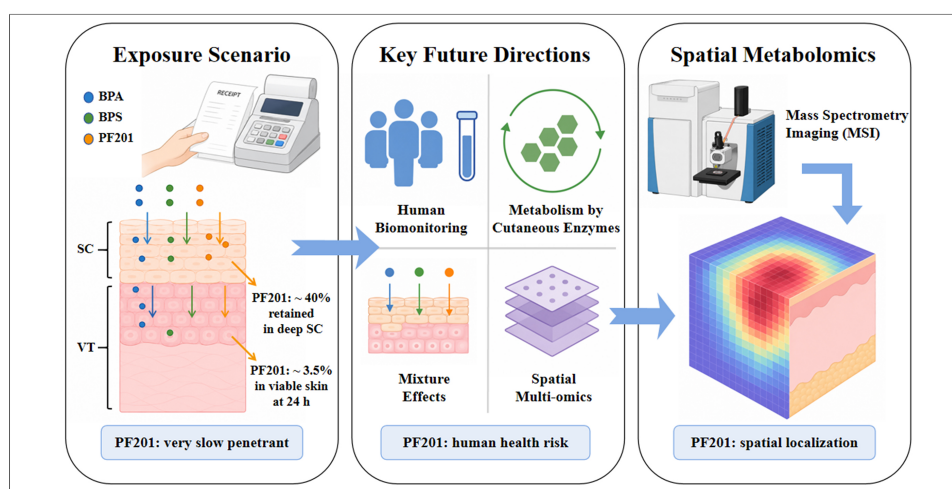
## Dermal exposure to Pergafast 201 in thermal paper: from absorption data to spatial metabolomics

Cairong Chen<sup>1</sup>, Yating Lin<sup>2,3</sup>, Yangen Huang<sup>4</sup>, Yingxin Yu<sup>2,3</sup>

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### INTRODUCTION

Bisphenol A (BPA), a prototypical endocrine disruptor, has received intense regulatory scrutiny over the past two decades. The European Union (EU) first imposed a ban on BPA in baby bottles in 2011<sup>[1]</sup>, and restriction limiting BPA to  $\leq 0.02\%$  in thermal paper in 2020<sup>[2]</sup>. As a result, it has driven a rapid market shift toward alternatives for BPA. Among the alternatives, bisphenol S (BPS), Pergafast 201 (PF201), bis(3-allyl-4-hydroxyphenyl) sulfone (TGSA), 2,4-bis(phenylsulfonyl)phenol (DBSP), and so on, have emerged as the color developer substitutes in thermal paper<sup>[3-7]</sup>. Especially for PF201, a non-phenolic urea derivative, it has rapidly increased and played a dominant color developer. In Switzerland, the share of PF201 in thermal paper soared from  $< 5\%$  in 2014 to  $> 60\%$



<sup>1</sup>Center for Reproductive Medicine, The Affiliated Qingyuan Hospital (Qingyuan People's Hospital), Guangzhou Medical University, Qingyuan 511518, Guangdong, China.

<sup>2</sup>Guangdong-Hong Kong-Macao Joint Laboratory for Contaminants Exposure and Health, Guangdong Key Laboratory of Environmental Catalysis and Health Risk Control, Institute of Environmental Health and Pollution Control, Guangdong University of Technology, Guangzhou 510006, Guangdong, China.

<sup>3</sup>Key Laboratory of City Cluster Environmental Safety and Green Development, Guangdong Basic Research Center of Excellence for Ecological Security and Green Development, School of Environmental Science and Engineering, Guangdong University of Technology, Guangzhou 510006, Guangdong, China.

<sup>4</sup>College of Chemistry and Chemical Engineering, Donghua University, Shanghai 201620, China.

**Correspondence to:** Prof. Yingxin Yu, Guangdong-Hong Kong-Macao Joint Laboratory for Contaminants Exposure and Health, Guangdong Key Laboratory of Environmental Catalysis and Health Risk Control, Institute of Environmental Health and Pollution Control, Guangdong University of Technology, Guangzhou 510006, Guangdong, China. E-mail: yuyingxin@gdut.edu.cn

by 2021<sup>[5,8]</sup>. Existing *in vitro* studies have shown that PF201 may interfere with endocrine pathways, exert potential immunomodulatory effects, and interact with multiple proteins implicated in a broad range of human diseases<sup>[9,10]</sup>. However, human exposure and toxicological data for the alternative lag far behind their market penetration.

