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Influence of hydrogel and porous scaffold on the magnetic thermal property and anticancer effect of Fe₃O₄ nanoparticles

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How to cite this article: Wang M, Sun R, Chen H, Liu X, Yoshitomi T, Takeguchi M, Kawazoe N, Yang Y, Chen G. Influence of hydrogel and porous scaffold on the magnetic thermal property and anticancer effect of Fe₃O₄ nanoparticles. *Microstructures* 2023;3:2023042. <https://dx.doi.org/10.20517/microstructures.2023.46>

Received: 5 Sep 2023 **First Decision:** 20 Sep 2023 **Revised:** 27 Sep 2023 **Accepted:** 19 Oct 2023 **Published:** 10 Nov 2023

Academic Editors: Yin Xiao, Chun-Xia Zhao **Copy Editor:** Fangyuan Liu **Production Editor:** Fangyuan Liu

Abstract

Magnetic hyperthermia uses magnetic nanoparticles (MNPs) for conversion of magnetic energy into thermal energy under an alternating magnetic field (AMF) to increase local temperature for ablation of cancer cells. The magnetic thermal capacity of MNPs not only depends on the intrinsic properties of MNPs but is also affected by the microenvironmental matrices surrounding the MNPs. In this study, the influence of agarose hydrogels and gelatin porous scaffolds on the magnetic thermal property and anticancer effect of Fe₃O₄ nanoparticles (NPs) were investigated with a comparison to free Fe₃O₄ NPs. Flower-like Fe₃O₄ NPs were synthesized and embedded in agarose hydrogels and gelatin porous scaffolds. Under AMF irradiation, the free Fe₃O₄ NPs had the best magnetic thermal properties and the most efficiently increased the local temperature to ablate breast cancer cells. However, the Fe₃O₄ NPs embedded in agarose hydrogels and gelatin porous scaffolds showed reduced magnetic-thermal conversion capacity, and the local temperature change was decreased in comparison to free Fe₃O₄ NPs during AMF irradiation. The gelatin porous scaffolds showed a higher inhibitory influence than the agarose hydrogels. The inhibitory effect of agarose hydrogels and gelatin porous scaffolds on magnetic-thermal conversion capacity resulted in a decreased anticancer ablation capacity to breast cancer cells during AMF irradiation. The Fe₃O₄ NP-embedded gelatin scaffolds showed the lowest anticancer effect. The results suggested that the matrices used to



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deliver MNPs could affect their performance, and appropriate matrices should be designed to maximize their therapeutic effect for biomedical applications.

Keywords: Fe₃O₄ NPs, hydrogel, scaffold, agarose, gelatin, magnetic hyperthermia, anticancer

INTRODUCTION

Magnetic nanoparticles (MNPs) can convert magnetic energy to thermal energy when subjected to alternating magnetic field (AMF)^[1,2]. The heat generated by MNPs under AMF irradiation increases the local temperature to ablate cancer cells, known as magnetic hyperthermia (MH)^[3-5]. MH has been developed as an effective approach for cancer treatment due to its good biocompatibility and deep strong tissue penetration. This approach has also been combined with radiotherapy, chemotherapy, and immunotherapy to further improve the anticancer effect^[6-12]. It is pivotal to increase magnetic-thermal conversion efficiency to achieve a maximized therapeutic effect with the minimized dosage of MNPs.

To achieve high magnetic-thermal conversion efficiency, many studies have reported the optimization of synthesis methods of MNPs^[13,14] by controlling their structure and magnetic characteristics, including shape, size, size distribution, dispersion and aggregation state, crystallinity, composition, and magnetic parameters^[15-26]. Except for the intrinsic properties of MNPs, their surrounding microenvironmental matrices can affect the magnetic-thermal conversion^[27,28]. Incorporation of MNPs in hydrogels has been reported to change their magnetic-thermal conversion property^[28-31]. Engelmann *et al.* immobilized them in acrylamide hydrogels and found that the heating efficiency of MNPs decreased when the hydrogel stiffness increased^[28]. Suto *et al.* compared the influence of polyvinyl alcohol hydrogel and water^[31]. The heating efficiency of MNPs dispersed in water was better than that dispersed in polyvinyl alcohol hydrogel, and their specific absorption rate value in hydrogel showed 67% less than that in water^[31]. These studies suggest the inhibitory effects of hydrogels on the magnetic-thermal conversion property of MNPs.

In recent years, localized delivery of photothermal nanoparticles (NPs) has been demonstrated as an efficient strategy to accumulate and constrain them in tumors to maximize the photothermal ablation effect while decreasing their side effect^[32-37]. Both hydrogels and porous scaffolds are good carriers for the local delivery of photothermal NPs. However, the influence of porous scaffolds on the magnetic-thermal conversion of MNPs remains elusive. It is desirable to compare the influence of aqueous solution, hydrogels^[38-45], and porous scaffolds^[46-48] on the magnetic-thermal conversion of MNPs to maximize their MH effect.

Based on the above considerations, in this study, different microenvironments of phosphate buffer saline (PBS), hydrogels, and porous scaffolds were used to investigate the magnetic-thermal conversion property and anticancer effect of MNPs under AMF irradiation [Figure 1]. The same concentration of Fe₃O₄ NPs was used in PBS, agarose hydrogels, and gelatin porous scaffolds to disclose the influence of the different matrices on magnetic thermal effects. Moreover, Fe₃O₄ NPs at different concentrations were incorporated in the same matrix to study the MNP concentration dependence. The free Fe₃O₄ NPs in PBS exhibited the best magnetic thermal property, while embedding in agarose hydrogels or gelatin porous scaffolds decreased the temperature change during AMF irradiation. The anticancer effect was investigated *in vitro* by incubating breast cancer cells (MDA-MB-231-Luc cells) with free Fe₃O₄ NPs, agarose/Fe₃O₄ hydrogels, and gelatin/Fe₃O₄ porous scaffolds under AMF irradiation. The free Fe₃O₄ NPs showed the highest anticancer effect under AMF irradiation, followed by Fe₃O₄ NPs in agarose hydrogels and then Fe₃O₄ NPs in gelatin porous scaffolds. The matrices for MNP delivery could affect the magnetic-thermal conversion property and anticancer effect of Fe₃O₄ NPs.

