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Low neutron cross-section FeCrVTiNi based high-entropy alloys: design, additive manufacturing and characterization

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Abstract

The development of high-entropy alloys (HEAs) based on the novel alloying concept of multi-principal components presents opportunities for achieving new materials with desired properties for increasingly demanding applications. In this study, a low neutron cross-section FeCrVTiNi-based HEA was developed for potential nuclear applications. A face-centred cubic (FCC) HEA with the nominal composition of FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3} is proposed based on the empirical thermodynamic models and the CALPHAD calculation. Verifications of the predictions were performed, including the additive manufacturing of the proposal material and a range of microstructural characterizations and mechanical property tests. Consistent with the prediction, the as-fabricated HEA consists of a dominant FCC phase and minor Ni₃Ti precipitates. Moreover, significant chemical segregation in the alloy, as predicted by the CALPHAD modelling, was observed experimentally in the produced dendritic microstructure showing the enrichment of Ni and Ti elements in the interdendritic regions and the segregation of Cr and V elements in the dendritic cores. Heterogeneous mechanical properties, including microhardness and tensile strengths, were observed along the building direction of the additively manufactured HEA. The various solid solution strengthening effects, due to the chemical segregation (in particular Cr and V elements) during



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solidification, are identified as significant contributing factors to the observed mechanical heterogeneity. Our study provides useful knowledge for the design and additive manufacturing of compositionally complex HEAs and their composition-microstructure-mechanical property correlation.

Keywords: High-entropy alloys, materials design, additive manufacturing, microstructures, materials characterization

INTRODUCTION

Developing structural materials that can withstand the harsh service environments of current and next-generation nuclear reactors, including high-radiation dose, high-temperature and corrosion, remains to be one of the main challenges in achieving their safe and economic operations. The structural materials and components for nuclear applications need to have a combination of low neutron cross-section, high radiation and corrosion resistance, good high-temperature mechanical performance, as well as favorable dimensional stability and resistance to swelling, flaking and thermal fatigue^[1,2]. Conventional alloys, including steels, nickel alloys, and zirconium alloys, have been extensively used in the nuclear industry for a variety of applications such as steam generators, pressure vessels, fuel assemblies and other reactor in-core components because of their compatibility and durability^[3]. However, they suffer from the problems of stress corrosion cracking, precipitation embrittlement and void swelling under reactor operational environments, which limit their service lifetime^[3-5]. Therefore, it is important to design and develop new alloys outside the paradigms of conventional alloys to advance next-generation nuclear technology applications.

High-entropy alloys (HEAs) have been emerging as a new class of materials since 2004^[6], which normally contain four to five principal elements and form a simple face-centred cubic (FCC), or body-centred cubic (BCC) single-phase solid-solution or a dual-phase of FCC + BCC. They have attracted significant interest in scientific and industrial communities because of their unique alloy design concept and promising properties under a range of environmental conditions^[7-9]. In particular, stimulated by the promising radiation resistance of HEAs essentially resulting from the severe lattice distortion, sluggish diffusion and chemical disorder in the materials^[10-12], research efforts have been devoted to developing particular HEA systems (reduced activation and/or low neutron cross-section compositions) with potentials for nuclear applications. For example, Ayyagari *et al.*^[13] developed the reduced activation Ta-Ti-V-Zr-X (X = Hf or W) HEAs of BCC structure by a combined experimental and molecular dynamics approach. King *et al.*^[14] proposed the Nb-Ti-V-Zr based HEAs, which consist of the dominant BCC phase as the high-temperature, low neutron cross-section materials for nuclear applications. Although the BCC phase normally contributes to improving the material's strength and forms the constitutions of refractory HEAs compatible for high-temperature applications, single-phase BCC HEAs are shown to have larger irradiation-induced volume swelling than FCC HEAs^[15], which are in contrast to the case of conventional alloys^[16]. As such, the development of low neutron cross-section, FCC HEAs with a good balance of strength and ductility together with superior dimensional stability under irradiation environments would be technologically important for future nuclear applications.

A comprehensive evaluation of the materials' properties (including the mechanical performance and radiation damage resistance) is necessary before certifying the candidate materials for nuclear applications. However, the first step is to achieve the design and manufacturing of the materials (especially for new materials) and establish an understanding on the relationship of the processing-microstructure-mechanical performance. The present work is towards the end of this first step. In this work, we target at developing

FCC HEAs based on a pentanary FeCrVTiNi system with the constituting elements of relatively low neutron cross-sections (approximately 2.5-6 barn), as tabulated in Table 1^[17,18]. We agree that future studies looking into the comprehensive materials properties (such as creep and radiation damage resistance) are needed. A combination of the currently known empirical calculation models for predicting the formation and stability of single-phase HEAs^[19] and the CALculation of PHase Diagrams (CALPHAD) approach with the Thermo-Calc program was utilized in the alloy design. To verify the predictions, a selected HEA of predicted FCC structure with a nominal composition of FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3} was produced experimentally using a new powder-bed arc additive manufacturing (PAAM) approach established in our previous work^[20]. This additive manufacturing approach is advantageous for its higher production rate and capability of fabricating components with complex geometries compared with conventional processes of arc melting or casting. However, due to its inherent characteristics of relatively high cooling rate and complex thermal history associated with the layer-by-layer depositions^[20-22], the produced microstructure and resultant properties of additively manufactured materials are often distinctive from those of the counterparts by conventional processes. Hence, the microstructure of the PAAM produced HEAs was characterized and correlated with the mechanical properties. The results show that a dominant FCC microstructure with minor Ni₃Ti precipitates was achieved for the designed HEA material. Additionally, a significant local compositional fluctuation exists along the building direction of the produced HEA, which is believed to result from the chemical segregation that occurred during the PAAM process and is responsible for the observed variations of the mechanical properties along the building direction.

MATERIALS AND METHODS

Materials design

The formation and stability of the solid-solution phases in HEAs are determined by their thermodynamics^[23]. There have been several empirical models^[18,23-26] predicting the crystalline structure and stability of the solid-solution phases in HEAs based on various thermophysical parameters, including the entropy of mixing (ΔS_{mix}), enthalpy of mixing (ΔH_{mix}), atomic size differences (δ), valence electron concentration (VEC), Pauling electronegativity difference ($\Delta\chi$) and the Ω parameter which is defined in the following:

$$\Omega = \frac{T_m \Delta S_{\text{mix}}}{|\Delta H_{\text{mix}}|} \quad (1)$$

where T_m is the melting temperature of the alloys estimated from the rule of mixtures of all elements. The entropy of mixing (ΔS_{mix}) of the multi-component HEAs is defined as:

$$\Delta S_{\text{mix}} = -R \sum_{i=1}^n c_i \ln c_i \quad (2)$$

where R is the ideal gas constant and c_i is the atomic percentage of the i th element. It has been well-accepted that the entropy of mixing (ΔS_{mix}) is not the only factor that determines the solid-solution single-phase formation in the HEAs^[27,28]. Hence, the other empirical parameters are also considered in the present work. The enthalpy of mixing (ΔH_{mix}) for a system is calculated as:

$$\Delta H_{\text{mix}} = \sum_{i=1, i \neq j}^n \Omega_{ij} c_i c_j \quad (3)$$

where $\Omega_{ij} = 4\Delta H_{AB}$, and ΔH_{AB} is the mixing enthalpy of binary AB alloys in the liquid state determined using Miedema's model^[29]. The VEC of the HEA system is calculated as:

$$\text{VEC} = \sum_{i=1}^n c_i (\text{VEC})_i \quad (4)$$

where $(\text{VEC})_i$ is the VEC for the i th element. The atomic size difference (δ) is defined as:

$$\delta = 100 \sqrt{\sum_{i=1}^n c_i \left(1 - \frac{r_i}{\bar{r}}\right)^2} \quad (5)$$

where $\bar{r} = \sum_{i=1}^n c_i r_i$, with c_i and r_i being the atomic percentage and the atomic radius of the i th element, respectively. The Pauling electronegativity difference ($\Delta\chi$) parameter is defined as:

$$\Delta\chi = \sqrt{\sum_{i=1}^n c_i (\chi_i - \bar{\chi})^2} \quad (6)$$

$$\bar{\chi} = \sum_{i=1}^n c_i \chi_i \quad (7)$$

where χ_i is the Pauling electronegativity of the i th element.