Dermal contact is an important exposure route of BPA for occupationally exposed individuals such as cashiers<sup>[11]</sup>. Color developers in thermal paper are present as free (unreacted) monomers, readily transferable to fingers. A single contact can transfer 0.05–6.0 µg of BPA, BPS, and PF201<sup>[12]</sup>. More concerning, the bisphenols are not confined to thermal paper; paper recycling introduces them into newspapers, napkins, toilet paper, and they have even been detected in infant clothing, diapers, and feminine hygiene products<sup>[11,13,14]</sup>. Therefore, assessing the percutaneous absorption and skin metabolism of these alternatives is a prerequisite for precision risk assessment.

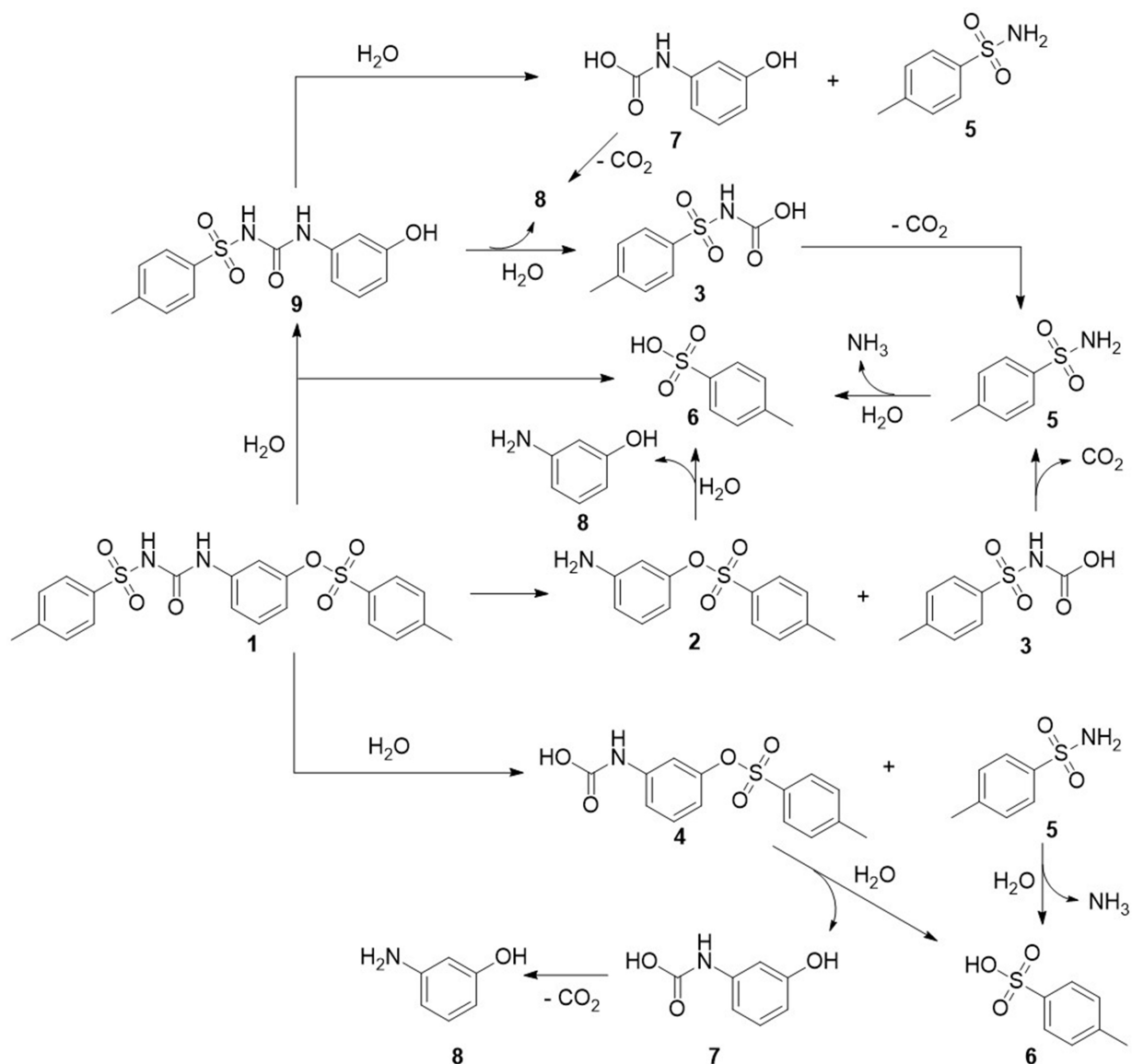
It is against this background that Tsykhotska *et al.* published “Assessment of Pergafast 201 absorption and metabolism in viable human skin: A comparative study with bisphenol A and bisphenol S” in *Environment International* in 2026<sup>[2]</sup>. Using tritium labelled compounds and the *ex vivo* human skin Franz diffusion cell system, the authors achieved for the first time a complete mass balance for PF201 (97.0% ± 5.9% recovery) and systematically compared the 24 h absorption and dermal retention of PF201 and compared with those of BPA and BPS<sup>[2]</sup>. Their study fills a critical data gap and provides key parameters for risk ranking of the alternative. The present commentary primarily highlights their research findings and contributions, and proposes possible future research directions.