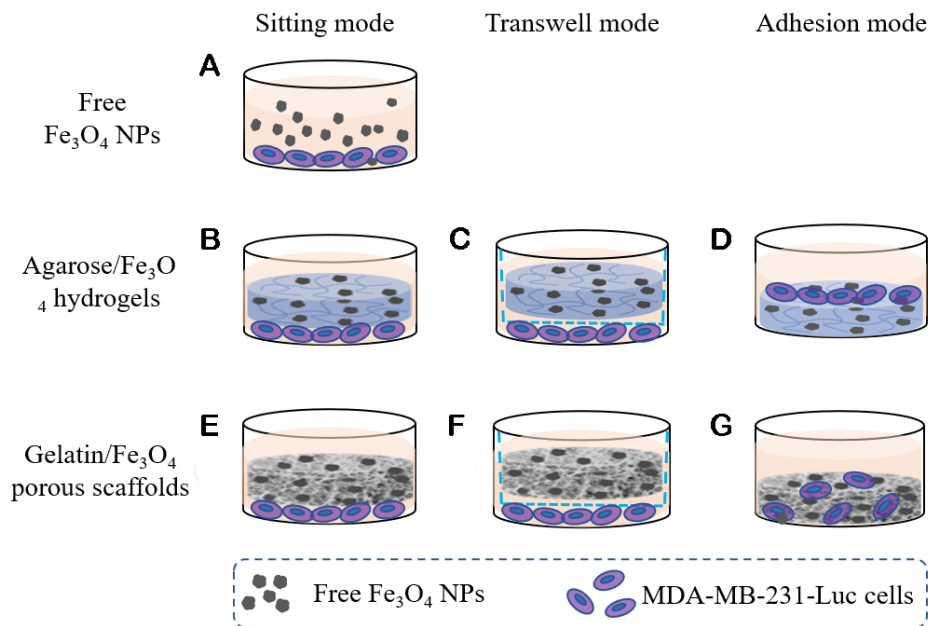


Figure 1. Anticancer experimental scheme of free Fe₃O₄ NPs (A), agarose/Fe₃O₄ hydrogels (B-D), and gelatin/Fe₃O₄ porous scaffolds (E-G). Three modes (sitting, transwell, and adhesion modes) were used to simulate the cells near or far away from or directly adhered to the matrices.

MATERIALS AND METHODS

Materials

Iron (II) chloride tetrahydrate (FeCl₂·4H₂O, ≥ 99%), iron (III) chloride hexahydrate (FeCl₃·6H₂O, ≥ 97%), and sodium citrate tribasic dihydrate (≥ 99%) were purchased from Sigma-Aldrich (St. Louis, MO, USA). Diethylene glycol (DEG, 99%), sodium hydroxide (NaOH, 99.99%), acetic acid, ethanol, ethyl acetate, 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDC), and N-hydroxysuccinimide (NHS) were obtained from Wako Pure Industries, Ltd. (Tokyo, Japan). N-methyldiethanolamine (NMDEA, 99%) was purchased from Tokyo Kasei. Kogyo Co., Ltd, and gelatin (porcine) was obtained from Nitta Gelatin (Osaka, Japan).

Synthesis and characterization of Fe₃O₄ NPs

Fe₃O₄ NPs were synthesized using Fe (II) and Fe (III) in a mixture solution of DEG and NMDEA (1:1, v/v)^[49,50]. First, 4 mmol FeCl₂·4H₂O and 2 mmol FeCl₃·z'6H₂O were added and dissolved in an 80 mL polyols mixture under a nitrogen atmosphere at room temperature. Next, 40 mL of 400 mM NaOH in a polyol mixture was dropped into the iron salts solution and stirred for 3 h. Then, the mixed solution was heated to 220 °C and reacted for 12 h under the protection of nitrogen. After the reaction, the black precipitates were collected and washed with ethanol/ethyl acetate solvent (1:1, v/v) three times to obtain bare Fe₃O₄ NPs. The obtained bare Fe₃O₄ NPs were redispersed in sodium citrate aqueous solution and reacted at 60 °C for 24 h to modify the Fe₃O₄ NPs with citrate anions. After centrifugation, washing, and lyophilization, the citrate-modified Fe₃O₄ NPs were obtained. The prepared Fe₃O₄ NPs were redispersed in PBS to obtain the Fe₃O₄ NPs colloidal solution at a concentration of 5 mg mL⁻¹, 10 mg mL⁻¹, and 20 mg mL⁻¹, which were named free Fe₃O₄-5 (5 mg mL⁻¹), free Fe₃O₄-10 (10 mg mL⁻¹), and free Fe₃O₄-20 (20 mg mL⁻¹), respectively. The structure and mean diameter of the citrate-modified Fe₃O₄ NPs were analyzed by a transmission electron microscope (TEM, JEOL 2100F, Japan). The hydrodynamic size of the citrate-modified Fe₃O₄ NPs was analyzed by dynamic light scattering (DLS, Beckman Coulter, USA).

Preparation and characterization of agarose/Fe₃O₄ hydrogels and gelatin/Fe₃O₄ scaffolds

The agarose/Fe₃O₄ hydrogels were prepared by dispersing the citrate-modified Fe₃O₄ NPs in agarose hydrogels. At first, 0.1 g agarose was dissolved in 5 mL PBS at 110 °C. After the temperature of the agarose solution was cooled to around 40 °C, the citrate-modified Fe₃O₄ NPs suspension solution was added to prepare 1% agarose solution with Fe₃O₄ NPs at 5 mg mL⁻¹, 10 mg mL⁻¹, and 20 mg mL⁻¹. Agarose hydrogels without Fe₃O₄ NPs were prepared as a control. After vortexing, the mixture was added into cylinder molds (Φ10 mm × H4 mm) and immediately transformed into a 4 °C refrigerator for 30 min to form agarose/Fe₃O₄ hydrogels. The obtained hydrogels with different NP concentrations were referred to as agarose/Fe₃O₄-5 (5 mg mL⁻¹), agarose/Fe₃O₄-10 (10 mg mL⁻¹), and agarose/Fe₃O₄-20 (20 mg mL⁻¹).

The porous scaffolds of gelatin and Fe₃O₄ NPs were prepared by using ice particulate porogen methods^[51-61]. Firstly, ice particulates with a diameter of 250-355 μm were obtained by spraying Milli-Q water into liquid nitrogen and sieved in a low-temperature chamber. Gelatin solution (8%, wt/v) in 70% acetic acid was mixed with the citrate-modified Fe₃O₄ NP suspension solution (1:1, v/v) to obtain gelatin/Fe₃O₄ NP mixture solution. Then, the temperature-balanced ice particulates (-4 °C) were added to the gelatin/Fe₃O₄ NP mixture solution (7:3, wt/v) in a -4 °C chamber, and the final concentrations of Fe₃O₄ NPs in the mixture solution were 5, 10, and 20 mg cm⁻³, respectively. The mixture was transformed into a silicone frame and frozen at -20 °C for 12 h and -80 °C for 4 h. Then, the lyophilized constructs were cross-linked by EDC (50.0 mM) and NHS (20.0 mM) in a series of ethanol/water mixture solvents [95/5, 90/10, and 85/5 (v/v)] each for 8 h. Finally, the cross-linked scaffolds were immersed in 0.1 M glycine solution to block the activated residual carboxyl groups. The lyophilized gelatin/Fe₃O₄ NPs composite scaffolds prepared with different concentrations of citrate-modified Fe₃O₄ NPs were referred to as gelatin/Fe₃O₄-5 (5 mg cm⁻³), gelatin/Fe₃O₄-10 (10 mg cm⁻³), and gelatin/Fe₃O₄-20 (20 mg cm⁻³). A control scaffold without Fe₃O₄ NPs was also prepared and referred to as a gelatin porous scaffold. An optical microscope (Olympus, Japan) and a scanning electron microscope (SEM, Hitachi S-4800, Tokyo, Japan) were used to observe the gross appearances and the inner pore structures of the lyophilized agarose/Fe₃O₄ hydrogels and gelatin/Fe₃O₄ scaffolds.