According to previous studies^[18,23-26], a single FCC solid-solution phase tends to form in a HEA while the empirical parameters of this alloy system fall into specific ranges of $0 \leq \delta \leq 8.5$, $-22 \leq \Delta H_{\text{mix}} \leq 7$ kJ/mol, $11 \leq \Delta S_{\text{mix}} \leq 19.5$, $\text{VEC} > 8$ and $\Omega \geq 1.1$. We have determined a range of compositions based on the FeCrVTiNi system, which meet these criteria. Details of the determined compositions and their parameter values can be found in [Supplementary Table 1](#). Moreover, we considered that in the investigated HEA system, the Ni-Ti has a low enthalpy of mixing which means their intermetallic is likely to form in the alloy system. In order to minimize the potential formation of Ni-Ti intermetallics in the alloys, we selected a HEA composition of FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3} with the lowest Ni content among the designed 57 potential compositions for further investigations in this work. The choice of a low Ni content in the designed alloy is also beneficial for reducing its potential activation after irradiation. The calculated empirical parameters for the selected FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3} alloy composition are listed in [Table 2](#). To verify the alloy design based on the empirical models, CALPHAD calculations of the FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3} system using the Thermal-Calc® software were further performed. As the Ni content is the highest in this composition, we simply took this alloy as a Ni-based alloy, and the preliminary CALPHAD calculations were based on the TCNI10 thermodynamic and MOBNI4 mobility databases for Ni-alloys.

Table 1. Neutron cross-section (σ), atomic radius (r), Pauling electronegativity (χ) and VEC of five elements in the designed HEAs of this work^[17,18]

Element	Fe	Cr	V	Ti	Ni
σ (barn)	2.56	3.1	5.06	6.1	4.5
r (Å)	1.241	1.249	1.316	1.462	1.246
χ	1.83	1.66	1.63	1.54	1.91
VEC	8	6	5	4	10

VEC: Valence electron concentration; HEAs: high-entropy alloys.

Table 2. Solid-solution phase stability parameter (Ω), the entropy of mixing (ΔS_{mix}), enthalpy of mixing (ΔH_{mix}), VEC, atomic size difference (δ), Pauling electronegativity difference ($\Delta\chi$) for the designed $FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3}$ HEA

Parameters	Ω	ΔS_{mix}	ΔH_{mix}	VEC	δ	$\Delta\chi$
$FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3}$	2.54	11.51	-7.226	8.03	4.35	0.12

VEC: Valence electron concentration; HEA: high-entropy alloy.

Additive manufacturing and characterization

To verify the alloy design experimentally, a new PAAM approach was used to produce the $FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3}$ HEA. Figure 1 presents the schematics of the PAAM setup and the extraction of samples for characterization. As schematically shown in Figure 1A, the PAAM device consists of a commercial gas tungsten arc welding system, a shielding gas cover and the powder holder. A mild steel plate with dimensions of 340 mm \times 70 mm \times 10 mm (length \times width \times height) was used as the substrate. High purity (99.99 wt.%) Fe, Cr, Ni, Ti and V elemental metallic powders with similar particle sizes of 180 μ m average diameters (80 mesh) were physically mixed in a ball miller according to the nominal composition of $FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3}$. The blended powders were served as the powder-bed in the PAAW system, and a HEA thin-wall structured bulk material of 15 deposited layers and dimensions of approximately 125 mm \times 11 mm \times 25 mm (length \times width \times height) was fabricated. To maintain the deposition stability and minimize the spattering of metallic powders during processing, the arc current and arc length were set to 120 A and 3.5 mm, respectively. The travelling speed was fixed to 50 mm/min to guarantee sufficient heat input and avoid the discontinuity or hump morphology of the deposition bead. Note that before the deposition, pre-heating of the substrate was performed by one-off arc scanning of the substrate plate before loading the powders. This pre-heating process is to facilitate the deposition of initial layers. Argon shielding gas was applied throughout the deposition process to minimize protentional oxidation. Additionally, analogous to the conventional arc melting process^[30,31], each deposited layer was remelted once during the PAAM process aiming to improve the homogeneity of the deposited material. Here the remelting process refers to the remelting of each deposited layer during PAAM using the identical arc melting processing parameters without feeding a new layer of powders on top of the as-deposited layer.

Figure 1B schematically shows the extraction of samples from the as-fabricated $FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3}$ thin-wall-structure material for various tests. Note that in the PAAM processing, the powder feedstocks and the deposited layers were constrained within the powder holder (see Figure 1A). It is believed that there should not be any significant horizontal offset of the individual layers that occurred during processing. The various widths of individual layers of the as-fabricated sample, as seen from Figure 1B are presumably caused by the variation of the melt flowability due to different thermal cycling conditions during processing. A cross-sectional sample for metallographic observations was extracted from the middle section of the as-fabricated materials and prepared by mounting, grinding, and polishing followed by electrical etching at room temperature in a solution consisting of 10 mL oxalic acid and 100 mL distilled water at 6 V DC for 20 s. To