## MAJOR CONTRIBUTIONS AND DISCUSSION

As reported by the authors<sup>[2]</sup>, mass balance calculations showed that all skin explants had recoveries falling within the 90%–110% range and the mean absorbed dose (the percentage of the applied dose reaching the receptor fluid) was 5.1% for BPA, 0.53% for BPS, and only 0.20% for PF201 at equal molar dose (1.39 nmol/cm<sup>2</sup>), which corresponded to 0.32, 0.35, and 0.64 µg/cm<sup>2</sup> on the skin surface, respectively. Also, four additional (lower) doses involving finite-dose conditions for PF201 were applied and the mean absorbed dose ranged from 0.15% to 0.31%. Dermal delivery (absorbed + skin retained) followed the same ranking: BPA > BPS > PF201. From dose-response and kinetic studies, the authors calculated PF201 permeability coefficients (K<sub>p</sub>) ranging from 0.63 × 10<sup>-6</sup> to 2.0 × 10<sup>-6</sup> cm/h, which classifies PF201 as a “very slow” penetrant. The mass balance directly satisfies Organisation for Economic Co-operation and Development (OECD) guidelines and the permeability coefficient data are ready for input into physiologically based pharmacokinetic models.

Despite minimal PF201 reaching the receptor fluid (< 0.5%), approximately 3.5% of the applied dose remained in the viable skin explant, and about 40% was retained in the deep stratum corneum (tape strips 3–20). This suggests that repeated occupational exposure (e.g., cashiers handling dozens of receipts daily) could lead to progressive accumulation of the chemical in the skin. Such a reservoir effect has been reported for BPA and BPS<sup>[15,16]</sup>, while the higher lipophilicity (LogP: 2.6 vs. 1.2) and molecular weight (460.52 vs. 250.27 g/mol) of PF201 than BPS may prolong the residence time of PF201 in the skin. The authors assumed that PF201 could continue to release into the systemic circulation long after exposure ceases, which warrants longer-term *ex vivo* studies, physiologically based pharmacokinetic (PBPK) modelling, as well as human intervention studies to directly test this.

In their study, radio-HPLC profiling of skin extracts showed PF201 predominantly as the parent compound (mean: > 99%), with only trace (mean: 3.3%) *m*-aminophenyl tosylate (AMT, a known by-product) detected in 28.6% samples (two out of seven explants). No *N*-(*p*-tosyl)carbamic acid methyl ester (MTC) or