Magnetic thermal property of Fe₃O₄ NPs in different matrices

The magnetic thermal properties of free Fe₃O₄ NPs, agarose/Fe₃O₄ hydrogels, and gelatin/Fe₃O₄ porous scaffolds were investigated under AMF irradiation. First, 300 μL of free Fe₃O₄-5, Fe₃O₄-10, and Fe₃O₄-20 solutions were added in 0.5 mL Eppendorf tubes, respectively. The samples were placed in the center of Double H CoilSets AMF (Frequency: 373.6 kHz; Field intensity: 130 Gauss) for 10 min using a D5 series machine (nB nanoScale BioMagnetics, Zaragoza, Spain). Subsequently, the IR1 thermal imaging system (nB nanoScale Biomagnetics, Zaragoza, Spain) was used to record the temperature change of different samples. Triplicate samples were used for each measurement.

Similarly, 300 μL of aqueous agarose solution without or with 5 mg mL⁻¹, 10 mg mL⁻¹, and 20 mg mL⁻¹ of Fe₃O₄ NPs were added in 0.5 mL Eppendorf tubes, respectively. All samples were immediately transformed into a 4 °C refrigerator for 30 min to form agarose hydrogel, agarose/Fe₃O₄-5, agarose/Fe₃O₄-10, and agarose/Fe₃O₄-20 hydrogel samples. Before exposure to AMF, the agarose/Fe₃O₄ hydrogel samples were balanced at room temperature for 2 h. The temperature change of the agarose/Fe₃O₄ hydrogel samples was measured and recorded during 10 min AMF irradiation at 373.6 kHz of frequency and a field intensity of 130 Gauss. Triplicate samples were used for each measurement.

The gelatin/Fe₃O₄ porous scaffolds were molded into cylinder discs (Φ10 mm × H4 mm) and hydrated with pure water (300 μL/disc) in silicone frames. The gelatin/Fe₃O₄ porous scaffold discs were exposed to AMF for 10 min (Frequency: 373.6 kHz; Field intensity: 130 Gauss), and the temperature change was measured.

Triplicate samples were used for each measurement.

Anticancer effect of Fe₃O₄ NPs in different matrices

The anticancer effect of free Fe₃O₄ NPs, agarose/Fe₃O₄ hydrogels, and gelatin/Fe₃O₄ porous scaffolds was investigated by incubating breast cancer cells in the different matrices under AMF irradiation^[55-61]. Three culture modes were used to simulate the cells directly adhered to, near, or far away from the matrices [Figure 1]. The cells were seeded in wells of cell culture plates, and then free Fe₃O₄ NPs, agarose/Fe₃O₄ hydrogels, or gelatin/Fe₃O₄ porous scaffolds were added to the adhered cells. The hydrogels and porous scaffolds were sitting on the adhered cells (sitting mode), which simulated the cells near the matrices. The second mode was a transwell mode by seeding cells in the bottom wells of the transwell plates and placing the hydrogels and porous scaffolds in the inserts, which simulated the cells far away from the matrices. The third mode was an adhesion mode by seeding cells on the hydrogels or in the porous scaffolds to allow the cells to adhere to the hydrogels or in the pores of the porous scaffolds, which simulated the cells directly adhering to the matrices.

Anticancer effect of free Fe₃O₄ NPs

The free Fe₃O₄ NPs could only be added in the culture medium. Therefore, only the sitting mode was used for investigating the anticancer effect of free Fe₃O₄ NPs. The sub-cultured MDA-MB-231-Luc cells were harvested and resuspended in a culture medium at a concentration of 2.5×10^5 cells mL⁻¹. A 200 μL cell suspension solution was seeded in the wells of a 48-well plate (5×10^4 cells well⁻¹). After culture in a humidified incubator (5% CO₂, 37 °C) for 24 h, the culture medium was removed, and another 200 μL fresh culture medium was added. Then, 300 μL medium, without or with free Fe₃O₄-5, free Fe₃O₄-10 and free Fe₃O₄-20, was added to the wells, respectively. After co-incubation for 2 h, the wells were exposed to AMF (frequency: 373.6 kHz of; field intensity: 130 Gauss) for 10 min. Cell viability before and after AMF irradiation was visualized by live/dead staining and quantitatively analyzed by WST-1 assay. Triplicate samples were used for each measurement.

Anticancer effect of agarose/Fe₃O₄ hydrogels

The three culture modes were used for the investigation of the anticancer effect of the agarose/Fe₃O₄ hydrogel. For the sitting mode, the MDA-MB-231-Luc cells were seeded and cultured in the wells of a 48-well plate, as mentioned above. After the culture medium was changed with another 200 μL fresh medium, the agarose and agarose/Fe₃O₄ hydrogel discs (Φ10 mm × H4 mm) were placed on the cells. After co-incubation for 2 h, the wells were exposed to AMF (frequency: 373.6 kHz of; field intensity: 130 Gauss) for 10 min. Cell viability was investigated by live/dead staining and WST-1 assay before and after AMF irradiation. Triplicate samples were used for each measurement.

For the transwell mode, the MDA-MB-231-Luc cells were seeded in the centers of the wells of 24-well plate by using donut-shaped silicone rings (inner diameter 10 mm, outer diameter: 16 mm, height: 2 mm). The seeded cell number was the same (5×10^4 cells well⁻¹). The agarose/Fe₃O₄ hydrogel discs were placed on the transwell inserts and co-cultured with the cells on the bottom wells. The transwell plates containing cells and discs were irradiated by AMF (frequency: 373.6 kHz; magnetic field: 130 Gauss) for 10 min. Before and after AMF irradiation, cell viability was investigated, as mentioned above. Triplicate samples were used for each measurement.

For the adhesion mode, the agarose/Fe₃O₄ hydrogel discs (Φ10 mm × H4 mm) were put in the wells of a 48-well plate. Subsequently, 200 μL of cell suspension solution was seeded on the agarose/Fe₃O₄ hydrogel discs. After 2 h incubation, the plates were exposed to AMF (frequency: 373.6 kHz; magnetic field: 130 Gauss) for

10 min, and cell viability was investigated, as mentioned above. Triplicate samples were used for each measurement.

Anticancer effect of gelatin/Fe₃O₄ porous scaffolds

The three culture modes were also used for the investigation of the anticancer effect of the gelatin and gelatin/Fe₃O₄ porous scaffolds. All the experiment procedures were the same as those of agarose/Fe₃O₄ hydrogel discs by using the porous scaffold discs (Φ10 mm × H4 mm). Triplicate samples were used for each measurement.

RESULTS

Characterization of Fe₃O₄ NPs

The morphology and size of the citrate-modified Fe₃O₄ NPs were characterized by TEM. As shown in [Figure 2A-C](#), the NPs displayed a flower-like shape, which should have a good magnetic-thermal conversion capacity for MH. They had an average size of 30.8 ± 5.7 nm from the TEM images. The hydrodynamic size of the citrate-modified Fe₃O₄ NPs was measured in aqueous solution by DLS, and the hydrodynamic size was 108.5 ± 28.5 nm [[Figure 2D](#)].