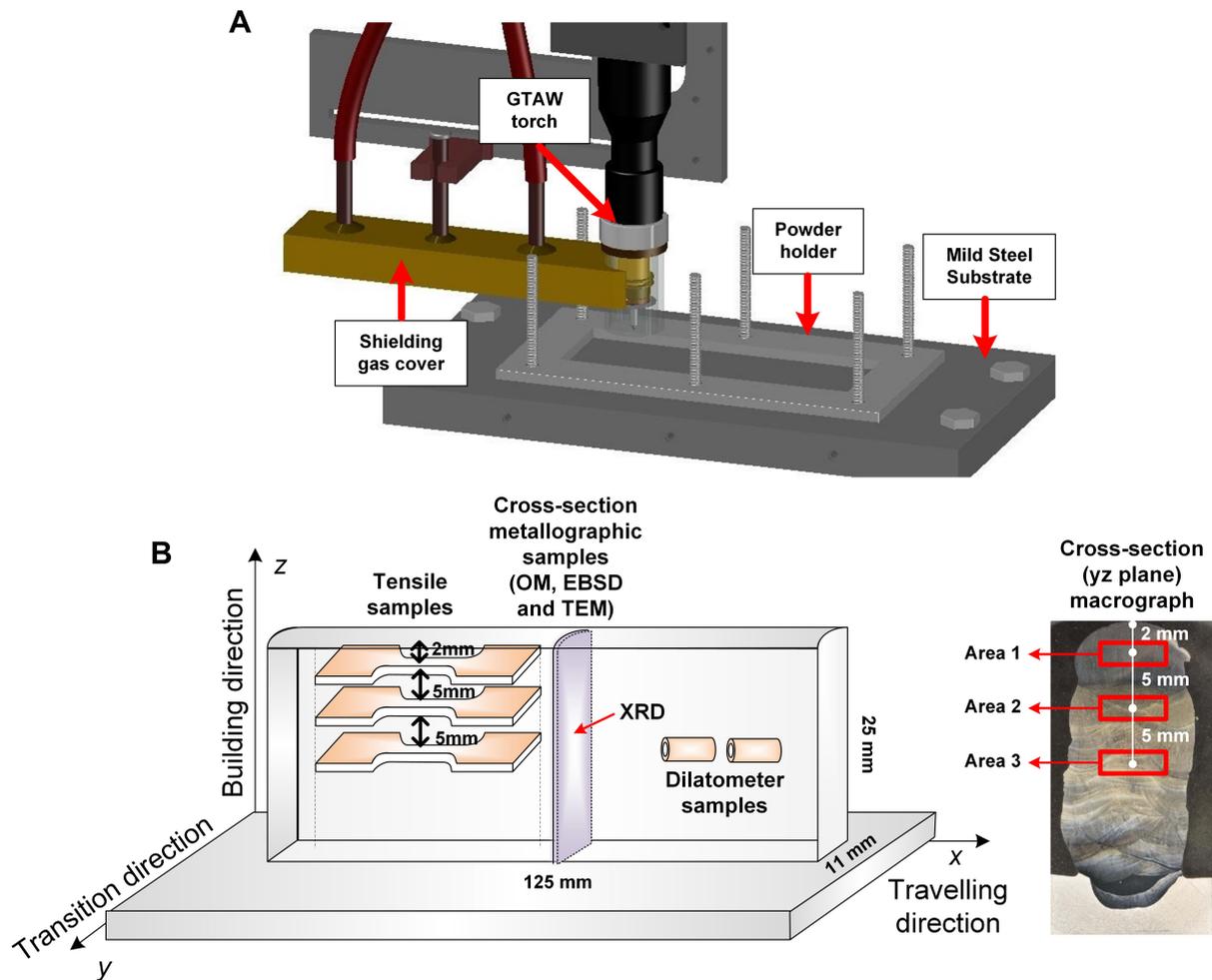


Figure 1. (A) Schematic of the powder-bed arc additive manufacturing (PAAM) system applied for producing high-entropy alloys (HEAs); (B) Schematic showing the dimensions of produced thin-wall HEA materials and the sampling locations for various characterization, together with a cross-sectional macrograph showing the areas for metallographic observations. Note that no macro-cracking or porosity was visible on the cross-section. OM: Optical microscope; EBSD: electron backscatter diffraction; TEM: transmission electron microscopy; XRD: X-ray diffraction.

avoid the possible compositional modification in the initial deposition layers due to the diffusion of elements such as Fe and C from the steel substrate into the deposited HEA, we focused on the middle and top parts of the deposited material. It is worth mentioning that as recognized in previous work^[32], not only the temperature gradient from subsequent deposition of additional layers but also the temperature cycling may cause the microstructural variations of additively manufactured materials and hence the heterogeneous materials mechanical properties along the building direction. Moreover, as this research involves a new material fabricated by a new process, it is considered necessary to understand if any differences exist in the microstructure and mechanical properties of various regions along the building direction (which experienced different thermal cycling). Three locations marked as Area 1, Area 2 and Area 3 in the cross-sectional sample (see Figure 1B) were selected for microstructural analysis. Preliminary metallographic observation was performed using a Nikon Eclipse LV100NDA Optical Microscope (OM). X-ray diffraction (XRD) was also performed on the cross-sectional sample for the phase identification using a GBC MMA X-ray diffractometer with CuK α radiation ($\lambda = 1.5406 \text{ \AA}$). Further observations were performed using a JEOL[®] JSM-6490LA scanning electron microscope (SEM) equipped with an X-Max energy dispersive X-ray

spectroscopy (EDS) detector. The EDS line scanning was conducted along the building direction from the top surface of the cross-sectional sample. Concurrent electron backscatter diffraction (EBSD) and EDS mappings were performed on the approximately middle region of the cross-sectional sample (Area 3) using a Zeiss® UltraPlus™ SEM equipped with EBSD and EDS detectors. Transmission electron microscopy (TEM) analyses were carried out using a JEM 2200FS TEM device operating at 200 kV, with the samples extracted from Area 3 and prepared by twin-jet polishing. In addition, the dilatometer measurements were performed by the Theta Industries Dilatoronic® Quence Dilatometer using hollow cylinder samples with dimensions of 10 mm length, 5 mm external diameter and 3.5 mm inner diameter. Two individual dilatometer tests were done in a temperature range from room temperature to 1250 °C at two different heating/cooling rates of 0.1 °C/s and 10 °C/s, respectively.

Three tensile samples were extracted from the HEA wall at the locations of 2 mm, 7 mm, and 12 mm away from the sample top surface, respectively, using electrical discharge machining. These sampling locations are consistent with those for metallographic observations (Areas 1, 2 and 3). The gauge volume of the tensile sample is 100 mm × 2 mm × 1.5 mm. An MTS370 load frame was used for the tensile tests at room temperature with a displacement rate of 0.6 mm/min. After the tensile tests, fractographic observation was performed using SEM. Moreover, Vickers microhardness measurement was performed by a Matsuzawa Via-F automatic Vickers hardness tester at a load of 500 N and a dwelling time of 10 s. The microhardness was measured along the central line of the cross-sectional sample at an interval of 0.5 mm.

RESULTS AND DISCUSSION

CALPHAD calculations

The CALPHAD modelling has been shown to be effective for predicting the phase formation and the solidification behavior in some HEA systems^[33,34]. Figure 2 presents the equilibrium phase diagram and solidification simulation results on the phase transformation and compositional segregation using Thermo-Calc for the $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ system. In agreement with the prediction from the empirical models, the Thermo-Calc calculated equilibrium phase diagram [Figure 2A] shows a dominant FCC phase in the designed alloy below 1300 °C. As expected, the Ni-Ti intermetallic in the form of Ni_3Ti is also predicted to form in the alloy below ~1070 °C, due to the low mixing enthalpy of Ti and Ni elements and resultant low Gibbs free energy of this intermetallic formation in the system. Taking into account the fast solidification process that occurred during PAAM, which is applied for producing the designed HEA in the present work, the non-equilibrium Scheil solidification simulation based on the Scheil-Gulliver model^[35] was performed. Figure 2B demonstrates the obtained liquid to solid phase transformation curve for the investigated HEA alloy, representing a fast solidification process away from the equilibrium. Furthermore, the elemental segregation during solidification of the HEA is calculated. As quantitatively shown in Figure 2C, with the progression of the solidification process in the studied HEA, the Ni and Ti elements appear to be enriched in the solid phase that forms in a later stage, whereas the other elements of Fe, Cr and V show depletion in the solid phase that solidifies subsequently. These simulations provide insights into the phase formation and the solidification process for the studied HEA system. In the following sections, the experimental characterization and analyses will be presented to further understand the microstructural characteristics and mechanical properties of the produced HEA using PAAM.

