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Predicting a novel two-dimensional BN material with a wide band gap

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Abstract

Based on density functional theory, a new two-dimensional boron nitride, *Pmma* BN, is proposed and studied in detail for the first time. The stability of *Pmma* BN is demonstrated using phonon spectra, ab initio molecular dynamics simulations at 300 and 500 K, and in-plane elastic constants. The orientation dependences of the Young's modulus and Poisson's ratio show that *Pmma* BN has large mechanical anisotropy. *Pmma* BN is an indirect band gap semiconductor material with a band gap of 5.15 eV and the hole and electron effective masses have high anisotropy. The electron carrier mobilities of *Pmma* BN along the *x* and *y* directions are similar, while the hole carrier mobility along the *y* direction is more than double that along the *x* direction. The band gap of *Pmma* BN remains indirect under the effect of uniaxial tensile strain and its adjustable range reaches 0.64 eV when the uniaxial strain is applied along the *x* direction. When uniaxial strain is applied along the *y* direction, the positions of the conduction band minimum and valence band maximum change. *Pmma* BN under uniaxial strain shows strong optical absorption capacity in the ultraviolet region. To explore its potential clean energy applications, the thermoelectric properties of *Pmma* BN are also investigated.



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Keywords: Two-dimensional boron nitride, band gap engineering, uniaxial strain, thermoelectric properties

INTRODUCTION

The first successful experiments on graphene^[1,2] opened a new research field, namely, two-dimensional (2D) materials. In recent years, 2D materials have attracted significant interest due to their unique structure and properties, as well as their potential application prospects. Therefore, the number of proposed and synthesized 2D materials is increasing every year^[3,4]. These materials have a variety of novel properties and phenomena and a wide range of application prospects. 2D materials exhibit a range of electronic properties, including metallic^[5], semiconductor^[6-9], insulator^[10], semimetal^[11], half-metal^[12-14] and superconductor^[15] properties. Since the electron orbitals of carbon atoms can hybridize in different ways, a large number of topological structures for 2D carbon, such as biphenylene^[16], graphdiyne^[17], carbon Ene-yne^[18], and other graphynes^[19,20], have been synthesized experimentally or predicted theoretically. The doping of nitrogen or boron atoms into carbon allotropes not only introduces new features^[21,22] but also generates new 2D systems^[23,24]. Similar to graphene, these 2D materials with different chemical compositions show unprecedented complementary properties, thereby making them cornerstones of many applications in the future^[25].

The molecular geometry, strength, and other properties of boron nitride and carbon allotropes are similar. Based on this, researchers realized a graphene-like boron nitride structure known as hexagonal boron nitride $(h-BN)^{[26]}$. Among the hexagonal, rhombic, cubic, and wurtzite BN phases, h-BN has the best stability at room temperature^[27]. h-BN is composed of an sp^2 bond network formed by the alternating arrangement of boron and nitrogen atoms. The B–N bond length in h-BN is 1.44 Å and the C-C bond length in graphene is $1.42 \text{ Å}^{[28]}$. However, the application of monolayer h-BN in electron transport and energy conversion is severely limited because it is a semiconductor with a band gap of ~5.7 eV^[27]. Taking the structure of phagraphene as the model, Pontes *et al.* proposed a new material, phaBN, and studied its electron optical and thermodynamic properties using first-principles calculations^[29]. This 2D structure is formed by boron and nitrogen atoms that are only bonded by simple bonds with equal atomic numbers of the two elements. PhaBN has been found to have possible optical and electronic applications. Furthermore, Li *et al.* predicted a novel 2D boron nitride material, di-BN, and showed that it could generate hydrogen molecules in a photocatalytic water cracking reaction. In addition, its carrier mobility is significantly higher than that of other popular 2D semiconductors, thereby giving di-BN significant application potential in the fields of electron and solar energy conversion^[30].

A novel 2D boron nitride, namely, *Pmma* BN, is proposed in the present work. We verify the structural stability of *Pmma* BN from a variety of angles and carefully study its mechanical, electronic, and thermoelectric properties using first-principles calculations.