Characterization of agarose/Fe₃O₄ hydrogels and gelatin/Fe₃O₄ porous scaffolds

The agarose/Fe₃O₄ hydrogel discs are shown in [Figure 3A](#). As the concentration of Fe₃O₄ NPs increased, the appearance of agarose/Fe₃O₄ hydrogel changed from transparent to black. SEM observation of the lyophilized agarose/Fe₃O₄ hydrogel discs showed that the agarose hydrogels with different amounts of Fe₃O₄ NPs had similar structures [[Figure 3B](#)]. They had spindle-shaped pores. The gelatin porous scaffold without Fe₃O₄ NPs was white, while the gelatin/Fe₃O₄ porous scaffolds became gray (gelatin/Fe₃O₄-5), dark gray (gelatin/Fe₃O₄-10), and black (gelatin/Fe₃O₄-20) [[Figure 3C](#)]. The gelatin and gelatin/Fe₃O₄ porous scaffolds had the same pore structures. They had large spherical pores that were surrounded by some small pores [[Figure 3D](#)]. The large spherical pores were controlled by the ice particulates that were used as a porogen material. The results indicated that the embedding of Fe₃O₄ NPs did not affect the pore structures of hydrogels and porous scaffolds.

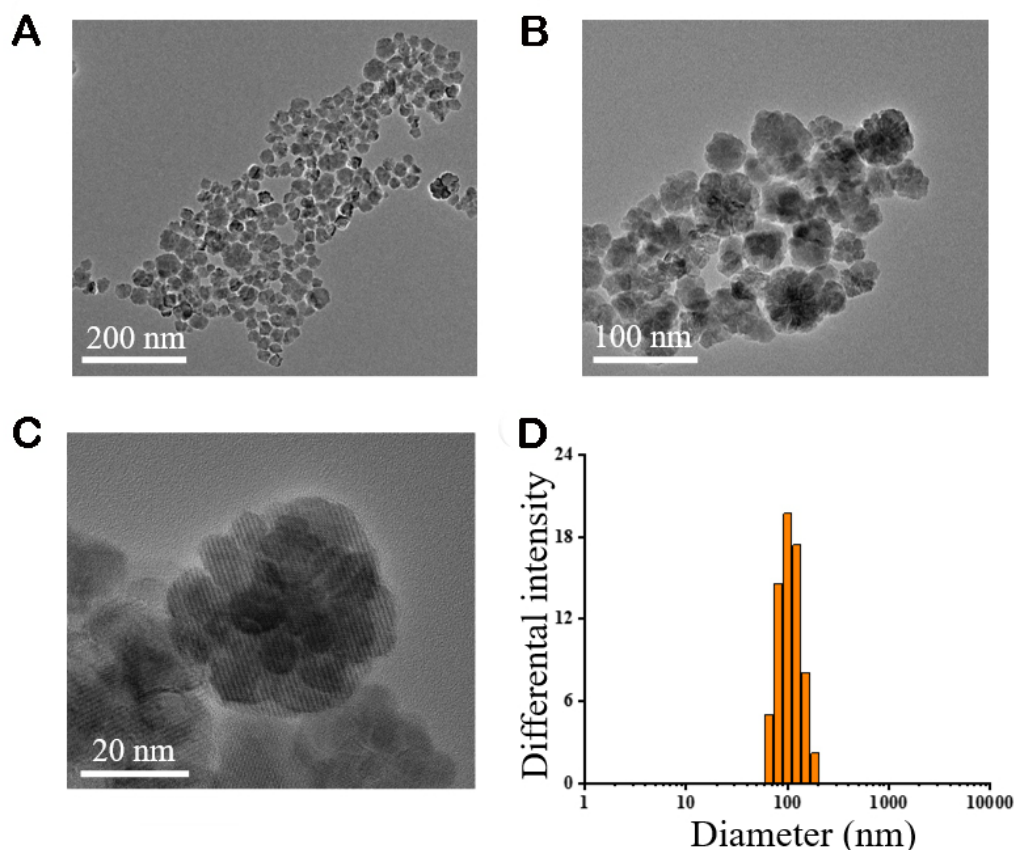
Magnetic thermal property of Fe₃O₄ NPs in different matrices

The magnetic thermal properties of Fe₃O₄ NPs in PBS, agarose hydrogels, and gelatin porous scaffolds were investigated by applying AMF (frequency: 373.6 kHz; field: 130 Gauss) for 10 min, and the results are shown in [Figure 4](#) and [Table 1](#). The temperature of PBS, agarose hydrogels, and gelatin porous scaffolds without Fe₃O₄ NPs had no obvious change after AMF irradiation [[Figure 4A](#) and [Table 1](#)]. The results suggested that PBS, agarose hydrogels, and gelatin porous scaffolds had no magnetic-thermal conversion capacity in the absence of Fe₃O₄ NPs. When Fe₃O₄ NPs were added to PBS, hydrogels, and porous scaffolds, the temperature change significantly increased under AMF irradiation.

The temperature change increased with the irradiation time and became plateau after 10 min AMF irradiation [[Figure 4B-D](#)]. The temperature change of free Fe₃O₄-5, agarose/Fe₃O₄-5, and gelatin/Fe₃O₄-5 was 24.1 ± 1.7 °C, 14.0 ± 0.3 °C, and 5.2 ± 0.3 °C, respectively, after 10 min AMF irradiation [[Figure 4B](#) and [Table 1](#)]. The temperature change of free Fe₃O₄-10, agarose/Fe₃O₄-10, and gelatin/Fe₃O₄-10 increased to 38.3 ± 1.1 °C, 22.8 ± 1.7 °C, and 9.1 ± 0.5 °C, respectively, after 10 min AMF irradiation [[Figure 4C](#) and [Table 1](#)]. The temperature change of free Fe₃O₄-20, agarose/Fe₃O₄-20, and gelatin/Fe₃O₄-20 increased to 65.7 ± 1.4 °C, 33.8 ± 1.0 °C, and 13.2 ± 0.4 °C, respectively, after 10 min AMF irradiation [[Figure 4D](#) and [Table 1](#)]. The results indicated that the temperature change of free Fe₃O₄ NPs, agarose/Fe₃O₄, and gelatin/Fe₃O₄ was positively correlated with the concentration of Fe₃O₄ NPs. Increasing the concentration of Fe₃O₄ NPs resulted in a bigger temperature change.

Table 1. Magnetic thermal property of Fe₃O₄ NPs in different matrices under AMF irradiation [Mean ± SD (n = 3)]

Sample	ΔT of PBS (°C)	ΔT of agarose hydrogel (°C)	Percentage compared to free NPs	ΔT of gelatin porous scaffold (°C)	Percentage compared to free NPs
No Fe ₃ O ₄ NPs	0.7 ± 0.2	1.0 ± 0.2	/	0.8 ± 0.2	/
Fe ₃ O ₄ -5 mg mL ⁻¹	24.1 ± 1.7	14.0 ± 0.3	58.1%	5.2 ± 0.3	21.6%
Fe ₃ O ₄ -10 mg mL ⁻¹	38.3 ± 1.1	22.8 ± 1.7	59.5%	9.1 ± 0.5	23.8%
Fe ₃ O ₄ -20 mg mL ⁻¹	65.7 ± 1.4	33.8 ± 1.0	51.5%	13.2 ± 0.4	20.1%

**Figure 2.** TEM images of citrate-modified Fe₃O₄ NPs at low (A), middle (B), and high magnifications (C). Hydrodynamic size distribution of citrate-modified Fe₃O₄ NPs (D).