**Figure 1.** The molecular structure of PF201 and possible metabolites. PF201: Pergafast 201.

conjugated metabolites were found. Positive control experiments using <sup>14</sup>C-7-hydroxycoumarin confirmed the activity of phase II xenobiotic metabolizing enzymes including sulfotransferases (SULT) and UDP-glucuronosyltransferases (UGT) in all skin explants. The authors observed that PF201 undergoes no significant metabolism or degradation in skin. While this conclusion seems simplifying toxicokinetic modeling (only the parent compound reaches systemic circulation), two caveats need to be considered. First, tissue homogenization and extraction may dilute or destroy low abundance metabolites; radio-detection is sensitive while lacks structural information. Second, human skin expresses a broad array of phase I enzymes including cytochromes P450 (CYPs), epoxide hydrolase, esterases and amidases<sup>[17,18]</sup>. PF201, being a urea derivative, contains an amide bond that could theoretically be hydrolyzed by cutaneous amidases or esterases, yielding aromatic amine metabolites. Although the authors did not detect significant accumulation of AMT (compound 2), other products or intermediates might have been missed due to lack of radiolabelled standards [Figure 1]. Thus, no major pathway was observed under the current detection conditions for PF201 rather than absolute biological inertness as the authors indicated. Other possible metabolites also remain worthy of investigation.

## FROM MACROSCOPIC PARAMETERS AND CONVENTIONAL HOMOGENIZATION METABOLISM TO SPATIAL METABOLOMICS

This study provides a solid quantitative foundation for PF201 skin risk assessment. To truly understand its health implications, we think there are several directions require further exploration.

### (1) Human biomonitoring and exposure markers

*Ex vivo* data need *in vivo* validation. Biomonitoring studies in cashiers or other occupationally exposed populations should be conducted, measuring PF201 and its potential metabolites (e.g., hydrolysis product AMT, oxidative or glutathione conjugates) in urine and blood. Given the expected very low absorbed doses, highly sensitive high performance liquid chromatography-tandem mass spectrometry (HPLC-MS/MS) methods (detection limits at pg/mL) are required. Controlled human volunteer studies (hand exposure to thermal paper with timed blood/urine collection) would provide human toxicokinetic parameters and allow comparison with *ex vivo* data to validate the Franz diffusion cell extrapolation. In addition, data from other *in vitro* models, such as human hepatic *in vitro* models<sup>[19]</sup>, regarding the metabolic fate of PF201 would also be required in order to identify metabolites that can be screened for in a human biomonitoring study.

### (2) Systematic evaluation of PF201 metabolism by cutaneous enzymes

As aforementioned, multiple phase I and II enzymes are expressed in skin, with different profiles across skin layers (e.g., the lipophilic stratum corneum and the hydrophilic viable tissue). While the authors confirmed phase II activity using 7-hydroxycoumarin, the more relevant questions are whether PF201 can be transformed by phase I enzymes including cutaneous amidases, esterases or CYPs as aforementioned.

### (3) Mixture exposure and joint skin absorption by PF201 and other bisphenols

In addition, although the present study focused solely on PF201, thermal paper often contains BPA, BPS, and PF201 together, with variable ratios. Their study tested each compound singly, but under mixture conditions competitive penetration (sharing stratum corneum lipid pathways) or barrier disruption (one component altering skin integrity and facilitating penetration of others) could occur. The potential combined effects of dermal co-exposure to these bisphenols should therefore also be performed to calculate whether the absorbed fraction of each component differs significantly from single compound exposure.

### (4) Spatial metabolomics by spatial mass spectrometry imaging (MSI)

All the above directions rely on conventional homogenization extraction quantification methods. While precise, these methods completely lose spatial information on compound distribution within the heterogeneous tissue microarchitecture. Skin is a highly heterogeneous organ, with distinct microstructures that differ enormously in their permeability and retention properties. Spatial MSI is a revolutionary tool for skin exposure research and it might unlock this “black box”. Basically, matrix assisted laser desorption/ionization (MALDI-), secondary ion mass spectrometry (SIMS-), or desorption electrospray ionization (DESI-) MSI can directly detect and localize exposure compounds and their metabolites or metabolites of endogenous substances produced by the toxic chemical exposure on tissue sections at spatial resolutions, generating ion density maps<sup>[20-23]</sup>. For example, more than 1,500 metabolites in organ can be visualized in an untargeted analysis by a sensitive air flow-assisted desorption electrospray ionization (AFADESI-) MSI method<sup>[23]</sup>. The technique does not require radiolabelling and can simultaneously detect parent compounds and multiple metabolites. Spatial metabolomics can accurately identify and locate the spatial distribution of multiple metabolites including endogenous metabolic products and enable downstream metabolomic analysis resulted from PF201 exposure. The application of MSI technology can achieve a multi-dimensional correlation of the transdermal behavior, metabolic transformation, and toxic endpoints, such as skin function damage, effects on lipid metabolism<sup>[24-26]</sup>, for PF201 considering its