Microstructure and stability

Phase formation

Figure 3 shows the XRD pattern measured from the cross-sectional sample extracted from the produced HEA by PAAM. A dominant FCC phase is identified, as evidenced by three strong reflections of (111), (200) and (220) present in the observed 2θ range of 20°-80°. The lattice parameter of the FCC phase in the studied HEA is determined to be $a = \sim 3.60 \text{ \AA}$ larger than that of pure FCC Ni ($a = \sim 3.52 \text{ \AA}$), due to the solid-

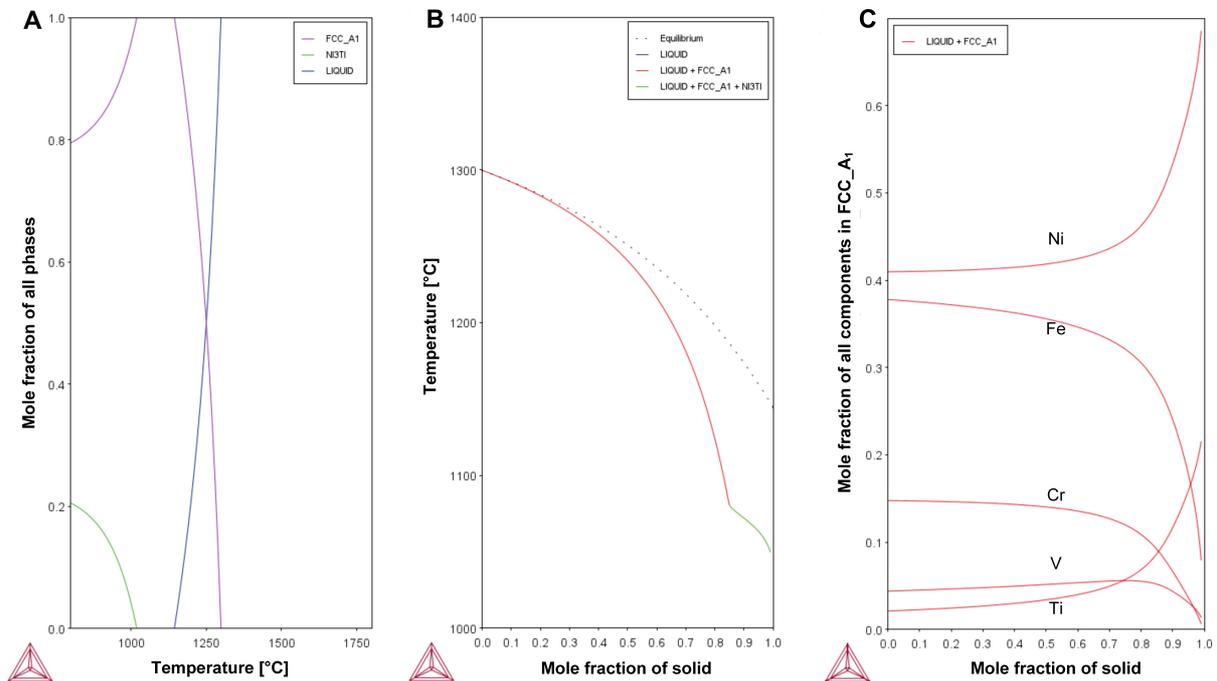


Figure 2. Thermo-Calc calculation results of the designed high-entropy alloy $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$: (A) equilibrium phase diagram; and Scheil solidification simulations showing the (B) phase transformation and (C) segregation of elements during solidification.

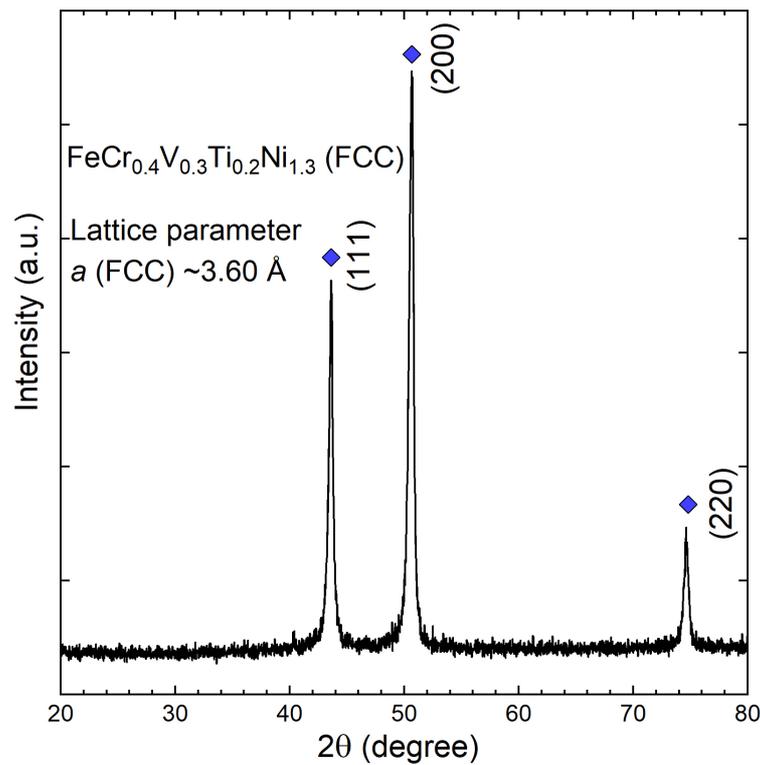


Figure 3. A typical X-ray diffraction pattern of the as-fabricated high-entropy alloy.

solution effect in the multi-component HEA. The Ni_3Ti phase in the HEA system predicted by Thermo-

Calc is not detected in the present XRD measurement, presumably due to the low volume fraction of this phase in the produced material, which is beyond the detection limit of the applied laboratory XRD.

Note that the present XRD analysis has limitations of only showing the dominant phase and its average structure in the HEA sample. To further understand the phase formation in the as-fabricated HEA material, TEM observations were performed on a typical sample extracted from Area 3 of the cross-sectional sample (as indicated in Figure 1). Figure 4 presents the obtained TEM results. Figure 4A shows a typical bright-field TEM image of the HEA sample together with a corresponding selected area electron diffraction (SAED) pattern under the $[112]$ zone axis. The SAED data in Figure 4A confirms the dominant phase in the HEA sample to be an FCC structure, consistent with the XRD result. Interestingly, it is occasionally observed that an ordered phase of L_{12} structure is present in the microstructure, as shown in Figure 4B. This finding is obtained from the observation of weak superlattice reflections in the corresponding SAED pattern (see inset in Figure 4B) collected under $[110]$ zone axis. This ordered L_{12} phase is not predicted in the equilibrium Thermo-Calc calculation [Figure 2A], suggesting that this phase is probably a metastable phase.

Figure 4C shows the needle-shaped precipitation observed in the microstructure. The corresponding EDS result shows that this precipitation is rich in Ni (~60 at.%) and Ti (~15 at.%), supporting the Thermo-Calc prediction [Figure 2B] for the formation of the minor Ni_3Ti type precipitation in the studied HEA. The Ni_3Ti type precipitation is also common in the maraging steels as a strengthening phase^[36]. Moreover, as seen in Figure 4D, the spherical precipitation with a diameter of ~0.3 μm is present. The EDS analysis shows that this spherical precipitate is rich in Ti (~42 at.%). As shown in the inset in Figure 4D, the high-resolution TEM data and the corresponding Fast Fourier Transform pattern were present for this spherical Ti-rich precipitate, revealing its Hexagonal-Closed Packed structure. The Ti-rich precipitation is not predicted in the equilibrium phase diagram by Thermo-Calc [Figure 2A], suggesting its non-equilibrium nature. Considering the low solid solubility of Ti in the Fe- and Ni-based solid-solution system, it is believed that during the PAAM processing of the present FeCrVTiNi-based HEA material, the oversaturated Ti solutes in the melt may precipitate out as the Ti-rich precipitates in the produced HEA during solidification. The Ti-rich precipitate was also found in a BCC-structured VCrMnTi-based HEA in the as-cast condition^[37]. Such precipitates may act as the getters for the O and H solutes in the alloys, potentially improving their resistance to oxidation and hydrogen embrittlement. However, the exact impact of the Ti-rich precipitation on the performance of the produced HEA remains to be understood.