The free Fe₃O₄ NPs in PBS showed the highest temperature change. The temperature change was reduced to 51.5%-59.5% when the Fe₃O₄ NPs were embedded in agarose hydrogels. The temperature change was further decreased to 20.1%-23.8% when the Fe₃O₄ NPs were embedded in gelatin porous scaffolds. The results indicated that the matrix where Fe₃O₄ NPs were embedded could significantly affect the magnetic-thermal conversion property of Fe₃O₄ NPs.

Anticancer effect of free Fe₃O₄ NPs

MH uses the Fe₃O₄ NPs to absorb and convert magnetic energy to heat and raise the local temperature, thereby killing the cancer cells. In this study, MDA-MB-231-Luc cells were cultured in a culture medium supplemented with free Fe₃O₄ NPs under different concentrations. Cell viability before and after AMF irradiation was investigated by live/dead staining and WST-1 assay [Figure 5]. Before AMF irradiation,

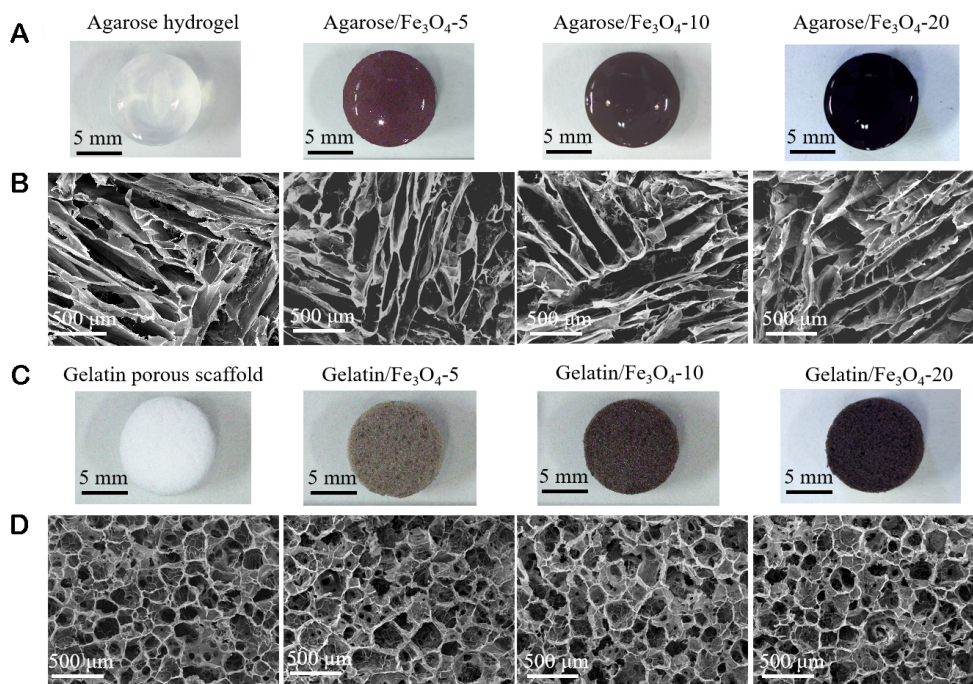


Figure 3. Gross appearances (A) and SEM images (B) of agarose and agarose/ Fe_3O_4 hydrogels with different concentrations of citrate-modified Fe_3O_4 NPs. Gross appearances (C) and SEM images (D) of gelatin and gelatin/ Fe_3O_4 porous scaffold with different concentrations of citrate-modified Fe_3O_4 NPs.

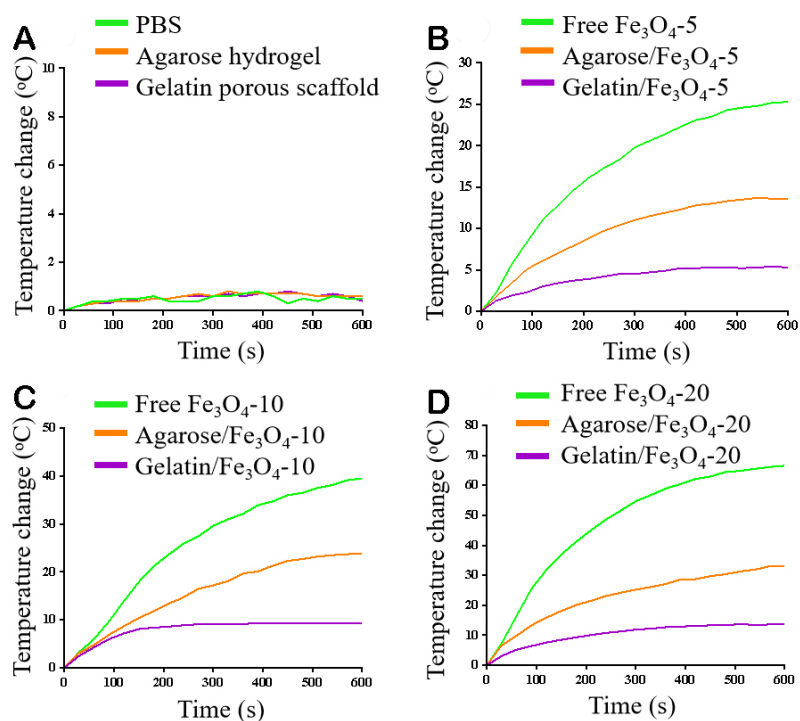


Figure 4. Heating curves of PBS, agarose hydrogels, and gelatin porous scaffolds without Fe_3O_4 NPs (A) and containing 5 mg mL^{-1} Fe_3O_4 NPs (B), 10 mg mL^{-1} Fe_3O_4 NPs (C), and 20 mg mL^{-1} Fe_3O_4 NPs (D) during AMF irradiation.