absorption in the skin although its absorption and metabolism are weak. Simultaneously, it can be compared with the toxicological data of BPA to explain that even with a lower skin absorption rate, long-term accumulation may still cause adverse health effects through low-dose long-term exposure making the risk discussion more complete.

Of course, MSI applied to skin exposure research is still in its early stages, facing challenges in sample preparation, ion suppression, and quantitative accuracy. Nevertheless, successful precedents exist, such as for drugs<sup>[27]</sup>. For example, Arancibia *et al.* reported a novel MSI-based automated workflow enables rapid visualization and evaluation of active pharmaceutical ingredient distribution across skin layers<sup>[28]</sup>. Sjövall *et al.* combined time-of-flight secondary ion mass spectrometry (TOF-SIMS) with structural imaging/quantitative absorption analysis to map the localization characteristics of endogenous lipids and exogenous carvacrol/ceramide in cross-sections of human skin, revealing that exogenous ceramide is mainly confined to the stratum corneum and exhibits a depth gradient<sup>[29]</sup>. For emerging contaminants like PF201, developing a standardized MSI workflow would itself be an innovative contribution. Specific applications of MSI to PF201 skin exposure research include: (1) Precise micro localization of PF201 in the skin; (2) *In situ* detection of low abundance metabolites, avoiding homogenization dilution, and visualization of dynamic processes; (3) 3D co-localization of mixture exposure such as BPA and BPS; (4) Integration with other spatial omics, and so on. I encourage the authors or other laboratories to establish such a workflow and apply it to comparative assessments of PF201 and other alternatives, thereby opening a new dimension for next generation skin toxicology and health risk assessment.

## CONCLUSIONS

Tsykhotska *et al.* provide, with rigorous design and complete data, the first comprehensive picture of the absorption kinetics and metabolic fate of PF201 in human skin<sup>[2]</sup>. They demonstrate that PF201 has much lower dermal absorption than BPA, yet it does penetrate and persist in the skin (mainly in stratum corneum). However, low absorption is not synonymous with no risk. The potential for long term skin reservoir accumulation, the likelihood of enhanced penetration under real world conditions, and the lack of systemic toxicological data all call for more studies. Future research should integrate human biomonitoring, systematic evaluation of skin enzyme-mediated metabolism, mixture effects with other bisphenols, and spatial MSI, which has the potential to elevate skin exposure studies from macroscopic mass balance to microscale fate maps, and associated adverse health risk. This would provide unprecedented insights into the fate of PF201 or more emerging contaminants, the situations encountered, the cells and enzymes encountered, and the toxic mechanisms of the toxicants using metabolomics.

## DECLARATIONS

### Authors' contributions

Methodology, literature collection and draft preparation: Chen, C.

Data collection: Lin, Y.

Metabolite prediction: Huang, Y.

Design, reviewing and editing: Yu, Y.

### Availability of data and materials

Not applicable.

### AI and AI-assisted tools statement

During the preparation of this manuscript, the AI tool DeepSeek (version V3) was used solely for language editing. The tool did not influence the study design, data collection, analysis, interpretation, or the scientific content of the work. All authors take full responsibility for the accuracy, integrity, and final content of the manuscript.

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### Conflicts of interest

Yu, Y. is an Editorial Board Member and a Guest Editor of the Special Issue "Topic: The Impact of Bisphenol Exposure on Human Health" of the journal *Journal of Environmental Exposure Assessment*. Yu, Y. was not involved in any steps of editorial processing, notably including reviewers' selection, manuscript handling and decision making. The other authors declare that there are no conflicts of interest.

### Ethical approval and consent to participate

Not applicable.

### Consent for publication

Not applicable.

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