THEORETICAL METHODS

First-principles calculations are employed in this work using the Vienna Ab initio Simulation Package $(VASP)^{[31,32]}$. The projected augmented wave pseudopotential is used to represent the electron-ion interaction and the Perdew–Burke–Ernzerhof functional^[33] is used to describe the exchange-correlation potential in the structural optimization. All simulation cells contain a 15 Å thick vacuum layer to obtain 2D models. The Broyden–Fletcher–Goldfarb–Shanno minimizer^[34] is also used for the unit cell optimization. The plane wave cutoff energy is set to 520 eV and a 5 × 7 × 1 Monkhorst Pack grid is used to sample the Brillouin region to ensure the convergence of the results. The electronic band structure, effective mass, and carrier mobility are obtained using the hybrid Heyd–Scuseria–Ernzerhof functional^[35]. Density functional

perturbation theory and the Phonopy $code^{[36]}$ are used to calculate the second-order force constants and phonon frequencies. Using a 3 × 4 × 1 supercell and a canonical ensemble^[37], ab initio molecular dynamics simulations (AIMD) are performed for 4 ps with a time step of 1 fs at 300 and 500 K, respectively. The thermoelectric properties of *Pmma* BN under uniaxial strain are calculated using Nanodcal^[38].

RESULTS AND DISCUSSION

Figure 1A shows the optimized atomic structure of the novel 2D BN in the *Pmma* phase. B and N atoms are represented by purple and pink spheres, respectively. It is clear from Figure 1A that *Pmma* BN consists of four-, six- and ten-membered rings of B and N atoms. The lattice constants of *Pmma* BN are a = 9.13 Å and b = 6.77 Å after geometric optimization [Table 1]^[39,40]. In addition, the conventional cell contains ten B and ten N atoms, with a B–N bond length of 1.47 Å, which is slightly larger than the value of 1.45 Å in h-BN. To further analyze the bonding properties of *Pmma* BN, the electron localization function (ELF)^[41,42] is shown in Figure 1B. Obviously, due to the localized electrons between B and N atoms, the B–N bond in *Pmma* BN is a polarized covalent bond and the high value of the bond center indicates a strong covalent bond. However, the results show that electrons are more inclined to N atoms.

The stability of the novel 2D BN in the *Pmma* phase is demonstrated by phonon spectra, AIMD simulations and elastic constants. The phonon spectra of *Pmma* BN are shown in Figure 2A, with no imaginary frequencies exhibited, thereby proving its dynamic stability. We next investigated the thermal stability of *Pmma* BN and AIMD simulations were performed with a $3 \times 4 \times 1$ supercell of *Pmma* BN for 4 ps at 300 and 500 K. The results are illustrated in Figure 2B and the atomic configurations at the end of the 4 ps AIMD simulations illustrated in Figure 2C demonstrate that we find no significant structural distortions at 300 and 500 K and observe no B–N bond breakage during the simulations or phase transitions. This indicates that *Pmma* BN is thermally stable up to at least 500 K.

This work obtains the in-plane elastic constants C_{11} , C_{22} , C_{12} and C_{44} of 2D materials by the second partial derivative of the elastic energy in Equation (1)^[43]:

$$U = \frac{1}{2}C_{11}\varepsilon_{xx}^{2} + \frac{1}{2}C_{22}\varepsilon_{yy}^{2} + C_{12}\varepsilon_{xx}\varepsilon_{yy} + 2C_{44}\varepsilon_{xy}^{2}$$
(1)

The three-dimensional plot of the strain-energy surface is shown in Figure 3 and the values of the calculated in-plane elastic constants of *Pmma* BN are displayed in Table 1. The elastic constants of 2D crystals should satisfy Equations (2) and (3)^[44]:

$$C_{44} > 0$$
 (2)

$$C_{11}C_{22} > C_{12}^2 \tag{3}$$

Clearly, the fact that these values satisfy the mechanical stability criteria demonstrates that *Pmma* BN is mechanically stable. According to the in-plane elastic constants of a 2D material, the in-plane Young's modulus (*E*) and Poisson's ratio (ν) along an arbitrary direction θ (angle referring to the *x* direction) can be obtained by Equations (4) and (5)^[43]:

$$E(\theta) = \frac{C_{11}C_{22} - C_{12}^2}{C_{11}\alpha^4 + C_{22}\beta^4 + \left(\frac{C_{11}C_{22} - C_{12}^2}{C_{44}} - 2C_{12}\right)\alpha^2\beta^2}$$
(4)

$$v(\theta) = \frac{\left(C_{11} + C_{22} - \frac{C_{11}C_{22} - C_{12}^{2}}{C_{44}}\right)\alpha^{2}\beta^{2} - C_{12}\left(\alpha^{4} + \beta^{4}\right)}{C_{11}\alpha^{4} + C_{22}\beta^{4} + \left(\frac{C_{11}C_{22} - C_{12}^{2}}{C_{44}} - 2C_{12}\right)\alpha^{2}\beta^{2}}$$
(5)

where $\alpha = \sin \theta$ and $\beta = \cos \theta$.

The orientation dependences of the Young's modulus and Poisson's ratio of *Pmma* BN are shown in Figure 4. The shape displayed in Figure 4 deviates greatly from the circle, which implies that *Pmma* BN has relatively large anisotropy with regards to the Young's modulus and Poisson's ratio. The largest Young's modulus appears in the *y* direction, while the smallest Young's modulus appears in the x + y direction. This indicates that *Pmma* BN deforms more easily along the x + y direction. Poisson's ratio is 0.23 along the *x* direction and 0.29 in the y direction, indicating that the expansion of the latter is smaller under tensile strain. For comparison, the calculated mechanical properties of the B₄N monolayer and h-BN are also shown in Table 1. The in-plane Young's moduli along the *x* and *y* directions of the B₄N monolayer are 153 and 268 N/m, respectively, exposing a stronger anisotropy than *Pmma* BN. However, the Young's modulus and Poisson's ratio along the *x* and *y* directions of h-BN are equal, indicating that it is isotropic.

We next analyze the electronic properties of *Pmma* BN through the band structure, band decomposed charge densities at the conduction band minimum (CBM) and valence band maximum (VBM) and the effective mass anisotropy of holes and electrons. Figure 5A exhibits the electric band structure and partial density of states (PDOS) of *Pmma* BN. The CBM is located between the G point (0, 0, 0) and the B point (0.5, 0, 0), while the VBM is located between the F point (0, 0.5, 0) and the G point. The results show that *Pmma* BN is an indirect band gap semiconductor material with a band gap of 5.15 eV. Although the G point may also appear to be the VBM, it is actually 0.015 eV lower in energy than the VBM between the F and G points. As shown in the PDOS, the CBM of *Pmma* BN mainly consists of B-*pz* orbitals, while the VBM mainly consists of N-*pz* orbitals.

The decomposed charge density at the CBM and VBM is shown in Figure 5B. The charge density is located around the B and N atoms at the CBM and around the N atoms at the VBM. This suggests that the VBM is entirely dominated by the N atoms, while the CBM is mainly dominated by the N atoms. The effective mass and mobility of carriers are important parameters for characterizing the carrier transport properties of semiconductor materials. Based on the effective mass approximation and the electroacoustic phonon scattering mechanism, the carrier mobility of 2D materials can be obtained by Equation (6)^[45,46]:

$$\mu_{2D} = \frac{e\hbar^3 C_{2D}}{k_B T m_e^* m^* E_1^2} \tag{6}$$

where C_{2D} is the in-plane elastic constant. The effective masses m^* of electrons and holes are obtained by Equation (7), which fits the band edge dispersions at the CBM and VBM, respectively^[47]:

$$\boldsymbol{m}^* = \hbar^2 \left(\frac{\partial^2 \boldsymbol{E}(\boldsymbol{k})}{\partial \boldsymbol{k}^2}\right)^{-1} \tag{7}$$

where $m_e^* = (m_x^* m_y^*)^{1/2}$ is the average effective mass and m_x^* and m_y^* are the effective masses in the *x* and *y* directions, respectively.