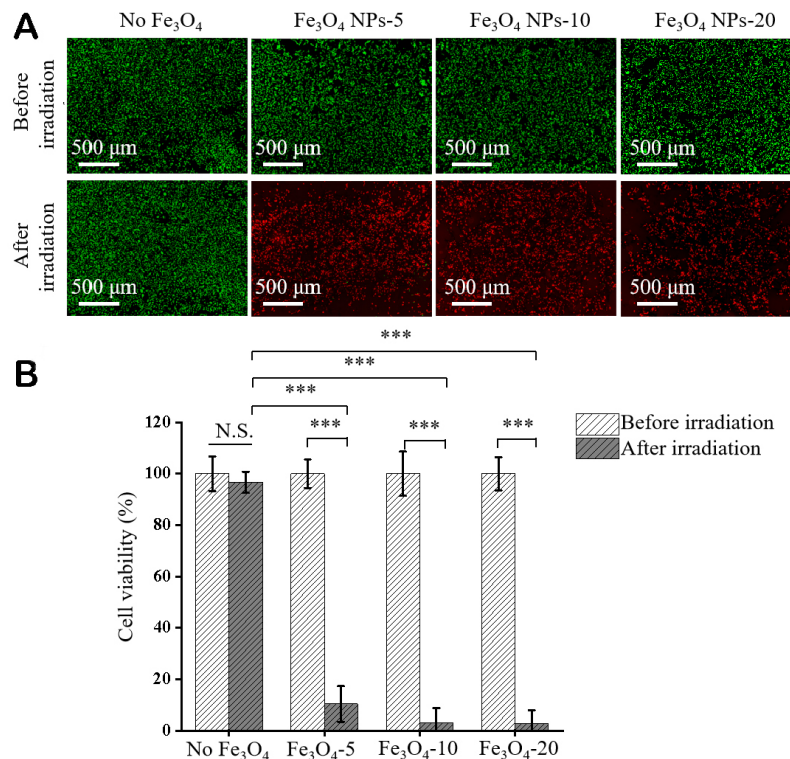


Figure 5. Anticancer effect of free Fe₃O₄ NPs. Live/dead staining of MDA-MB-231-Luc cells cultured without or with free Fe₃O₄ NPs before and after AMF irradiation (live cell: green fluorescence, dead cells: red fluorescence) (A). Quantified cell viability during culture without or with free Fe₃O₄ NPs before and after AMF irradiation (B). Cell viability was normalized to that cultured with PBS without free Fe₃O₄ NPs before AMF irradiation. Data are the mean ± SD ($n = 3$). Significant difference: *** $P < 0.001$. N.S.: no significant difference.

almost all the cells were alive [Figure 5A]. After AMF irradiation, the cells cultured without Fe₃O₄ NPs were still alive, while almost all the cells cultured with Fe₃O₄-5, Fe₃O₄-10, and Fe₃O₄-20 were dead. The results indicated that the cells cultured with free Fe₃O₄ NPs at a concentration of 5, 10, and 20 mg mL⁻¹ were completely killed after AMF irradiation. Some of the dead cells cultured with 20 mg mL⁻¹ free Fe₃O₄ NPs detached from the culture wells.

The WST-1 assay showed that the cells cultured without or with free Fe₃O₄ NPs had the same high viability before AMF irradiation [Figure 5B]. After AMF irradiation, the viability of cells cultured without free Fe₃O₄ NPs did not change significantly, while the viability of cells cultured with free Fe₃O₄ NPs significantly decreased after AMF irradiation. Viability of cells cultured with 5, 10, and 20 mg mL⁻¹ free Fe₃O₄ NPs decreased to 10.3% ± 6.9%, 3.1% ± 5.8%, and 2.7% ± 5.2%, respectively. All the live/dead staining and WST-1 assay results indicated that the breast cancer cells could be killed by the free Fe₃O₄ NPs under AMF irradiation. A higher concentration of free Fe₃O₄ NPs resulted in a higher killing effect. The killing effect of free Fe₃O₄ NPs should be due to the high temperature generated by free Fe₃O₄ NPs during AMF irradiation [Figure 4]. A higher concentration of the Fe₃O₄ NPs could generate higher temperatures and enhance the killing efficiency.

Anticancer effect of agarose/Fe₃O₄ hydrogels

The interaction between cells and agarose/Fe₃O₄ hydrogels should be different from that between cells and free Fe₃O₄ NPs. The cells could be near the hydrogels without adhesion to the hydrogels. The cells could also

be far away from the hydrogels or directly adhere to the hydrogels. To simulate these interactions between the breast cancer cells and hydrogels, three culture modes were used to investigate the MH anticancer effect of the agarose/Fe₃O₄ hydrogels. When the MDA-MB-231-Luc cells were cultured near the agarose/Fe₃O₄ hydrogel discs (sitting mode), most of the cells near the agarose/Fe₃O₄-5 hydrogel discs were dead, and cell viability decreased significantly after AMF irradiation [Figure 6A and B]. Almost all the cells near the agarose/Fe₃O₄-10 and agarose/Fe₃O₄-20 were dead, and their viability further decreased to the lowest level. The cell viability near agarose/Fe₃O₄-5, agarose/Fe₃O₄-10, and agarose/Fe₃O₄-20 decreased to 11.3% ± 4.8%, 3.9% ± 2.2%, and 3.6% ± 5.1%, respectively, after AMF irradiation [Figure 6B]. However, the cells near the agarose hydrogel discs without Fe₃O₄ NPs were alive, and their viability did not change before and after AMF irradiation.

When the breast cancer cells were cultured far away from the agarose/Fe₃O₄ hydrogel discs (transwell mode), the agarose/Fe₃O₄ hydrogels with a high concentration of Fe₃O₄ NPs (agarose/Fe₃O₄-10 and agarose/Fe₃O₄-20) could kill almost all the cells and significantly decrease cell viability to very low level by AMF irradiation [Figure 6C and D]. However, only a small number of cells cultured with the agarose/Fe₃O₄-5 were killed by AMF irradiation. After AMF irradiation, cell viability in the transwell of agarose/Fe₃O₄-5, agarose/Fe₃O₄-10, and agarose/Fe₃O₄-20 decreased to 85.7% ± 5.7%, 6.0% ± 3.5% and 2.8% ± 5.9%, respectively [Figure 6D].

The breast cancer cells adhered on the agarose/Fe₃O₄ hydrogel discs (adhesion mode) were most efficiently killed by AMF irradiation. Almost all the cells cultured with all the agarose/Fe₃O₄ hydrogel discs were dead, and their viability was the lowest compared to the sitting and transwell modes [Figure 6E and F]. Viability of breast cancer cells adhered on the agarose/Fe₃O₄-5, agarose/Fe₃O₄-10, and agarose/Fe₃O₄-20 decreased to 7.1% ± 1.2%, 2.7% ± 4.6%, and 2.3% ± 6.5%, respectively, after AMF irradiation [Figure 6F]. This should be due to the direct heating effect of the cells by the agarose/Fe₃O₄ hydrogels when the cells adhered to the hydrogels.

Anticancer effect of gelatin/Fe₃O₄ porous scaffolds

The anticancer effect of gelatin/Fe₃O₄ porous scaffolds was investigated by using the same methods as those used for the agarose/Fe₃O₄ hydrogels [Figure 7]. The cells cultured with the gelatin porous scaffold before and after AMF irradiation and with the gelatin/Fe₃O₄ porous scaffolds before AMF irradiation were alive with high viability. However, after AMF irradiation, some dead cells were observed in the gelatin/Fe₃O₄-5 and gelatin/Fe₃O₄-10, and almost all the cells were dead in the gelatin/Fe₃O₄-20 when the cells were cultured near the scaffolds [Figure 7A] or adhered in the scaffolds [Figure 7E]. When the cells were cultured far away from the scaffolds, a small part of the cells cultured with the gelatin/Fe₃O₄-20 were killed [Figure 7C]. After AMF irradiation, the viability of breast cancer cells cultured with gelatin/Fe₃O₄-5, gelatin/Fe₃O₄-10, and gelatin/Fe₃O₄-20 was 94.2% ± 9.1%, 80.3% ± 7.8%, and 6.2% ± 5.2% in sitting modes [Figure 7B], 99.8% ± 5.0%, 96.0% ± 5.1%, and 87.0% ± 3.4% in transwell modes [Figure 7D] and 90.8% ± 2.5%, 67.8% ± 3.2%, and 4.7% ± 3.4% in adhesion modes [Figure 7F], respectively.