Stability

To verify the stability of the phase(s) in the as-fabricated $FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3}$ alloy, two individual dilatometer tests were performed at the heating and cooling rates of 0.1 $^{\circ}C/s$ and 10 $^{\circ}C/s$, respectively. Figure 5 shows the recorded dilatometry curves. For both experiments, during the heating period, the strain of the dilatometer sample observed from its length changes almost linearly increases with increasing the temperature, suggesting that there is no evident phase transformation during the heating stage. Note that a smear-out slope discontinuity was clearly observed during cooling at 0.1 $^{\circ}C/s$ when the temperatures approach the range of ~850 $^{\circ}C$ to ~800 $^{\circ}C$. This slope change is not conclusively identifiable for the case at the rate of 10 $^{\circ}C/s$. According to the TEM investigation, the matrix of alloy consists of a disordered FCC phase and a small amount of ordered L_{12} structure. It is considered that the clear slope change in the dilatometer curve during cooling at a slow cooling of 0.1 $^{\circ}C/s$ was linked to the disorder to order phase transformation, similar to the observation of Ti_2AlNb alloy^[38]. It is also speculated that this phase transformation, as indicated by the slope change in the curve during cooling, would only readily occur at a relatively slow cooling for kinetics reasons. Because the additive manufacturing process used in this work involves a much higher cooling rate^[39], the FCC matrix mainly remained in a disordered state in the as-

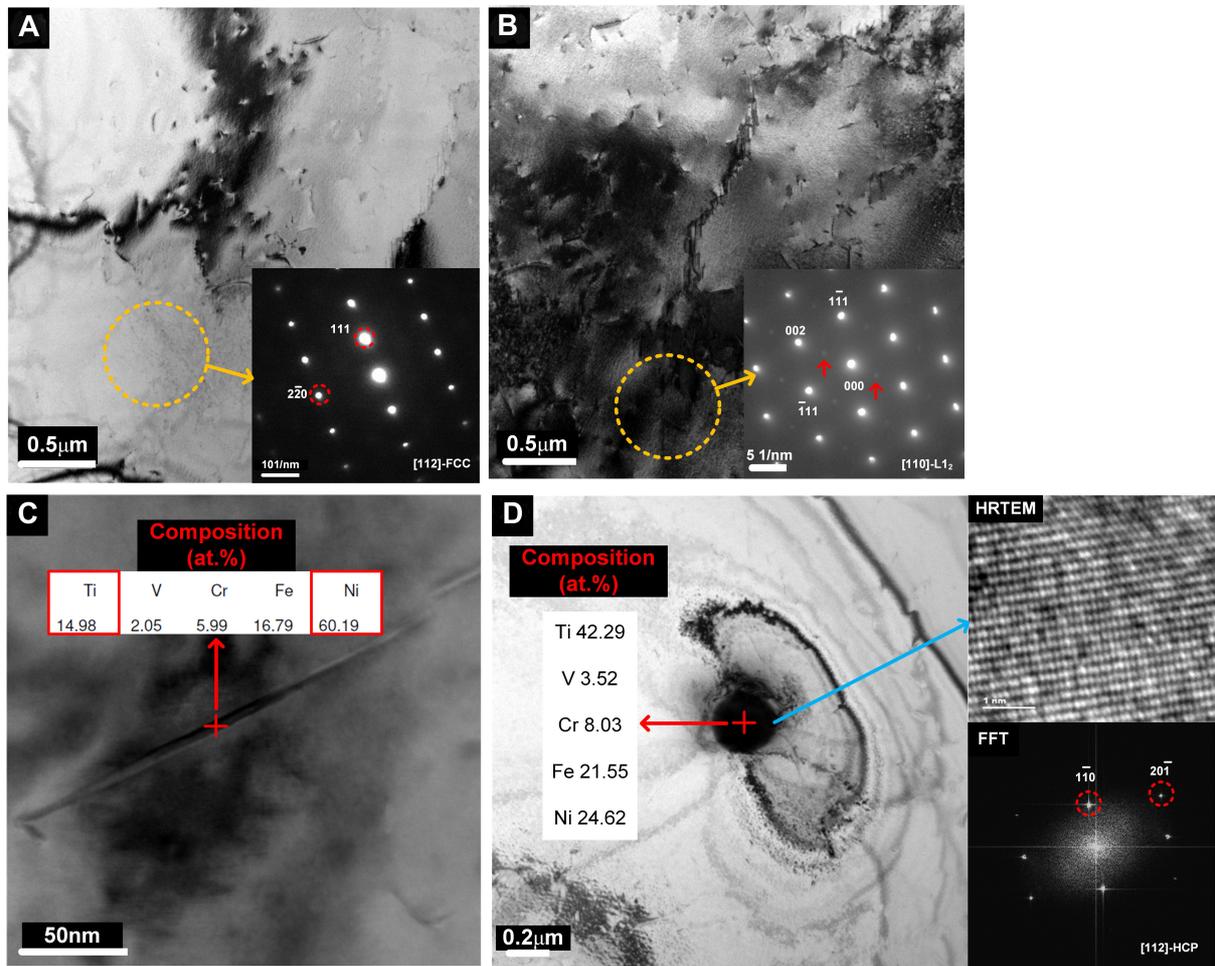


Figure 4. Transmission electron microscopy (TEM) observation of the produced high-entropy alloy (HEA) with the sample extracted from Area 3 (as indicated in Figure 1), showing the co-existence of matrix phases of the (A) disordered face-centred cubic (FCC) and (B) ordered L12 structures as well as the (C) needle-shaped Ni₃Ti type precipitation and (D) spherical Ti-rich precipitation. The insets in (A) and (B) are corresponding selected area electron diffraction (SAED) patterns. The energy dispersive X-ray spectroscopy (EDS) results from the precipitates are included in (C) and (D). High-resolution TEM (HRTEM) data and the fast Fourier transform (FFT) pattern from the spherical precipitation are also shown in (D).

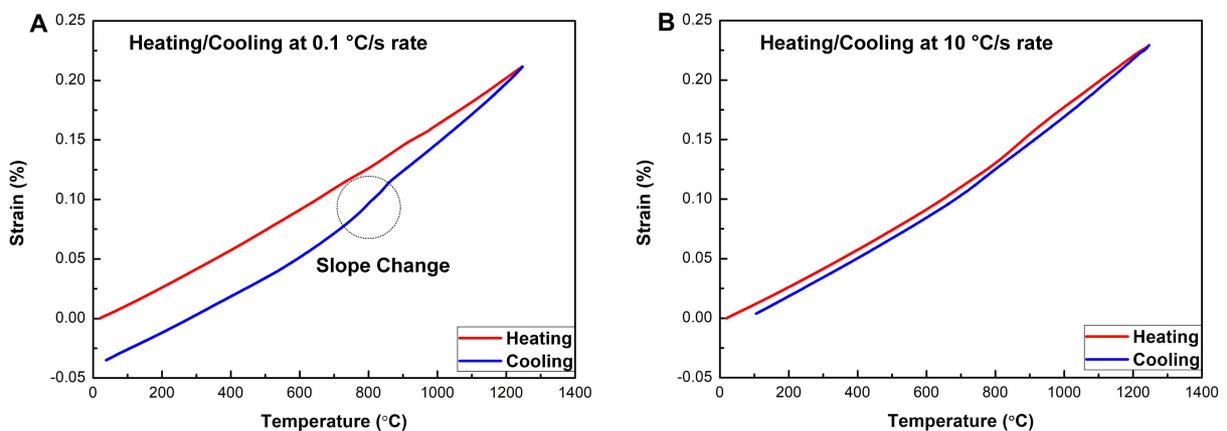


Figure 5. Dilatometry curves of the sample length changes (strain) as a function of temperature for the as-fabricated FeCr_{0.4}V_{0.3}Ti_{0.2}Ni_{1.3} high-entropy alloy at different rates of (A) 0.1 °C/s and (B) 10 °C/s.