The effective mass anisotropy of holes and electrons is displayed in Figure 5C. The angle refers to the *x* direction and the variation factor of the hole and electron effective masses is 5. The hole and electron effective masses show high anisotropy, with a minimum in the *x* direction $(0.61/0.39 m_0)$ and a maximum in the *y* direction $(26.63/17.46 m_0)$. The hole effective mass is greater than the electron effective mass in both the *x* and *y* directions. The obtained results of C_{2D} , m_d^* , E_1 , and μ_d for electrons and holes along the *x* and *y* directions are shown in Table 2. E_1 is the deformation potential constant, which is obtained by the linear fits to the band-edge positions E_{CBM} and E_{VBM} are with respect to the vacuum energy E_{vacuum} as a function of lattice compression or dilatation. E_{1x} and E_{1y} represent the deformation potential constants along the *x* and *y* directions, respectively. The energy variations of the CBM and VBM with uniaxial strain using a step of 0.5% are presented in Figure 6. From Table 2, the carrier mobilities for electrons in *Pmma* BN are almost the same along the *x* and *y* directions, while the value for holes along the *y* direction is more than double the value along the *x* direction.

In addition, we investigate the effect of uniaxial tensile strain on *Pmma* BN. Table 1 lists the in-plane elastic constants for *Pmma* BN under uniaxial strain. For C_{11} , C_{22} , and C_{12} , the values gradually decrease as the uniaxial strain increases. In addition, the change in the elastic constant is more pronounced with increasing uniaxial strain along the *y*-axis relative to applying uniaxial strain along the *x*-axis. Applying uniaxial strain along the *y*-axis increases the value of C_{44} , whereas increasing the uniaxial strain along the *x*-axis irregularly changes the value of C_{44} and when 10% uniaxial strain is applied along the *x*-axis, the value of C_{44} becomes small with a value of only 5. Fortunately, the in-plane elastic constants of *Pmma* BN under uniaxial strain are consistent with the mechanical stability criterion of 2D materials and are all mechanically stable.

The fracture strength of *Pmma* BN is studied by calculating the variation in stress with uniaxial tensile strain. As seen from the strain-stress curves in Figure 7A, the maximum stresses are 24.02 N/m at 17.5% uniaxial strain along the *y* direction and 23.74 N/m at 22% uniaxial strain along the *x* direction. A fracture occurs when the uniaxial strain reaches 18.5% along the *y* direction. After that, the effect of uniaxial tensile strain on lattice stability is further studied by calculating the phonon spectra with increasing uniaxial strain increases. As shown in Figure 7B, there are no imaginary phonon modes of the phonon spectra at 19% uniaxial strain along the *x* direction, while it appears in the figure when the uniaxial strain along the *x* direction but have imaginary phonon modes when the uniaxial strain along the *y* direction is 19%. This indicates that the lattice of *Pmma* BN becomes unstable when the uniaxial strain along the *x* direction is 20% and 19%, respectively, while the higher fracture strain is beneficial to the future application of this material in strain engineering.

Using the HSE06 functional, we calculate the band structure of *Pmma* BN and study the effect of uniaxial strain on its electronic properties. The uniaxial strain along the *x* direction of the band structure is up to 19%, while the uniaxial strain along the *y* direction is up to 18% and the step length is 1%. The band structures with uniaxial strains of 0%, 5%, 10%, 15%, and 19% along the *x* direction are shown in Figure 8A. The band gap under uniaxial strain along the *x* direction is shown in Figure 8B. The band gaps of *Pmma* BN

	Uniaxial strain	а	b	C ₁₁	C ₂₂	C ₁₂	C ₄₄	E _x	E _y	V _x	V _y
Pmma BN	0%	9.13	6.77	195	256	58	34	182	239	0.23	0.29
	5% (x-axis)	9.58	6.77	183	231	42	67				
	10% (<i>x</i> -axis)	10.04	6.77	158	205	28	5				
	15% (x-axis)	10.50	6.77	115	184	15	59				
	5% (y-axis)	9.13	7.11	176	215	39	12				
	10% (y-axis)	9.13	7.45	154	159	23	20				
	15% (y-axis)	9.13	7.79	133	92	10	24				
B ₄ N ^[39]	0%			153	267	-4.9		153	268	-0.02	-0.03
h-BN ^[40]	0%			290	290	64		276	276	0.22	0.22