The above results indicated that the breast cancer cells could be killed by either free Fe₃O₄ NPs, agarose/Fe₃O₄ hydrogels, or gelatin/Fe₃O₄ porous scaffolds. However, the anticancer effect depended on the matrices and interactions. The free Fe₃O₄ NPs showed the highest killing effect, while the gelatin/Fe₃O₄ porous scaffolds showed the lowest. Matrices used to embed Fe₃O₄ NPs showed an inhibitory influence on the killing effect of Fe₃O₄ NPs. In particular, the porous scaffolds had a more inhibitory influence than the hydrogels. This should be due to the influence of matrices on the magnetic thermal property of the Fe₃O₄ NPs. The interaction between breast cancer cells and Fe₃O₄ NP-embedded hydrogels or porous scaffolds

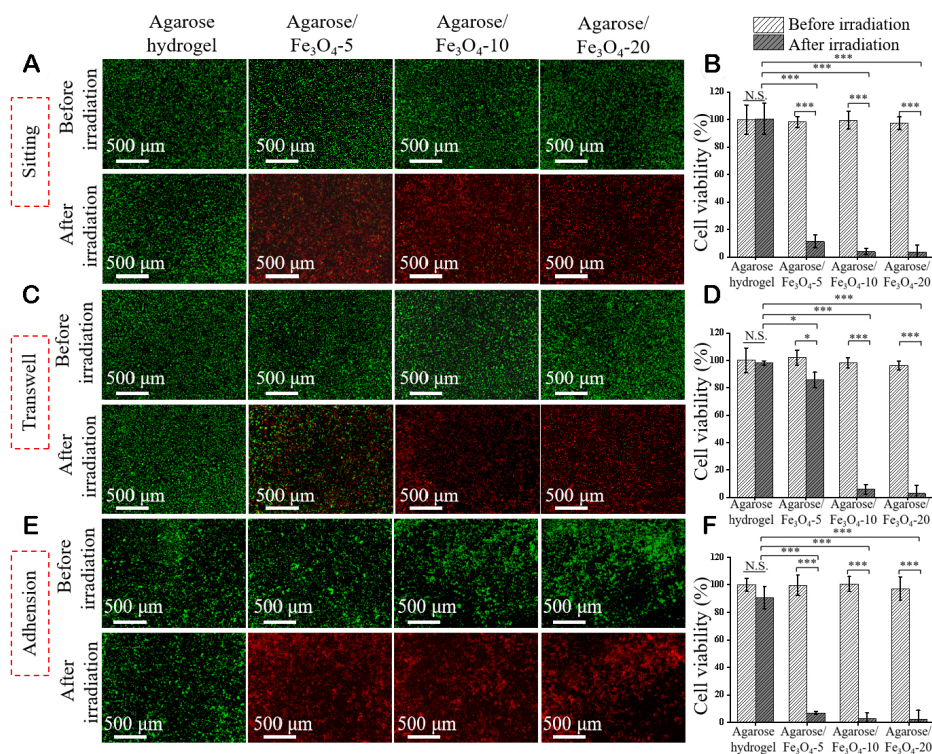


Figure 6. Anticancer effect of agarose/ Fe_3O_4 hydrogels. Live/dead staining of MDA-MB-231-Luc cells cultured with the agarose/ Fe_3O_4 hydrogel discs via sitting (A), transwell (C), and adhesion (E) modes before and after AMF irradiation (live cell: green fluorescence, dead cells: red fluorescence). Quantified cell viability during culture without or with agarose/ Fe_3O_4 hydrogel discs via sitting (B), transwell (D), and adhesion (F) modes before and after AMF irradiation. Cell viability was normalized to that cultured with agarose hydrogel without free Fe_3O_4 NPs before AMF irradiation. Data are the mean \pm SD ($n = 3$). Significant difference: * $P < 0.05$, *** $P < 0.001$. N.S.: no significant difference.

could also affect their killing effect. The cells directly adhered to the hydrogels or in the porous scaffolds could be more easily killed by AMF irradiation. The cells that were far away from the hydrogels or scaffolds were less affected by the heat generated by the Fe_3O_4 NP-loaded hydrogels or porous scaffolds.

DISCUSSION

Fe_3O_4 NPs have been widely used for MH because of their excellent magnetic-thermal conversion property. Investigation of the influence of matrices surrounding Fe_3O_4 NPs on the magnetic-thermal conversion property and anticancer effect of Fe_3O_4 NPs is important for the biomedical application of Fe_3O_4 NPs to maximize their therapeutic effect. In this study, the magnetic-thermal conversion property and anticancer effect of free Fe_3O_4 NPs and Fe_3O_4 NPs embedded in agarose hydrogels and gelatin porous scaffolds were compared because agarose hydrogels and gelatin porous scaffolds have been frequently used to embed therapeutic drugs and NPs. During AMF irradiation, the temperature change of free Fe_3O_4 NPs, agarose/ Fe_3O_4 , and gelatin/ Fe_3O_4 was positively correlated with the concentration of Fe_3O_4 NPs and irradiation time. Increasing the concentration of Fe_3O_4 NPs resulted in an increase of temperature alteration.

The matrices used to embed the Fe_3O_4 NPs could affect the magnetic thermal properties of Fe_3O_4 NPs. The free Fe_3O_4 NPs showed the best magnetic thermal properties. Embedding in hydrogels or porous scaffolds decreased the temperature change of Fe_3O_4 NPs during AMF irradiation. The temperature change of Fe_3O_4