fabricated $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ HEA. Therefore, during the heating period of the dilatometer test, the strain of both samples shows a quasi-linear relationship with temperature. However, when the cooling rate is relatively low at $0.1\text{ }^\circ\text{C/s}$, the alloy seems to be more inclined to form an ordered L_{12} structure with the disorder to order transformation (which is likely kinetically limited) occurred during cooling at around $850\text{ }^\circ\text{C}$.

Dendritic microstructure and chemical segregation

Figure 6 shows the typical dendritic microstructure of the produced HEA by PAAM, observed by OM on the cross-section of Areas 1, 2 and 3 (as indicated in Figure 1). In general, cellular-like dendrites were formed in Area 1 as shown in Figure 6A, while the columnar grains were developed in Areas 2 and 3 along the building direction as shown in Figure 6B and C, respectively. The fusion line produced by arc welding between different deposited layers in the direction perpendicular to the x-z plane can be readily observed. Above the layer boundary, it exhibited relatively small columnar grains, which are regarded as the columnar prior grains^[40] in Figure 6B. The length of the columnar grains in Areas 2 and 3 reaches up to $\sim 1\text{ mm}$, while the width of these grains is approximately $15\text{ }\mu\text{m}$. The driving force of epitaxial columnar grains with dendrites along the deposition direction growth comes from the heat transfer from subsequent layers, which follows the highest thermal gradient^[41]. The grains of preceding deposited layers are partially remelted by thermal cycling and format prior grains serving as the nucleus of the following grain growth. Therefore, the preferred grain growth direction is along the buildup direction, which is perpendicular to the direction of the solid-liquid interface^[42,43]. In contrast, for the last deposition layers, the lack of following thermal cycling and accelerated cooling induced by the atmosphere transfers the dendrites into equiaxed morphologies^[44].

To characterize the chemical segregation in the produced HEA as predicted by Thermo-Calc [Figure 2C], concurrent EBSD and EDS mappings across the dendritic microstructure on the cross-section (Area 3) of the as-fabricated HEA sample were performed, and the results are shown in Figure 7. The orientation map shows the development of small-angle grain boundaries (represented by fine lines) within the columnar grains. The corresponding EDS maps show that the elemental segregation across the dendrites occurred during solidification. Specifically, the Ni and Ti are positive segregation elements that are enriched in the interdendritic regions (solidified later compared with the dendritic cores), whereas the Fe and Cr are negative segregation elements that tend to segregate in the dendritic cores. Note that V is also a negative segregation element as predicted by Thermo-Calc [Figure 2C]. However, the segregation of V element in the microstructure was not clear from the experimental EDS map [Figure 7] due to its relatively low content in the produced $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ HEA material and the limit of the EDS detection resolution. Overall, these experimental characterizations are in line with the Thermo-Calc prediction [Figure 2C], providing evidence and description of the micro-segregation process that is occurred during the solidification of the produced HEA by PAAM.

Heterogeneous mechanical properties and their microstructural origins

The mechanical properties of the as-fabricated $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ HEA are accessed and correlated with their microstructural characteristics. To evaluate the homogeneity of the mechanical performance of the produced HEA material by PAAM, the microhardness was measured along the building direction on the cross-sectional sample, and the mechanical tensile tests were performed at room temperature on the samples extracted from different locations along the building direction. Figure 8 presents the obtained mechanical property results. As shown in Figure 8A, a significant variation of the microhardness ranging from 188 HV to 358 HV was observed along the building direction. The engineering stress-strain curves and tensile strengths and elongation results from the tensile tests on the samples extracted on similar heights of Areas 1, 2 and 3 (as indicated in Figure 1) are demonstrated in Figure 8B. Analogous to the microhardness results, distinctive tensile properties (strength and ductility) were observed at different sampling locations

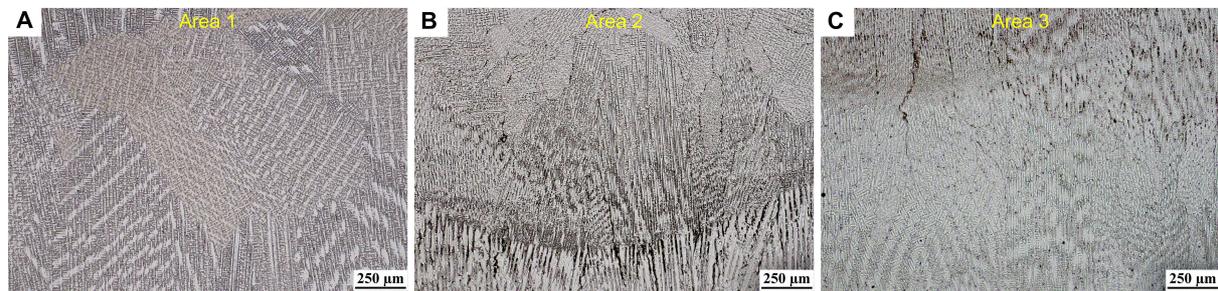


Figure 6. Cross-sectional optical micrographs of the as-fabricated $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ high-entropy alloy sample at selected areas indicated in Figure 1: (A) Area 1; (B) Area 2; and (C) Area 3.