Table 1. Lattice constants (Å), in-plane elastic constants (N/m), Young's moduli (N/m) and Poisson's ratios along the x and y directions for *Pmma* BN under uniaxial strain

Table 2. In-plane elastic constants C_{2D} (N/m), effective mass m_d^* (in m_e), deformation potential E_1 (eV) and carrier mobility μ_d (cm² V⁻¹ s⁻¹) of holes and electrons in *Pmma* BN

Carrier types	C _x	C _y	m_x^*/m_o	m_y^*/m_o	E _{1x}	E _{1y}	μ_{x}	μ_y
Electrons	195	256	0.39	17.46	6.32	0.98	143	172
Holes	195	256	0.61	26.63	4.76	0.56	106	228



Figure 1. (A) Top and side views of optimized Pmma BN atomic structure. (B) ELF for Pmma BN with an isosurface value of 1.

remain indirect under uniaxial strain and the uniaxial strain applied along the *x* direction makes the band gap gradually smaller. When the uniaxial strain along the *x* direction reaches 19%, the band gap is the smallest at 4.51 eV. The band gap adjustable range of *Pmma* BN at the uniaxial strain along the *x* direction reaches 0.64 eV. Figure 8C exhibits the band structures with uniaxial strains of 0%, 1%, 2%, 10% and 18% along the *y* direction. When 1% uniaxial strain is applied along the *y* direction, the CBM is transferred from the G and B points to the G and A points (0.5, 0.5, 0) and the band gap decreases by 0.02 eV. When the uniaxial strain is applied to 2% along the *y* direction, the CBM occurs between the A and F points, while the VBM is transferred from the F and G points to the G point and the band gap decreases to 5.08 eV. The band gap under uniaxial strain along the *y* direction is shown in Figure 8D. The band gap decreases to 4.58 eV (18%) as the uniaxial strain increases along the *y* direction.

To explore its possible application fields, the optical absorption spectra of *Pmma* BN under uniaxial strain with a step size of 2% are studied by Cambridge Serial Total Energy Package (CASTEP)^[48] and PBE^[33] functional. Since the spectral lines do not change significantly with a step size of 2%, in order to make the optical absorption spectra look clearer, we plotted Figure 9 with a step size of 4%. Obviously, *Pmma* BN and its structures under uniaxial strain have a strong optical absorption capacity in the ultraviolet region.



Figure 2. (A) Phonon spectra of *Pmma* BN. (B) Fluctuations in total energy as a function of AIMD simulations at 300 and 500 K of *Pmma* BN. (C) Atomic configurations at the end of the 4 ps AIMD simulations.



Figure 3. Three-dimensional plot of strain-energy surface.



Figure 4. Orientation dependences of (A) Young's modulus and (B) Poisson's ratio for Pmma BN.



Figure 5. (A) Electronic band structure and PDOS of *Pmma* BN. (B) Band-decomposed charge densities at the VBM and CBM. The isosurface value is $0.025 \text{ eV}/\text{Å}^3$. (C) Orientation-dependent carrier effective mass. The angle refers to the *x* direction.

Figure 9A and B show the optical absorption spectra of *Pmma* BN under uniaxial strain along the x and y direction, respectively. For the x direction, the optical absorption spectra exhibit a slight red shift and a promising increase in absorption in the visible region, which is mainly caused by the contraction of the band gap under uniaxial strain. For the y direction, the increase in uniaxial strain significantly improves the optical absorption capacity of *Pmma* BN after 100 nm in the optical absorption spectra. The optical absorption spectra indicate that applying uniaxial strain along the y direction has a greater effect on the optical absorption capacity of *Pmma* BN in the ultraviolet region. The adjustable wide band gap and strong optical absorption in the ultraviolet region may allow *Pmma* BN to meet the needs of photodetectors.