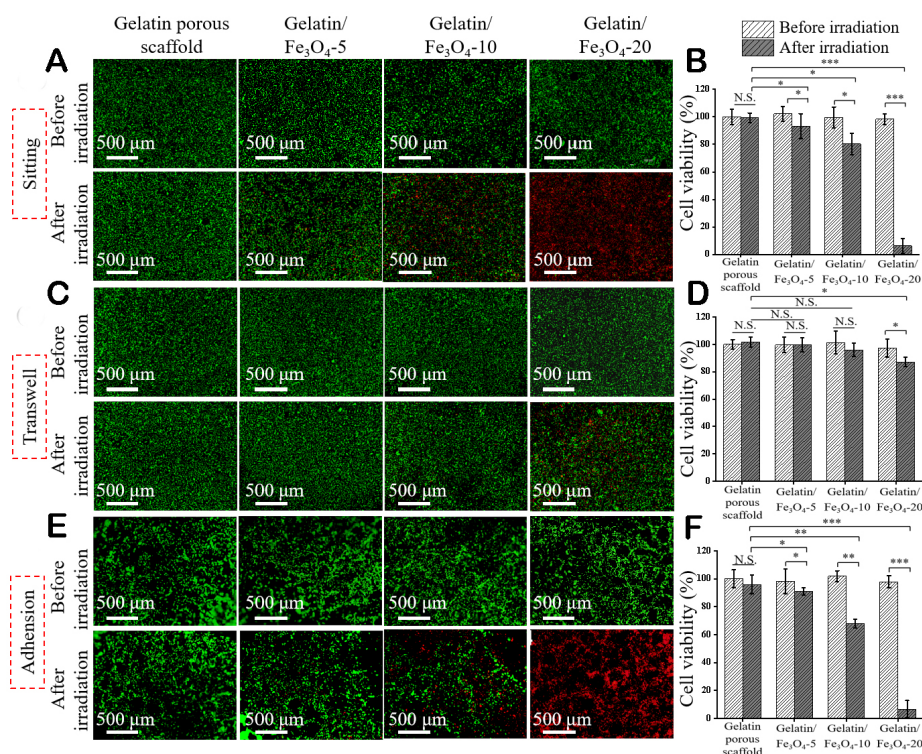


Figure 7. Anticancer effect of gelatin/Fe₃O₄ porous scaffolds. Live/dead staining of MDA-MB-231-Luc cells cultured with gelatin/Fe₃O₄ scaffold discs via sitting (A), transwell (C), and adhesion (E) modes before and after AMF irradiation (live cells: green fluorescence, dead cells: red fluorescence). Quantified cell viability after culture with gelatin/Fe₃O₄ scaffold discs via sitting (B), transwell (D), and adhesion (F) modes before and after AMF irradiation. Cell viability was normalized to that cultured with gelatin porous scaffold without free Fe₃O₄ NPs before AMF irradiation. Data are the mean ± SD (n = 3). Significant difference: *P < 0.05, **P < 0.01, ***P < 0.001. N.S.: no significant difference.

NPs in agarose hydrogels and porous scaffolds was decreased to 51.5%-59.5% and 20.1%-23.8% of that of free Fe₃O₄ NPs, respectively. The results of Fe₃O₄ NPs embedded in hydrogels were the same as the previously reported influence on their magnetic thermal property. Embedding in porous scaffolds further decreased the magnetic-thermal conversion capacity of Fe₃O₄ NPs.

The matrix influence should be due to the variation of Brownian relaxation of Fe₃O₄ NPs in the matrices. The heat generation mechanism of MNPs exposed to AMF includes Néel relaxation and Brownian relaxation^[62]. Néel relaxation refers to the heating due to the energy loss produced by the rotation of individual magnetic moments within the MNPs under AMF irradiation, and Brownian relaxation refers to the rotation of entire MNPs to produce heat^[63,64]. The Néel relaxation of Fe₃O₄ NPs in the hydrogels and porous scaffolds might not change. However, the matrices should affect the Brownian relaxation of Fe₃O₄ NPs. The Brownian relaxation of Fe₃O₄ NPs under AMF irradiation should be partially suppressed in the agarose hydrogel, leading to a decreased heating effect of Fe₃O₄ NPs. When the Fe₃O₄ NPs were embedded in gelatin porous scaffolds, the Fe₃O₄ NPs were tightly constrained in the gelatin fibers, and the Brownian relaxation of Fe₃O₄ NPs should be heavily inhibited. Therefore, the Fe₃O₄ NPs in the gelatin porous scaffolds generated heat predominantly through Néel relaxation.

Due to the influence of matrices on the magnetic thermal property of Fe₃O₄ NPs, the anticancer effect of Fe₃O₄ NPs was also dependent on the matrices. The breast cancer cells cultured with free Fe₃O₄ NPs were most efficiently killed by AMF irradiation. Embedding in agarose hydrogels and gelatin porous scaffolds

significantly decreased the killing effect of Fe_3O_4 NPs. Embedding in gelatin porous scaffolds showed the lowest killing effect. The decreased anticancer effects of Fe_3O_4 NPs embedded in agarose hydrogels and gelatin porous scaffolds could be explained by the decreased temperature change under AMF irradiation. The results of this study revealed that the magnetic thermal property and anticancer effect of Fe_3O_4 NPs were affected by their surrounding microenvironmental matrices. Although cancer cells and normal cells have different cytoplasmic properties^[65,66] and cancer cells are more sensitive to heat compared to normal cells^[67,68], the heat generated by Fe_3O_4 NPs under AMF irradiation should also affect viability of normal cells. The influence of Fe_3O_4 NP-embedded hydrogels and porous scaffolds on normal cell viability needs further investigation for controlling temperature to efficiently ablate cancer cells while minimizing negative effects on normal cells.

Furthermore, the interaction between the breast cancer cells and the Fe_3O_4 NPs could affect the killing effect. The free Fe_3O_4 NPs could be uptaken by breast cancer cells and generate heat inside the cells under AMF irradiation. When the Fe_3O_4 NP-embedded agarose hydrogels were applied, the breast cancer cells could be adhered to the hydrogel, near the hydrogels, or far away from the hydrogels. For the Fe_3O_4 NP-embedded gelatin porous scaffolds, breast cancers could enter the scaffolds and adhere in the scaffolds, near the scaffolds, or far away from the scaffolds. Three cultured models (sitting, transwell, and adhesion modes) were used to simulate the interaction between the cells and matrices. The Fe_3O_4 NPs embedded in the agarose hydrogels and gelatin porous scaffolds should be less or not uptaken by the cells if the hydrogels and scaffolds were not degraded. Therefore, the heat should be generated by the scaffolds and then transmitted to the breast cancer cells for ablation. The breast cancer cells adhered to the hydrogels or in the scaffolds were most efficiently ablated. The cells far away from the hydrogels and porous scaffolds were less affected. The results should be due to the heat transmission effect of the hydrogels and porous scaffolds.

CONCLUSION

In this study, the magnetic thermal property and ablation effect of free Fe_3O_4 NPs and Fe_3O_4 NPs embedded in agarose hydrogels and gelatin porous scaffolds were investigated to elucidate the influence of microenvironmental matrices on these properties. The flower-like Fe_3O_4 NPs were embedded in agarose hydrogels and gelatin porous scaffolds. Their magnetic thermal property and anticancer effects were compared with those of free Fe_3O_4 NPs. Under AMF irradiation, the free Fe_3O_4 NPs showed the highest temperature increase. Embedding in agarose hydrogels and gelatin porous scaffolds inhibited the heating capacity of Fe_3O_4 NPs and decreased the temperature change. The gelatin porous scaffolds had the highest inhibitory influence. The anticancer effect of Fe_3O_4 NPs was also dependent on the matrices. The free Fe_3O_4 NPs could most efficiently kill breast cancer cells under AMF irradiation. However, the ablation capacity of Fe_3O_4 NPs embedded in the agarose hydrogels and gelatin porous scaffolds significantly decreased under AMF irradiation compared to that of free Fe_3O_4 NPs. The reduced killing capacity of Fe_3O_4 NPs in agarose hydrogels and gelatin porous scaffolds was due to the inhibitory effect of the matrices on their magnetic thermal property. These results suggested that the matrices surrounding MNPs could affect the magnetic thermal property of MNPs and, therefore, affect their ablation capacity to cancer cells. The results should provide useful information for the design and application of MNPs for biomedical applications.

DECLARATIONS

Authors' contributions

Made substantial contributions to the conception and design of the study and performed data analysis and interpretation: Wang M, Sun R, Kawazoe N, Chen G

Performed data acquisition and provided administrative, technical, and material support: All authors

Availability of data and materials

Data will be made available upon request.

Financial support and sponsorship

This research was supported by the JSPS KAKENHI Grant Number 19H04475.

Conflicts of interest

All authors declared that there are no conflicts of interest.

Ethical approval and consent to participate

Not applicable.

Consent for publication

Not applicable.

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