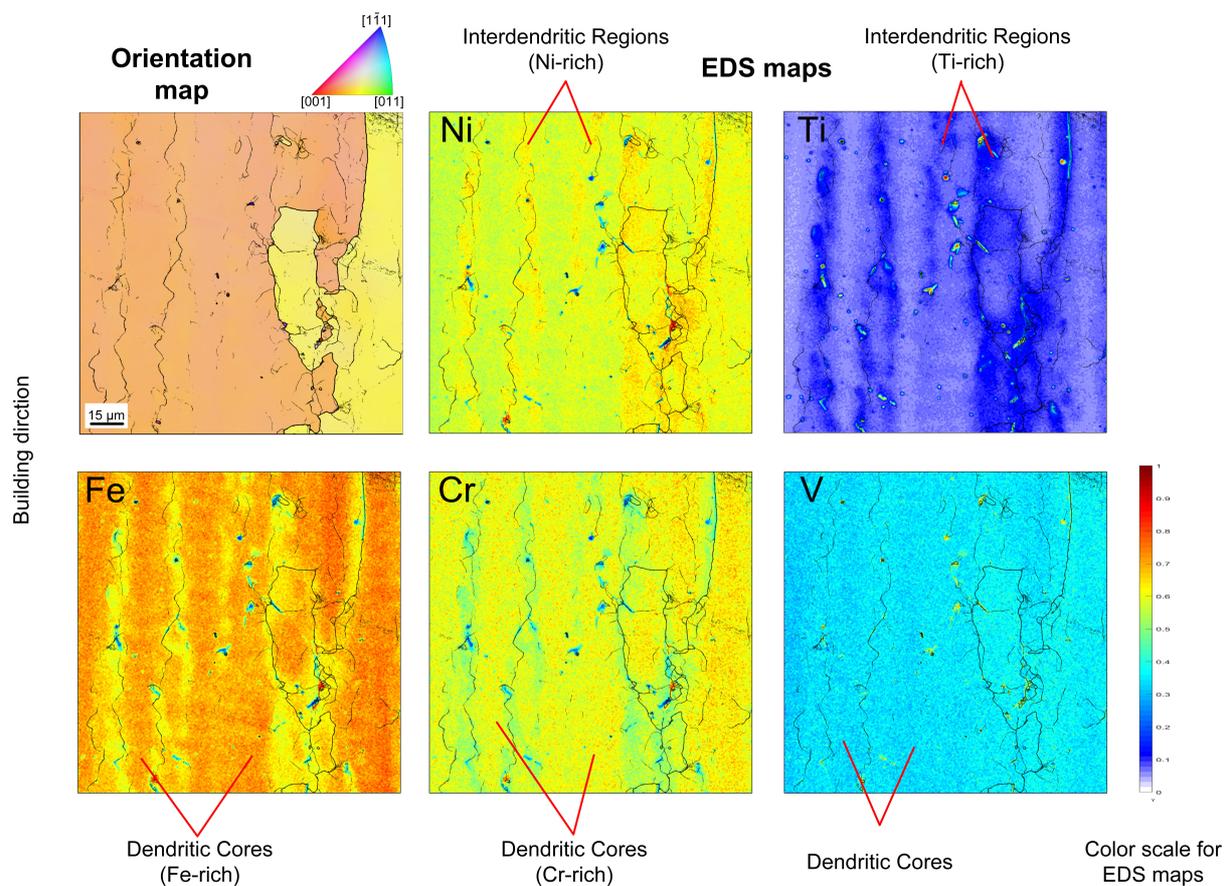


Figure 7. EBSD-based orientation map and corresponding energy dispersive X-ray spectroscopy (EDS) maps of Ni, Ti, Fe, Cr and V elements were collected from the cross-sectional sample in Area 3, as indicated in Figure 1. The thicker lines in the maps show the large-angle grain boundaries (misorientation angles $> 7^\circ$), and fine lines represent small-angle grain boundaries (misorientation angles 0.15° – 7°). The intensities of EDS maps are normalized to the same color scale. EBSD: Electron backscatter diffraction.

along the building direction. Specifically, the tensile test with the sample extracted at the height near the Area 1 shows the yield strength (YS), ultimate tensile strength (UTS) and elongation of ~ 468 MPa, ~ 719 MPa and 43%, respectively. Strengthening while at the expense of lowered ductility was observed for the samples away from the top surfaces of the as-fabricated HEA. At the locations near Area 2 and 3, the YS of the tested sample increased significantly from ~ 516 MPa to ~ 766 MPa and the UTS remarkably improved from ~ 772.4 MPa to ~ 976 MPa. In contrast, the corresponding tensile elongation dropped from about 35%

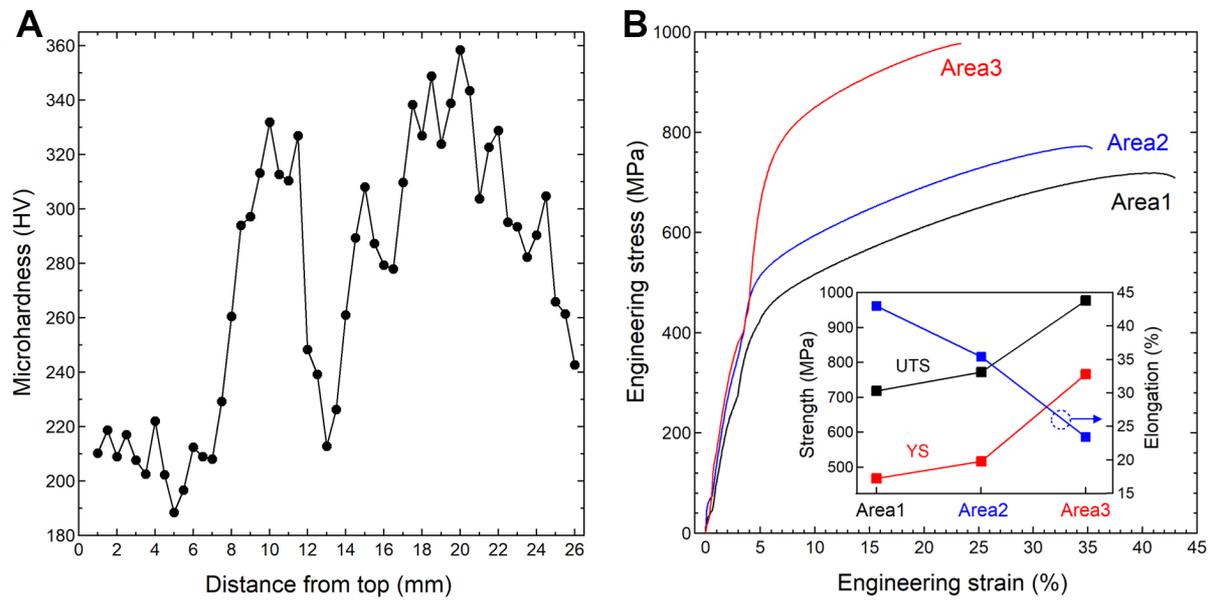


Figure 8. Mechanical properties of the as-fabricated $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ alloy: (A) microhardness as a function of the distance from the sample top measured along the cross-section; (B) engineering stress-curves measured from similar heights of three areas (1, 2 and 3 as indicated in Figure 1), the inset shows the obtained ultimate tensile strength (UTS), yield strength (YS) and elongation.

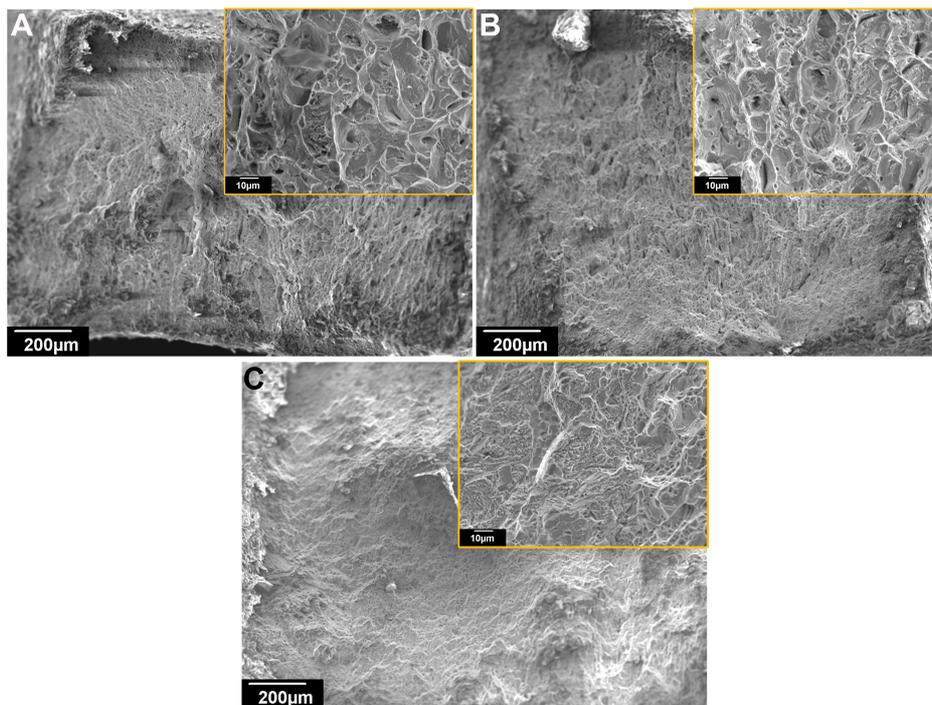


Figure 9. Scanning electron microscope (SEM) images showing the fracture morphologies after tensile testing of high-entropy alloy (HEA) samples extracted from various areas: (A) Area 1; (B) Area 2; and (C) Area 3.