Finally, considering the increasing importance of green energy production, the thermoelectric properties, such as the electrical conductivity, Seebeck coefficient, phononic thermal conductivity, and thermoelectric figure of merit (ZT), of *Pmma* BN under uniaxial strain in the temperature range of 50–1950 K are studied. The results as a function of temperature are displayed in Figure 10. A large Seebeck coefficient and electrical conductivity and a small thermal conductivity favor high-performance thermoelectric materials. As seen from Figure 10A, *Pmma* BN has the largest electrical conductivity at 18% uniaxial strain along the *y* direction and the smallest electrical conductivity at 19% uniaxial strain along the *x* direction. From Figure 10B, the Seebeck coefficients of *Pmma* BN under uniaxial strain are similar at 50–250 K and *Pmma* BN without uniaxial strain has the largest Seebeck coefficient after 250 K. In Figure 10C and D, the phononic thermal conductivity and ZT show a monotonic increase with temperature for *Pmma* BN under uniaxial strain. Additionally, the ZT of *Pmma* BN without uniaxial strain after 250 K increases more rapidly than others and reaches 0.54 at 1950 K. The results demonstrate that the application of uniaxial strain reduces the thermoelectric figure of merit of *Pmma* BN and affects the thermoelectric properties. Therefore,



Figure 6. (A) E_{CBM} and (B) E_{VBM} of *Pmma* BN relative to the vacuum energy E_{vacuum} as a function of small uniaxial strain along the x direction. (C) E_{CBM} and (D) E_{VBM} of *Pmma* BN relative to the vacuum energy E_{vacuum} as a function of small uniaxial strain along the y direction.

when making a heterojunction, the lattice match should be as small as possible.

CONCLUSIONS

In this study, a new type of 2D boron nitride named *Pmma* BN is proposed and studied using firstprinciples calculations. First, the dynamic, thermal, and mechanical stability of the novel 2D BN in the Pmma phase are demonstrated by phonon spectra, AIMD simulations, and elastic constants. Second, the orientation dependences of the Young's modulus and Poisson's ratio of Pmma BN show that it has mechanical anisotropy. Additionally, Pmma BN is an indirect band gap semiconductor material with a band gap of 5.15 eV and the hole and electron effective masses show high anisotropy. The electron carrier mobility of *Pmma* BN along the x and y directions is basically the same, while the hole carrier mobility along the y direction is more than double that along the x direction. The influence of uniaxial strain on the physical properties of BN is then studied. The band gaps of Pmma BN remain indirect under uniaxial strain and its adjustable range at the uniaxial strain along the *x* direction reaches 0.64 eV. When 2% uniaxial strain is applied along the y direction, the CBM is between A and F points, the VBM is transferred from the F and G points to the G point and the band gap is reduced to 5.08 eV. Pmma BN under uniaxial strain shows strong optical absorption capacity in the ultraviolet region. Finally, considering the increasing importance of green energy production, the thermoelectric properties of *Pmma* BN show that the application of uniaxial strain reduces its ZT and affects its thermoelectric properties. Therefore, these results provide a new 2D boron nitride material with excellent material properties for possible applications.



Figure 7. (A) Strain-stress curves for uniaxial tensile strain in *Pmma* BN. (B) Phonon spectra at 19% and 20% uniaxial strains along the *x* direction. (C) Phonon spectra at 18% and 19% uniaxial strains along the *y* direction.



Figure 8. (A) Electronic band structures of *Pmma* BN under 0%, 5%, 10%, 15% and 19% uniaxial strain along the *x* direction. (B) Band gap under uniaxial strain along the *x* direction. (C) Electronic band structures under 0%, 1%, 2%, 10% and 18% uniaxial strain along the *y* direction. (D) Band gap under uniaxial strain along the *y* direction.



Figure 9. Absorption spectra of *Pmma* BN under uniaxial strain along the (A) x direction and the (B) y direction.



Figure 10. Pmma BN under uniaxial strain: (A) electrical conductivity; (B) Seebeck coefficient; (C) phononic thermal conductivity; (D) thermoelectric figure of merit.

DECLARATIONS

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Authors' contributions

Funding acquisition: Fan Q, Yun S Investigation: Fan Q, Zhou H, Zhao Y, Yun S Writing – original draft: Fan Q, Zhou H Project administration: Fan Q

Availability of data and materials

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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Conflicts of interest

All authors declared that there are no conflicts of interest.

Ethical approval and consent to participate

Not applicable.

Consent for publication

Not applicable.

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