to 23%. These results indicate that the heterogeneities in the microhardness and tensile properties exist along the building direction of the produced HEA material. Fractographic observations shown in Figure 9 confirm the ductile fracture for all the samples, evidenced by the dimple morphologies on the fracture

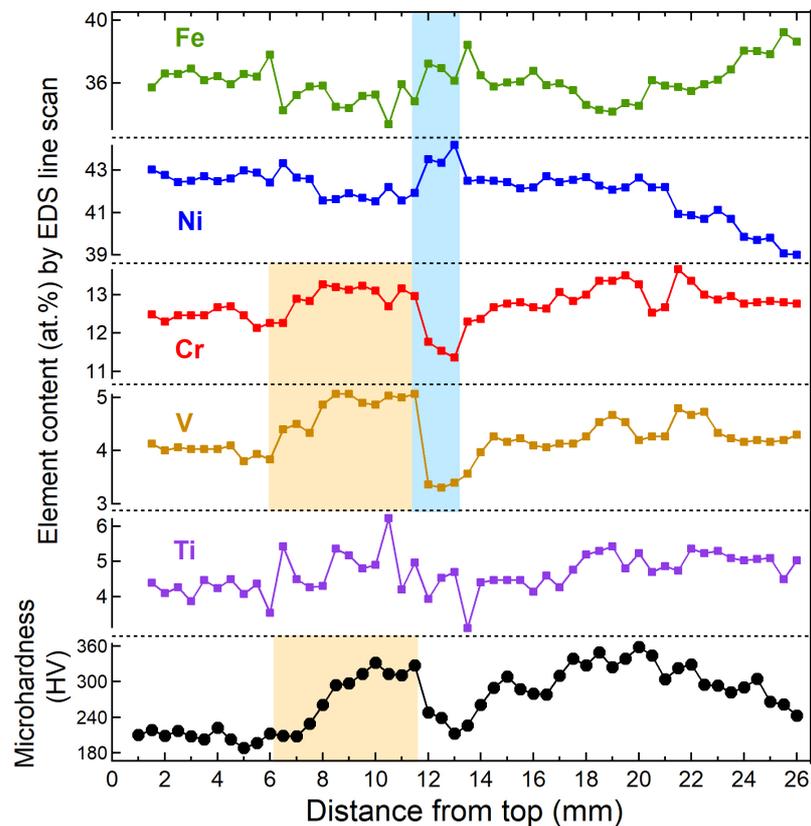


Figure 10. Energy dispersive X-ray spectroscopy (EDS) line scanning on the cross-section across the dendritic microstructure, showing the compositional variation as a function of the distance from the sample top. The errors of the EDS data for the measured elements (Fe, Cr, V, Ti and Ni) are estimated to be within ± 1 at.% based on the counting statistics of X-ray spectra in the EDS measurements. To follow the correlation between the composition and microhardness, the microhardness data in [Figure 7A](#) is also included.

surface. The dimple sizes decrease with increasing the sampling locations, in line with the observation of the decreased tensile ductility.

Considering that a fundamental strengthening mechanism of HEAs is the solid-solution strengthening, we explored the correlation between the observed microhardness changes with the compositional variation in the designed alloy that is occurred due to segregation during solidification. The EDS line scanning on the cross-sectional sample was performed, and the obtained result is plotted together with the microhardness data and shown in [Figure 10](#). Note that the quantification of these elements by EDS is in experimental error (within ± 1 at.%). However, the comparison of their relative contents measured from the same experiment is valid. It is interesting to observe that the distribution profiles of Cr and V elements resemble that of the microhardness data measured along the building direction, as highlighted by the yellow boxes in [Figure 10](#). The elemental segregation as observed in [Figure 7](#) is also confirmed by the EDS line scanning, as exemplified by the light blue box in [Figure 10](#), which shows the enrichments of Ni and Fe elements are accompanied by the deficiencies of Cr and V elements. The Ti elemental segregation is not clearly identifiable in [Figure 10](#). Based on these results, it is considered that the Cr and V elements play significant roles in the solid-solution strengthening in the designed alloy. However, other factors that may influence the strengthening, like residual stress and local texture changes, may not be ruled out, and further investigations are needed to advance the understanding.

CONCLUSIONS

In this work, the currently known empirical thermodynamic models and the CALPHAD approach (Thermo-Calc) were used to design a low neutron cross-section, FCC HEA, with a nominal composition of $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$. The results predicted that a dominant FCC phase and minor Ni_3Ti precipitation would form in the designed system. To verify the predictions, a new PAAM approach was used to fabricate the designed HEA material. A range of microstructural characterizations and mechanical tests were also performed. The principal conclusions in the following are obtained.

(1) The phase characterization acquired from XRD and TEM are generally coherent with the CALPHAD prediction. The $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ HEA mainly consists of FCC single phase with a small quantity of Ni_3Ti precipitations. However, a part of L_{12} ordered phase was detected by TEM investigation. The further dilatometer results suggested that the $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ HEA prefers to generate L_{12} structure under equilibrium state, but the disordered phase was mainly produced in the as-fabricated alloy because of the high cooling rate of PAAM system.

(2) Microstructure evaluation of $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ HEA buildup wall in three areas along the building direction were compared. In Area 1, cellular-like dendrites were formed, while columnar grains with elongated dendrites were exhibited in Areas 2 and 3. This phenomenon can be attributed to the thermal gradient along the building direction during the deposition process.

(3) The EBSD analysis support to the Thermo-Calc calculation about the partitioning of primary elements. The Ni and Ti are more inclined to solidified firstly and form the interdendritic region when the Fe, Cr and V gathered at the dendritic cores.

(4) The hardness of the as-fabricated HEA is heterogeneous from top to bottom, which would be caused by the variation of composition. The solid solution strengthening effect of Cr and V contributes to the increase of hardness obviously. Meanwhile, the tensile strength and 0.2% yield strength of $\text{FeCr}_{0.4}\text{V}_{0.3}\text{Ti}_{0.2}\text{Ni}_{1.3}$ wall increased from Area 1 to Area 3 (719 MPa to 976 MPa and 468 MPa to 766 MPa, respectively). The trade-off effect led to the decrease of ductility along with the height of the deposition wall, from 43% to 23%.

DECLARATIONS

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Authors' contributions

Made substantial contributions to conception and design of the study and performed data analysis and interpretation: Dong B, Wang Z, Li H

Performed data acquisition, as well as provided administrative, technical, and material support: Zhu H, Muránsky O, Qiu Z, Shen C, Pan Z

Availability of data and materials

The raw/processed data required to reproduce these findings cannot be shared at this time as the data is used as part of an ongoing study.

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Conflicts of interest

All authors declared that there are no conflicts of interest.

Ethical approval and consent to participate

Not applicable.

Consent for publication

Not applicable.

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